



SCR Catalyst Management

2005 Duke / WPCA NO_x

Seminar

June 7, 2005

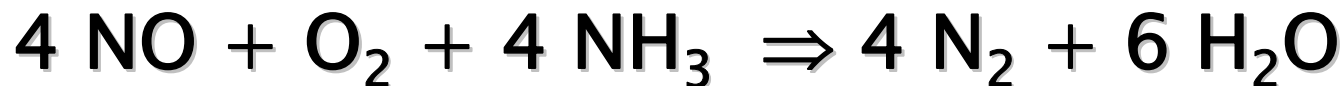
by

T. Nathan White, Haldor Topsoe, Inc.

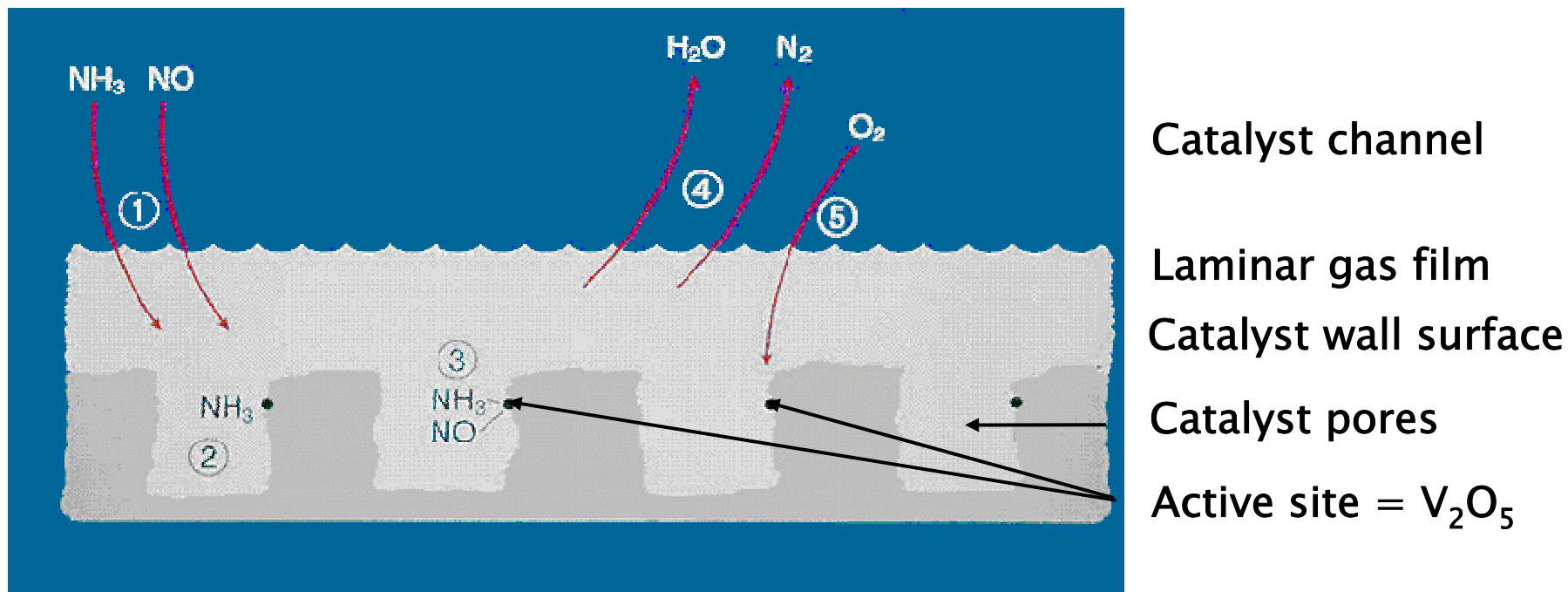


- DeNOx reaction
- Fuel impact
 - Ash
 - Amount
 - Particle size distribution
 - Minerals
 - Trace Elements
 - % Sulfur
- Catalyst sampling / analysis
- Inspections / housekeeping
- Catalyst replacement / cleaning
- Catalyst on line monitoring

The DeNO_x Reaction



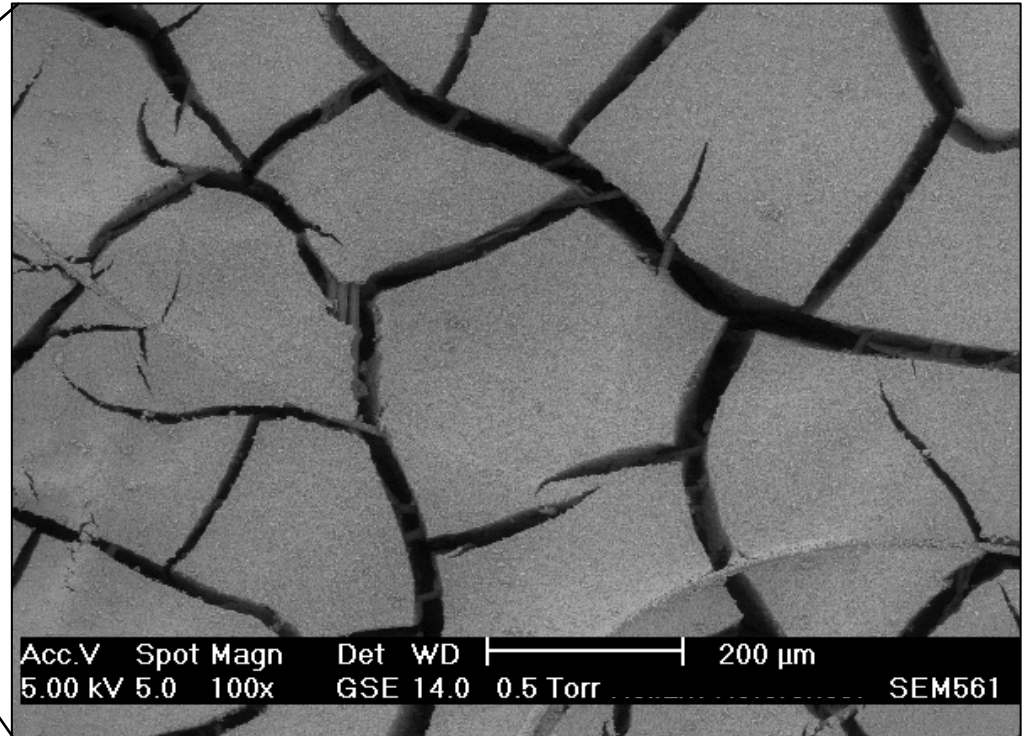
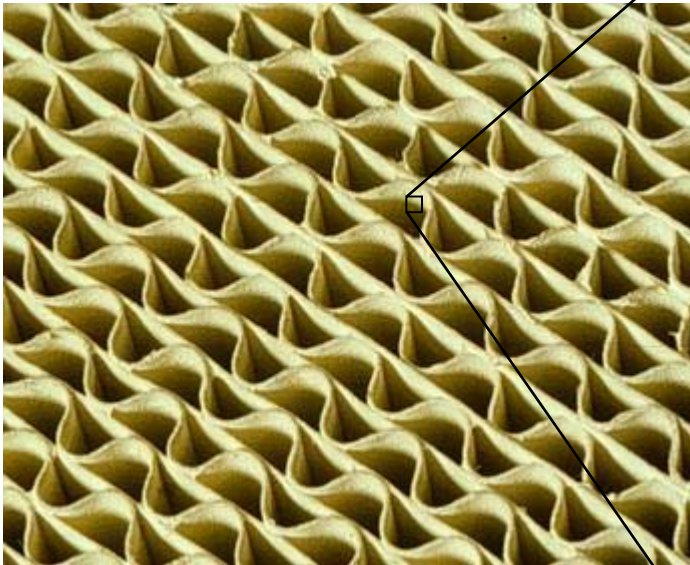
First order reaction – mass transport limited



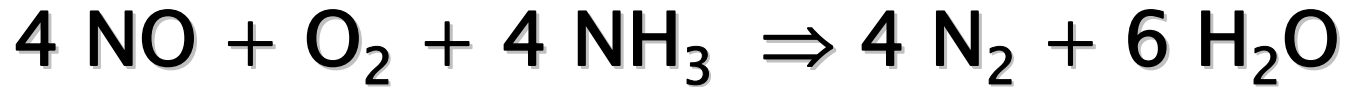
DNX Catalyst: V_2O_5 and WO_3 on TiO_2

The DeNO_x Reaction

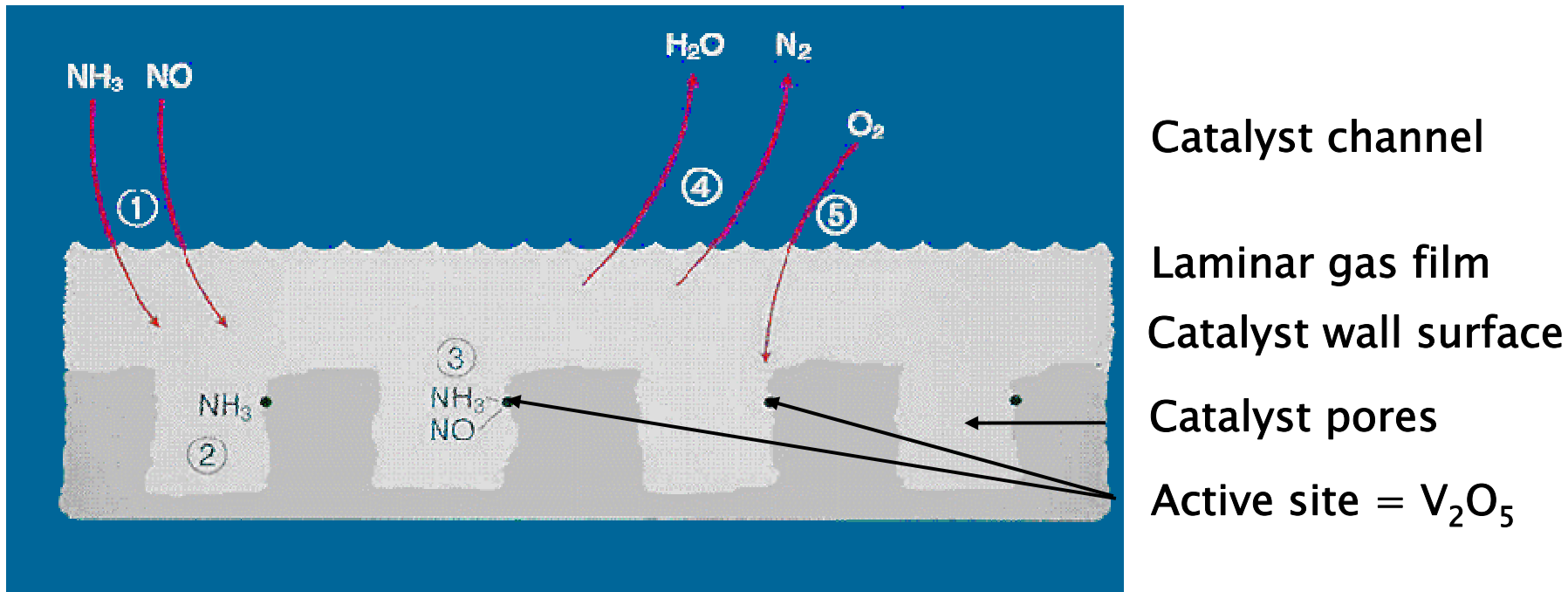
DeNO_x reaction is diffusion limited
more highways = high diffusion rate = higher activity



The DeNOx Reaction

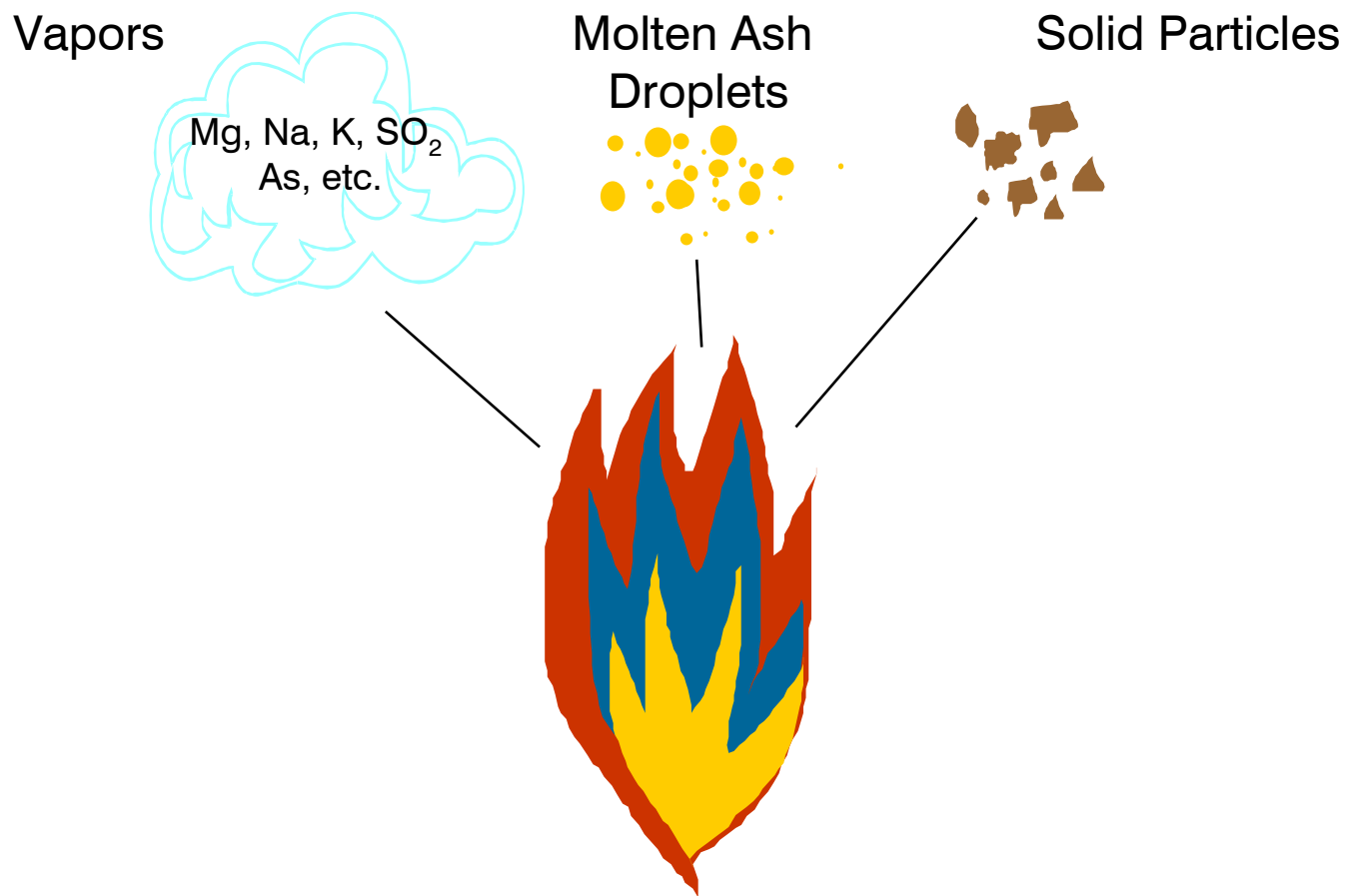


First order reaction – mass transport limited



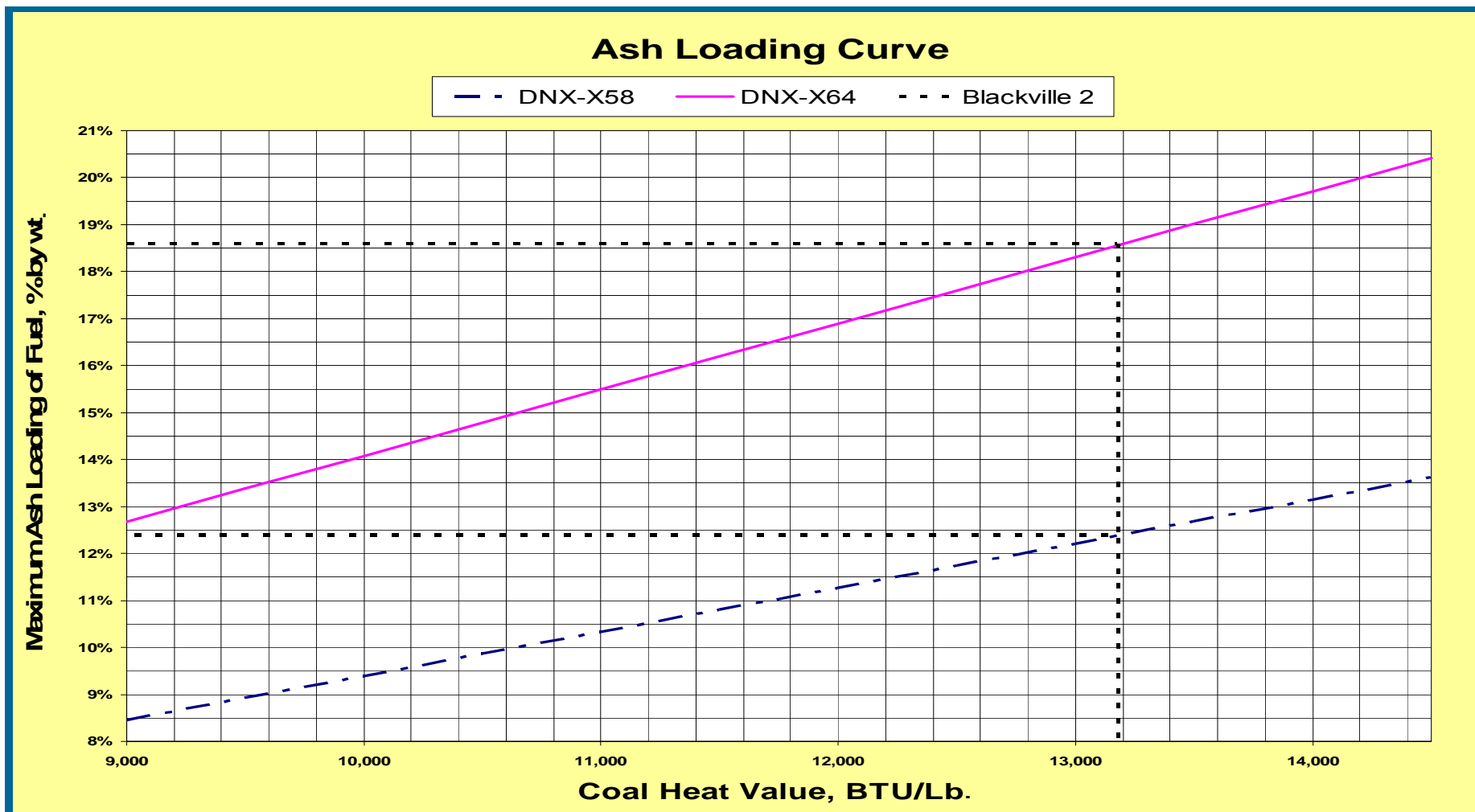
Catalyst deactivation results when one of the above is made less active.

Fossil Fuel Combustion

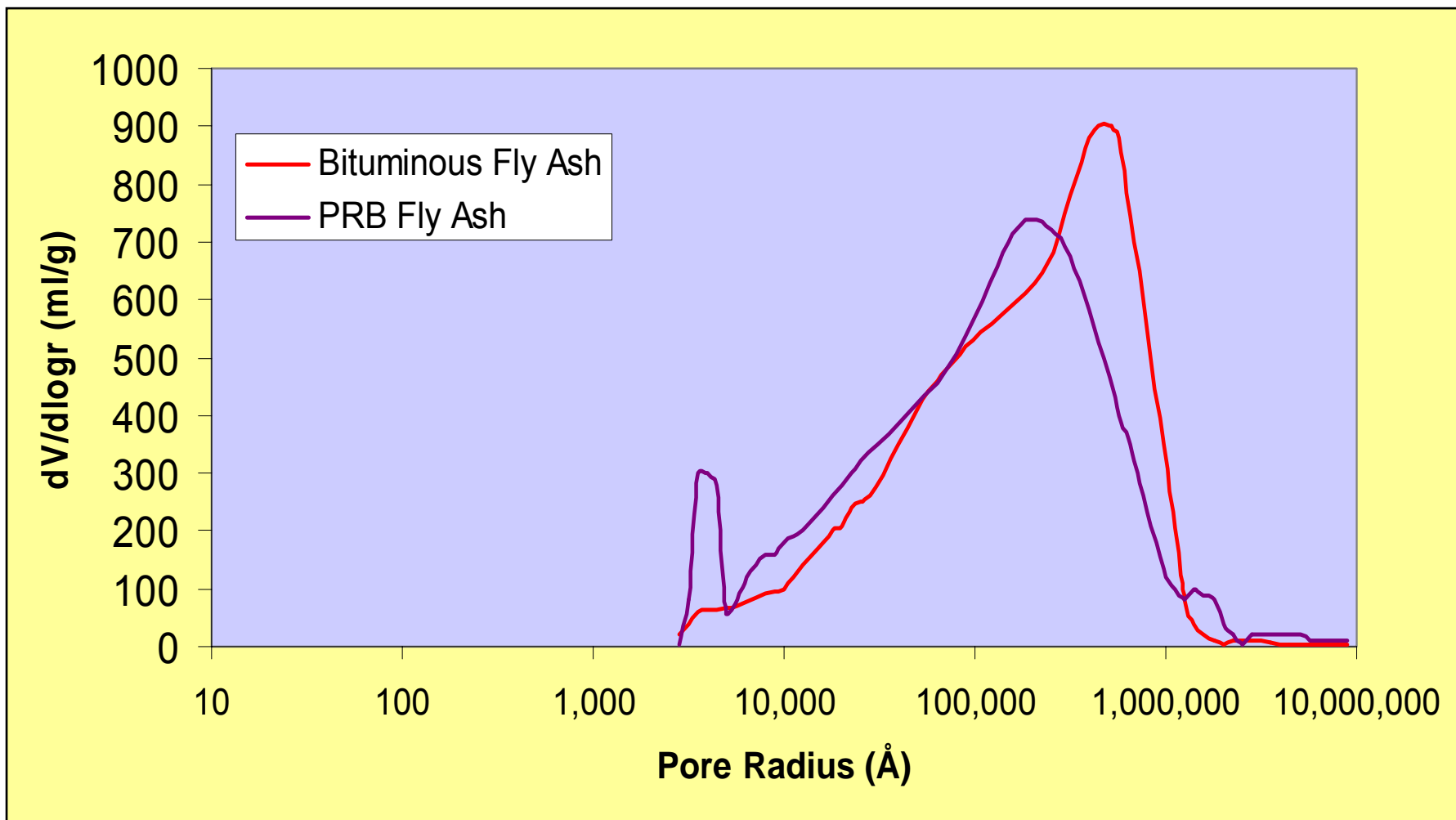


Fuel Impact - Ash

■ Catalyst pitch – amount of ash



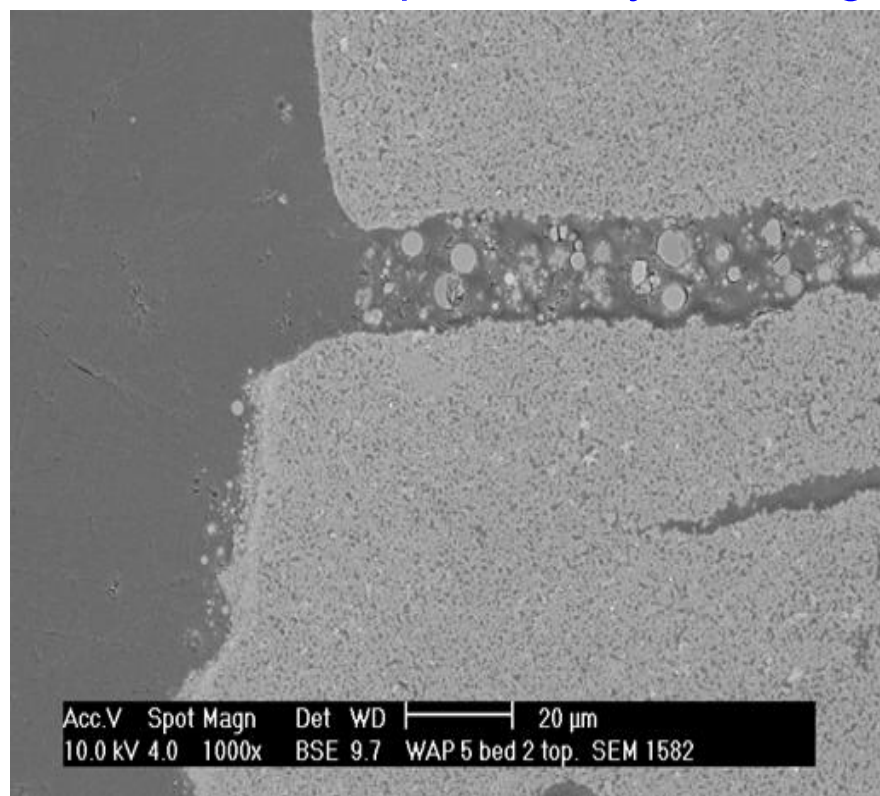
■ Catalyst pitch – particle size distribution



Fuel Impact – Ash Mineral Analysis

Coal Ash Mineral Composition	Bituminous Coal
<i>Al₂O₃, %</i>	22
<i>Fe₂O₃, %</i>	20.3
<i>SiO₂, %</i>	42.6
SO₃, %	5
CaO, %	5.3
K₂O, %	1.4
MgO, %	0.9
Na₂O, %	0.9

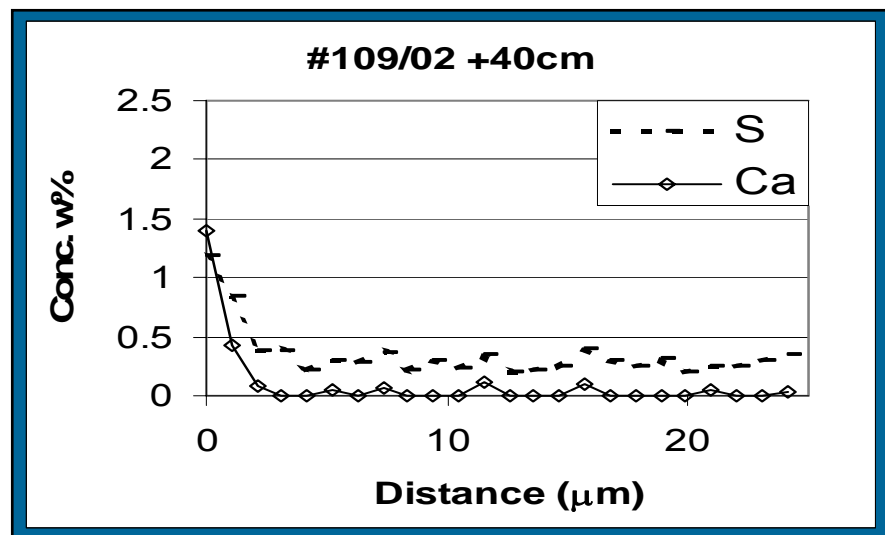
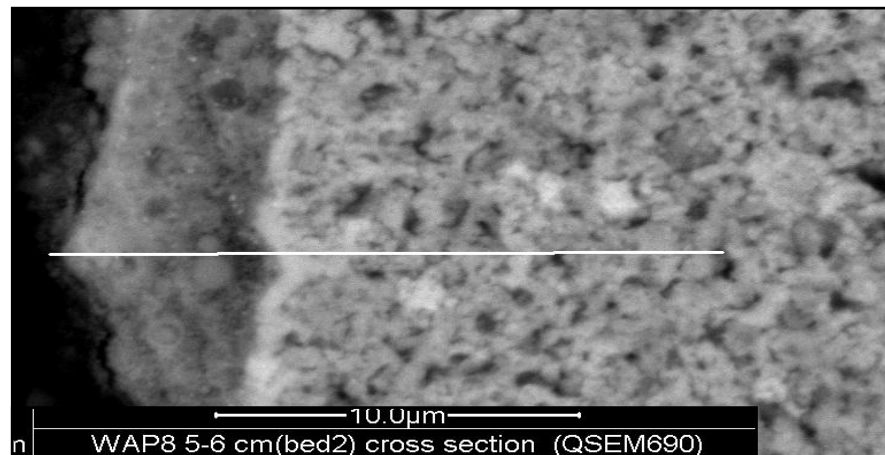
- Al, Fe and Si impacts catalyst erosion
- Fe impacts SO₂-oxidation (+ / -)
- Si can also impact catalyst fouling



Fuel Impact – Ash Mineral Analysis

Coal Ash Mineral Composition	Bituminous Coal
Al ₂ O ₃ , %	22
Fe ₂ O ₃ , %	20.3
SiO ₂ , %	42.6
SO ₃ , %	5
CaO, %	5.3
K ₂ O, %	1.4
MgO, %	0.9
Na ₂ O, %	0.9

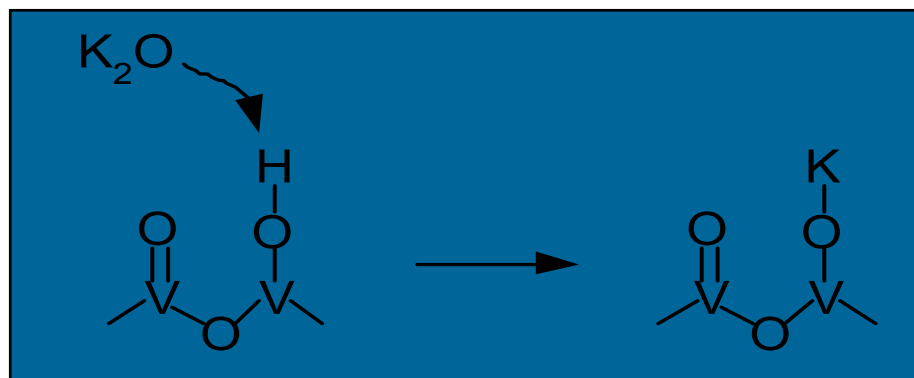
■ CaSO₄ impacts catalyst fouling



Fuel Impact – Ash Mineral Analysis

Coal Ash Mineral Composition	Bituminous Coal
Al_2O_3 , %	22
Fe_2O_3 , %	20.3
SiO_2 , %	42.6
SO_3 , %	5
CaO , %	5.3
K_2O , %	1.4
MgO , %	0.9
Na_2O , %	0.9

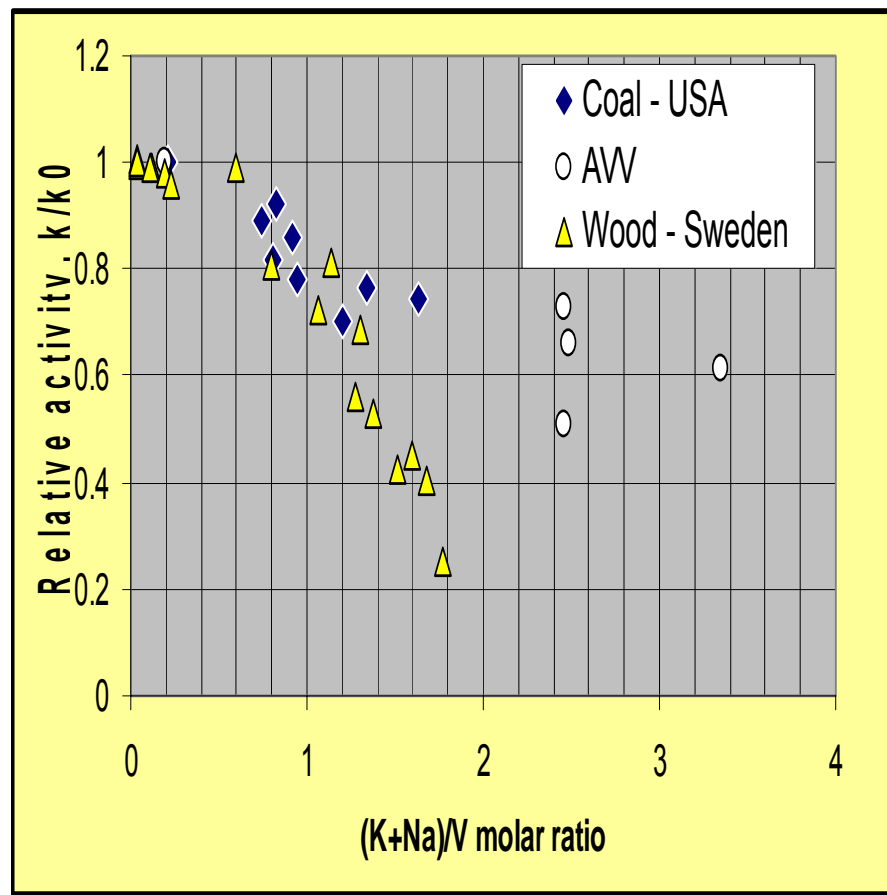
- Ca, Mg, Na and K are all catalyst chemical poisons
- Acid soluble Na in ash >1.0%
- Acid soluble K in ash > 0.5%
- Risk is considered minor in bituminous coal, as most of the alkaline metals are not water soluble
- PRB, wood, straw fuel the risk is higher



Fuel Impact – Ash Mineral Analysis

Coal Ash Mineral Composition	Bituminous Coal
Al_2O_3 , %	22
Fe_2O_3 , %	20.3
SiO_2 , %	42.6
SO_3 , %	5
CaO , %	5.3
K_2O , %	1.4
MgO , %	0.9
Na_2O , %	0.9

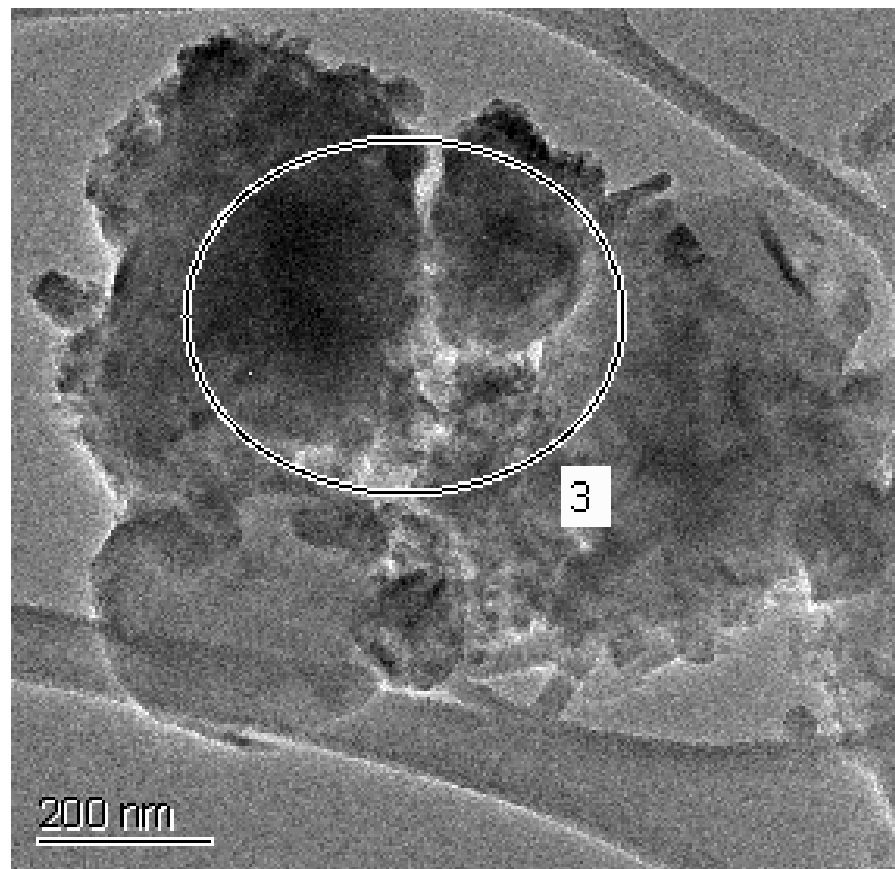
■ SO_3 in the fuel reduces poisoning



Fuel Impact – Trace Elements

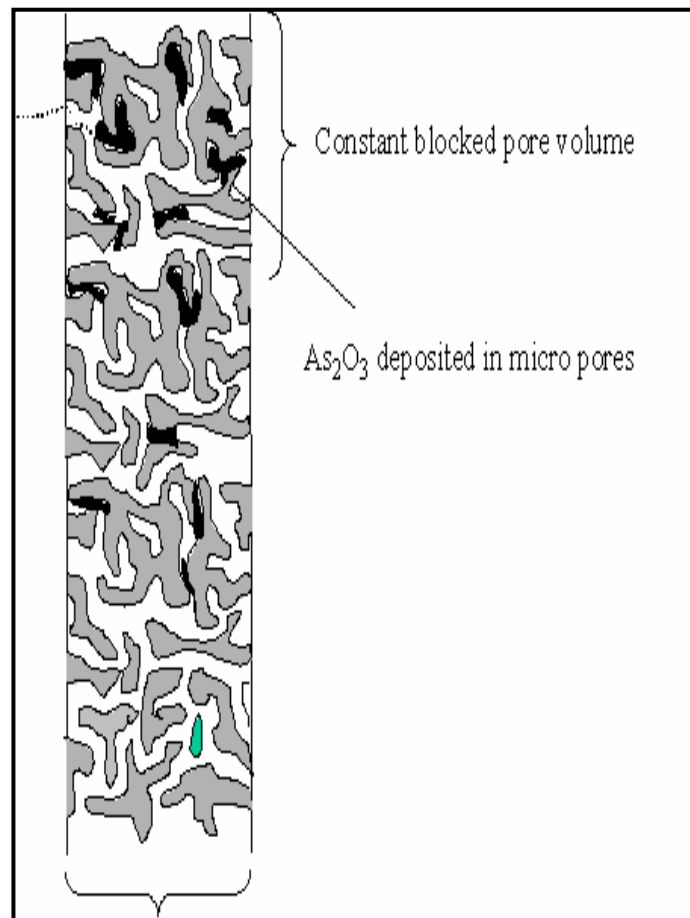
Coal Trace Elements	ppm
<i>Arsenic ppm</i>	<i>13</i>
Chrome ppm	43
Vanadium ppm	45
Copper ppm	21
Lead ppm	18
Mercury ppm	413
Nickel ppm	37

- As is a catalyst poison
- Mechanism – pore blockage



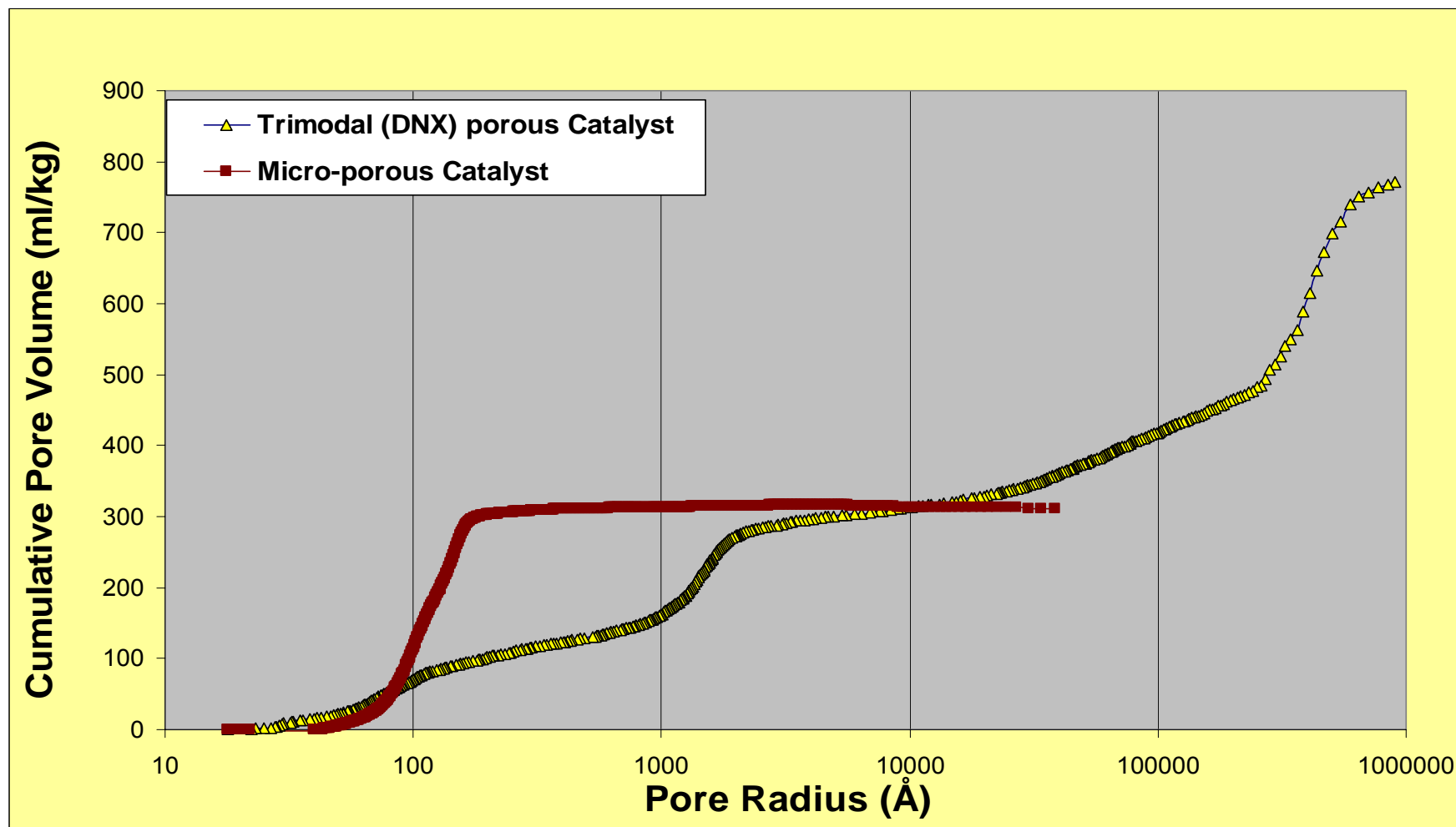
Fuel Impact – Trace Elements

- Arsenic poisoning mechanism
- As in coal is vaporized to As_2O_3 ,
- if the vapour pressure of As_2O_3 in the gas phase is higher than equilibrium, As_2O_3 will precipitate in the catalyst pores due to capillary forces,
- As condenses in the very small micro pores of the catalyst,
- As could react with the vanadium in the micro pores to form a vanadia-arsenate compound,
- this compound is so stable it inhibits V^{5+} -O-H reaction cycle.



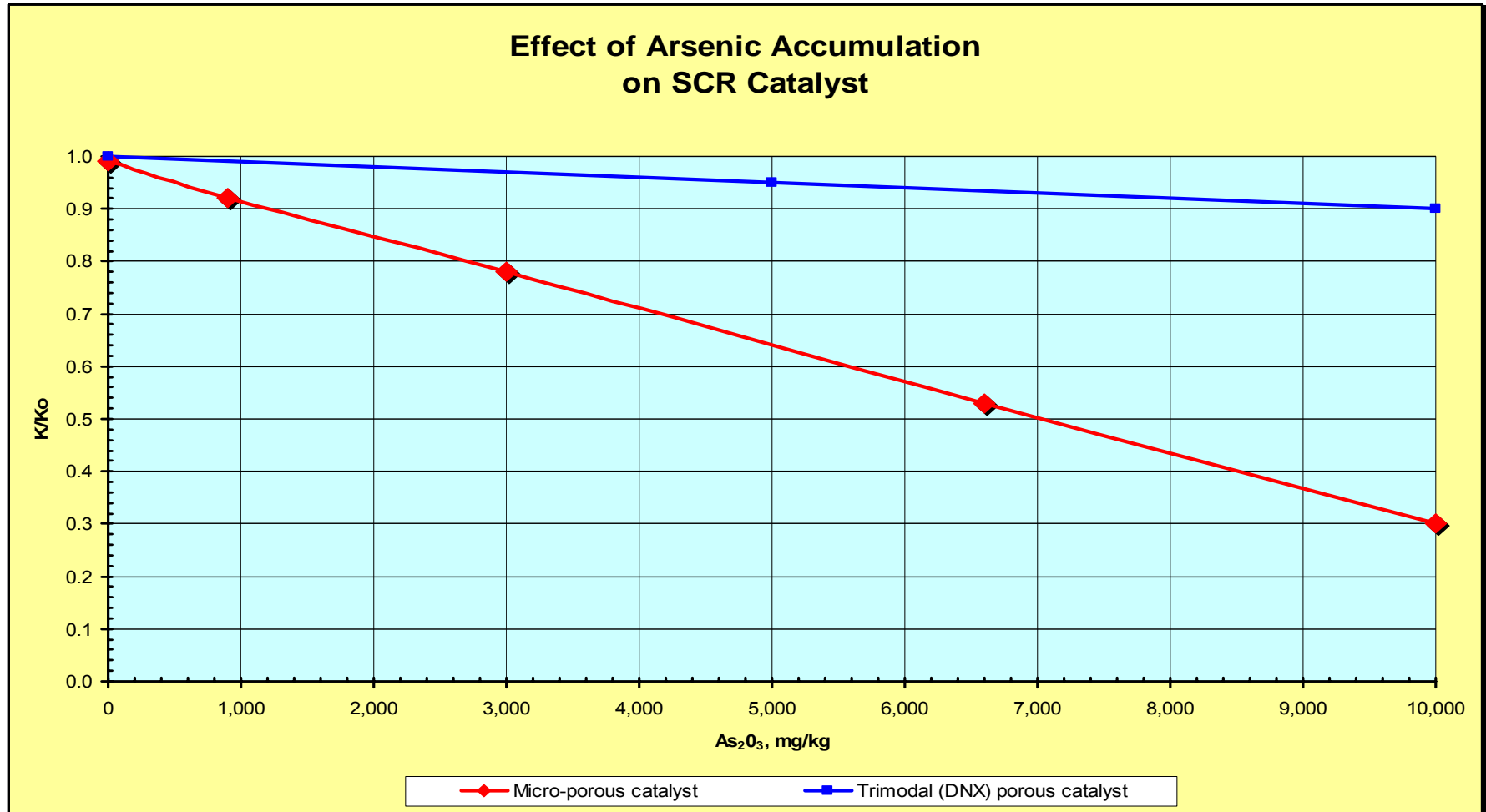
Fuel Impact – Trace Elements

- A catalyst with a diverse and accumulated larger pore volume



Fuel Impact – Trace Elements

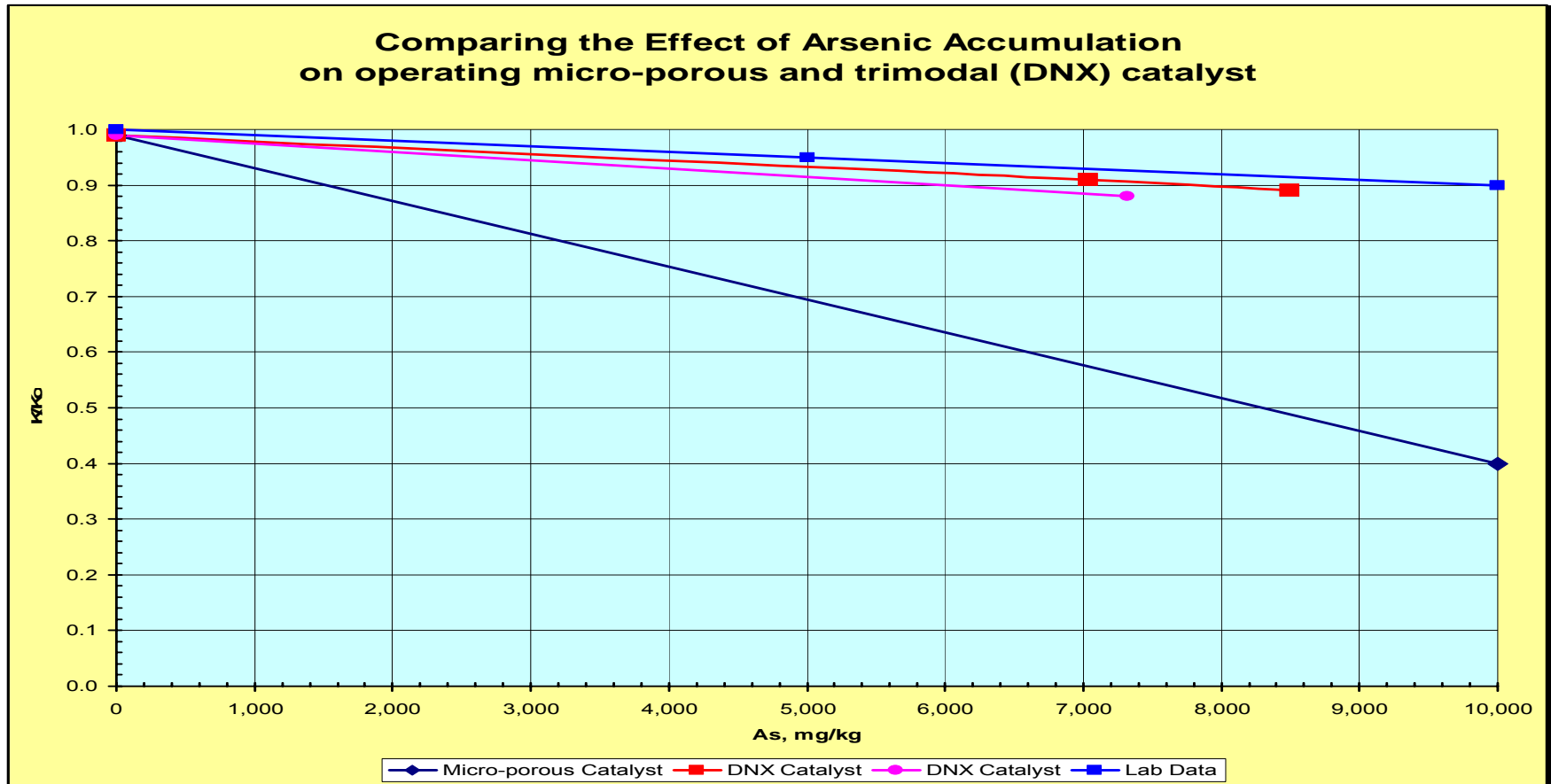
■ is more resistant to As poisoning



Fuel Impact – Trace Elements

■ As containing fuel

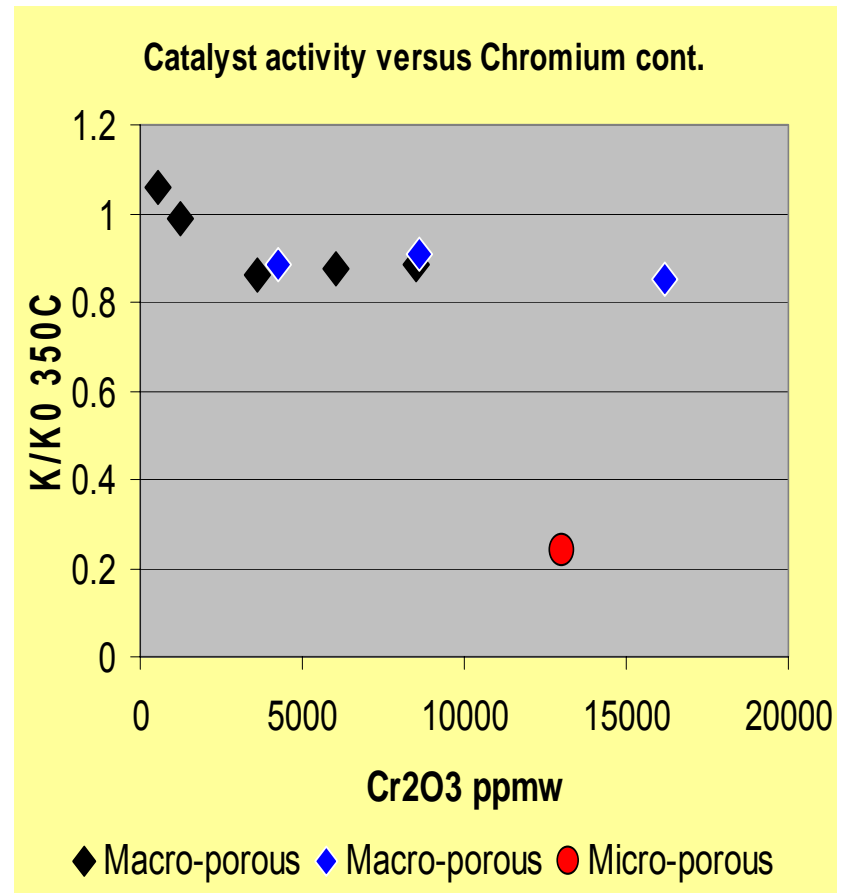
- Micro-porous catalyst increase CaO.
- Use a catalyst with a diverse catalyst pore volume, w/ coal levels up to 25 ppm



Fuel Impact – Trace Elements

Coal Trace Elements	ppm
Arsenic ppm	13
<i>Chrome ppm</i>	<i>43</i>
Vanadium ppm	45
Copper ppm	21
Lead ppm	18
Mercury ppm	413
Nickel ppm	37

- Cr is a catalyst pore blockage poison



Fuel Impact – Trace Elements

Coal Trace Elements	ppm
Arsenic ppm	13
Chrome ppm	43
<i>Vanadium ppm</i>	<i>45</i>
Copper ppm	21
Lead ppm	18
Mercury ppm	413
Nickel ppm	37

- VOSO_4 will accumulate on the on the catalyst surface
- Can deactivate catalyst, collapses active sites, if accumulation is high enough
- Increases SO_2 - oxidation as V accumulates on the catalyst
- Oil, Pet coke

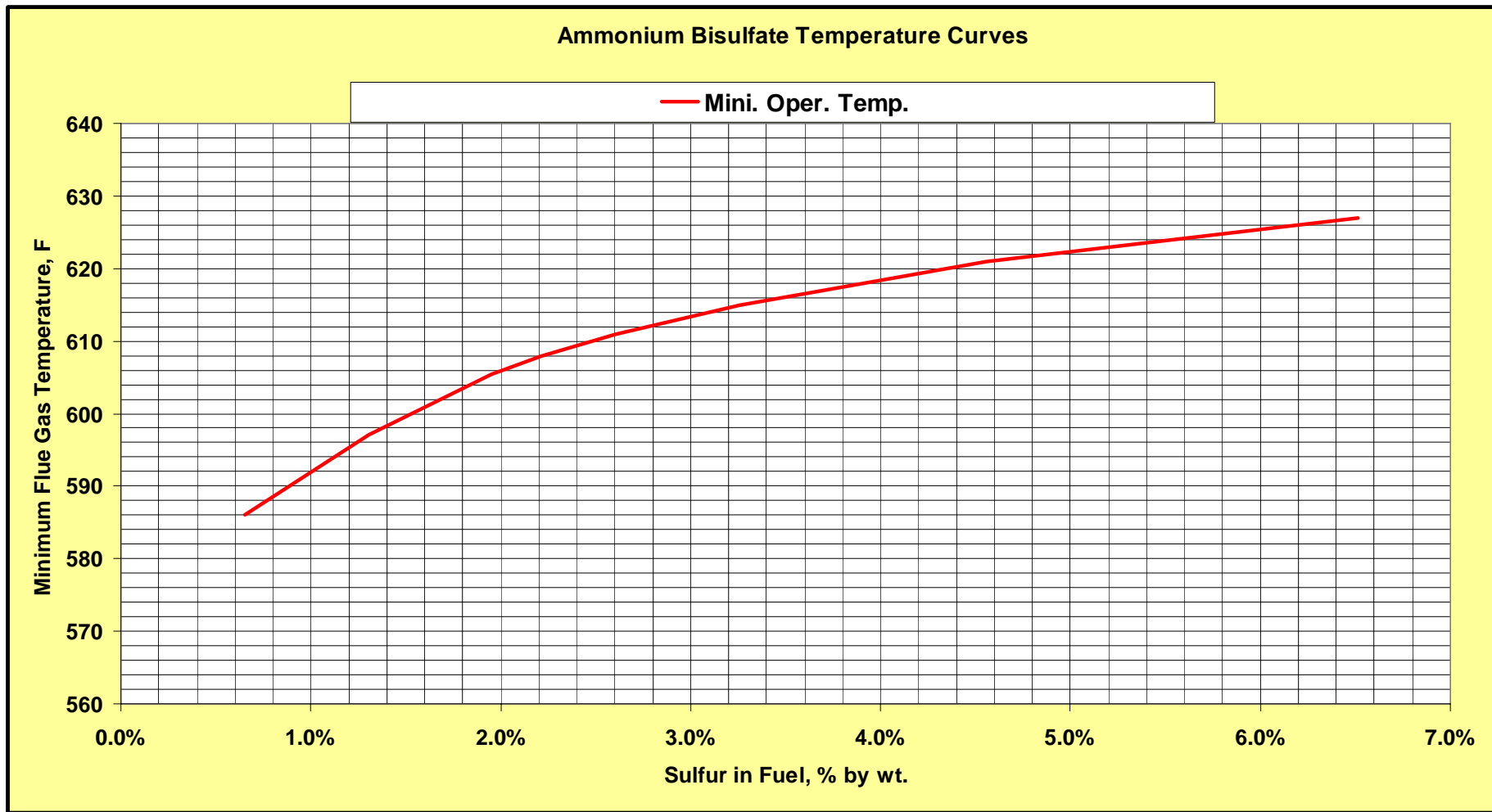
Fuel Impact – Trace Elements

Coal Trace Elements	ppm
Arsenic ppm	13
Chrome ppm	43
Vanadium ppm	45
<i>Copper ppm</i>	<i>21</i>
<i>Lead ppm</i>	<i>18</i>
<i>Mercury ppm</i>	<i>413</i>
<i>Nickel ppm</i>	<i>37</i>

- Heavy and Base metals – antimony, chrome, copper, lead, mercury, nickel, tin, zinc
- Alkali Metals – cesium, lithium, rubidium
- Alkaline Earth Metals – barium, strontium
- Phosphorous

Fuel Impact - % Sulfur

- To prevent operating catalyst below Ammonium Bisulfate (ABS) formation temperature



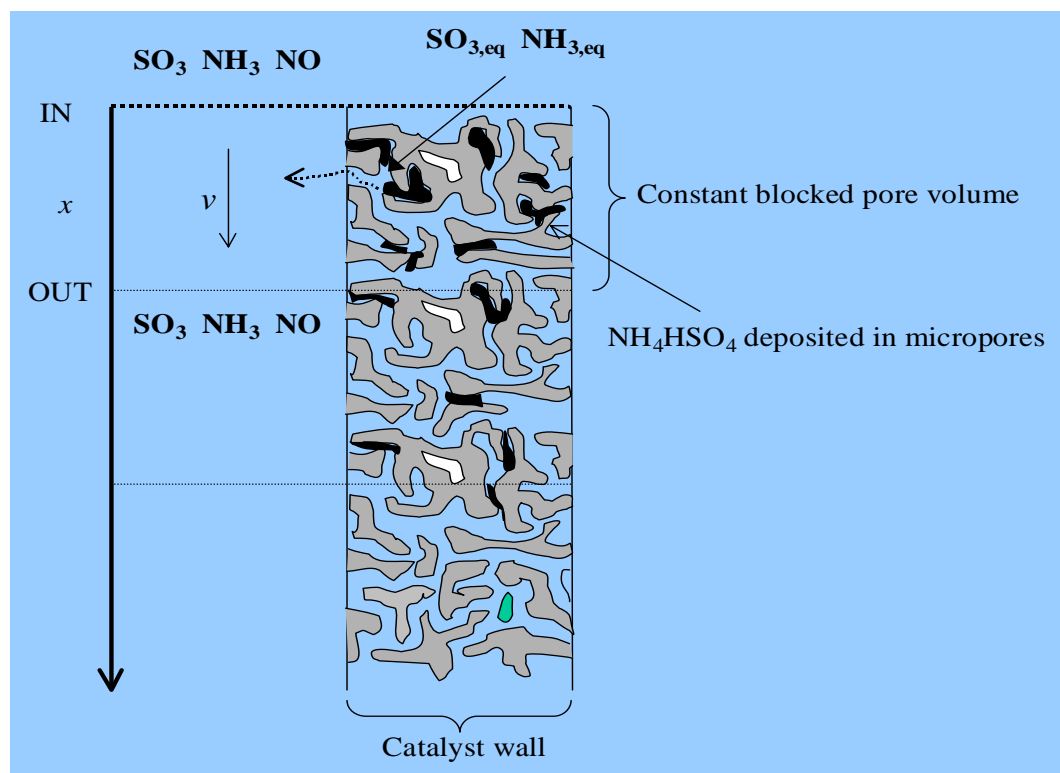
Fuel Impact - % Sulfur

■ ABS - NH_4HSO_4



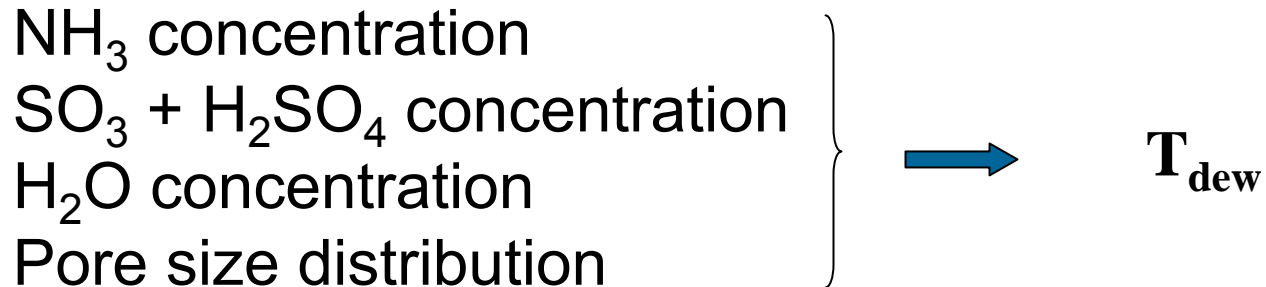
■ Mechanism

- Pore blockage by capillary condensation.



Fuel Impact - % Sulfur

■ Mechanism - ABS dew point calculation



Clausius/Clapeyron equation:

$$\ln(P_{\text{NH}_3} \cdot P_{\text{H}_2\text{SO}_4})_{\text{eq,bulk}} = 27.97 - \frac{26671}{T[\text{K}]}$$

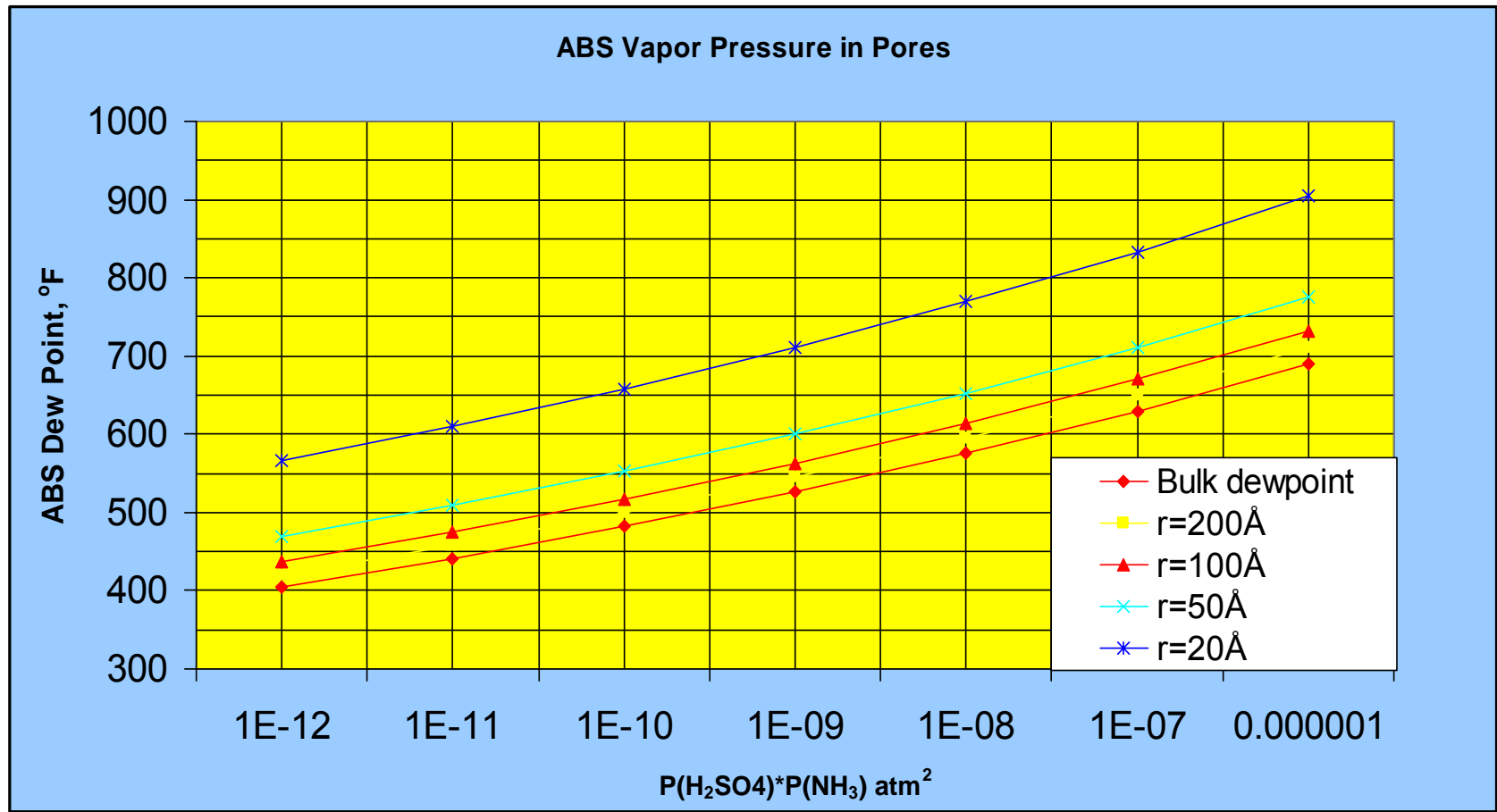
Matsuda, S., 1982

Kelvin equation:

$$\ln \frac{(P_{\text{NH}_3} \cdot P_{\text{H}_2\text{SO}_4})_{\text{eq,pore}}}{(P_{\text{NH}_3} \cdot P_{\text{H}_2\text{SO}_4})_{\text{eq,bulk}}} = - \frac{2 \cdot \sigma \cdot M}{\rho \cdot r_{\text{pore}} \cdot R \cdot T}$$

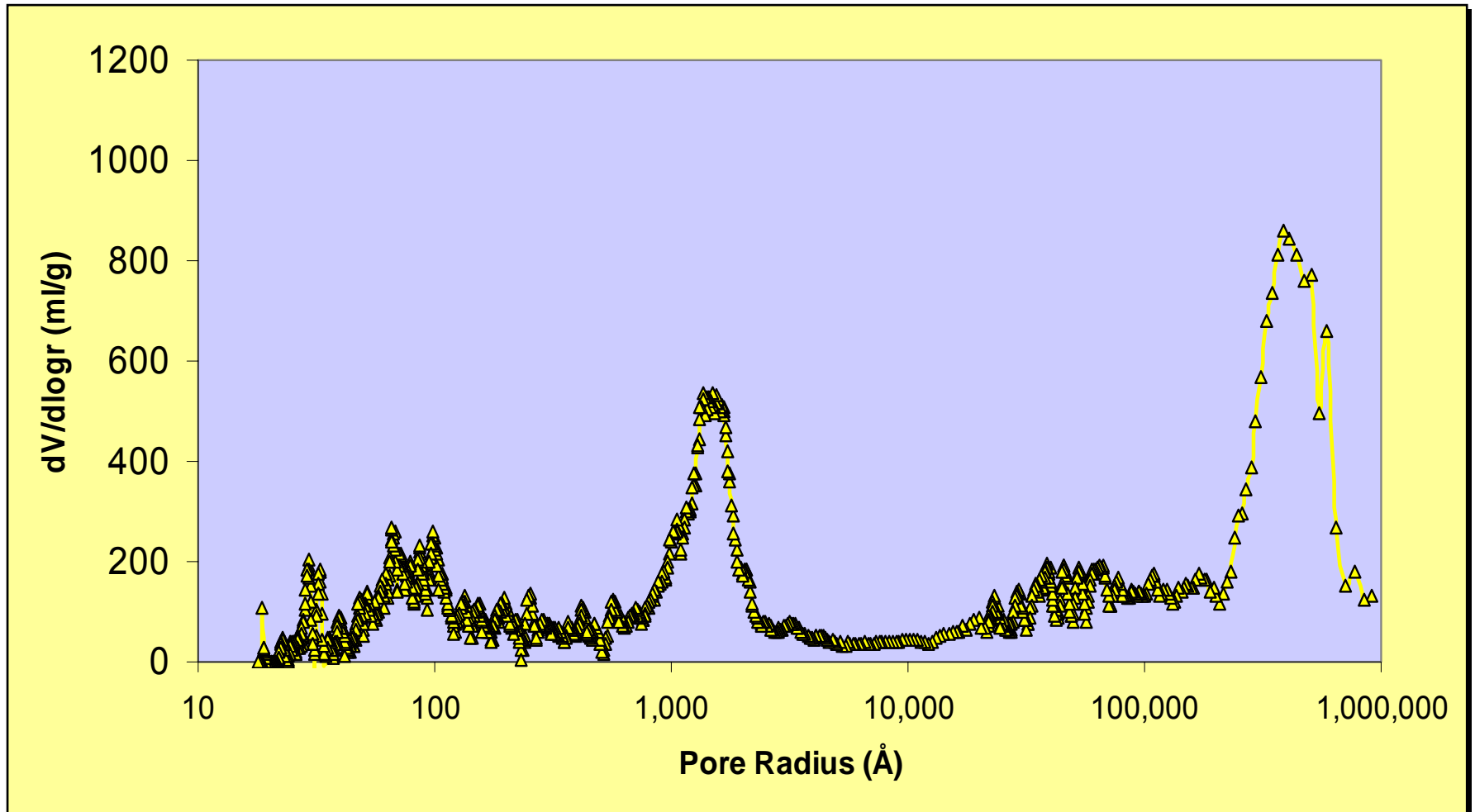
Fuel Impact - % Sulfur

- ABS dew point temperature depends on the catalyst pore distribution



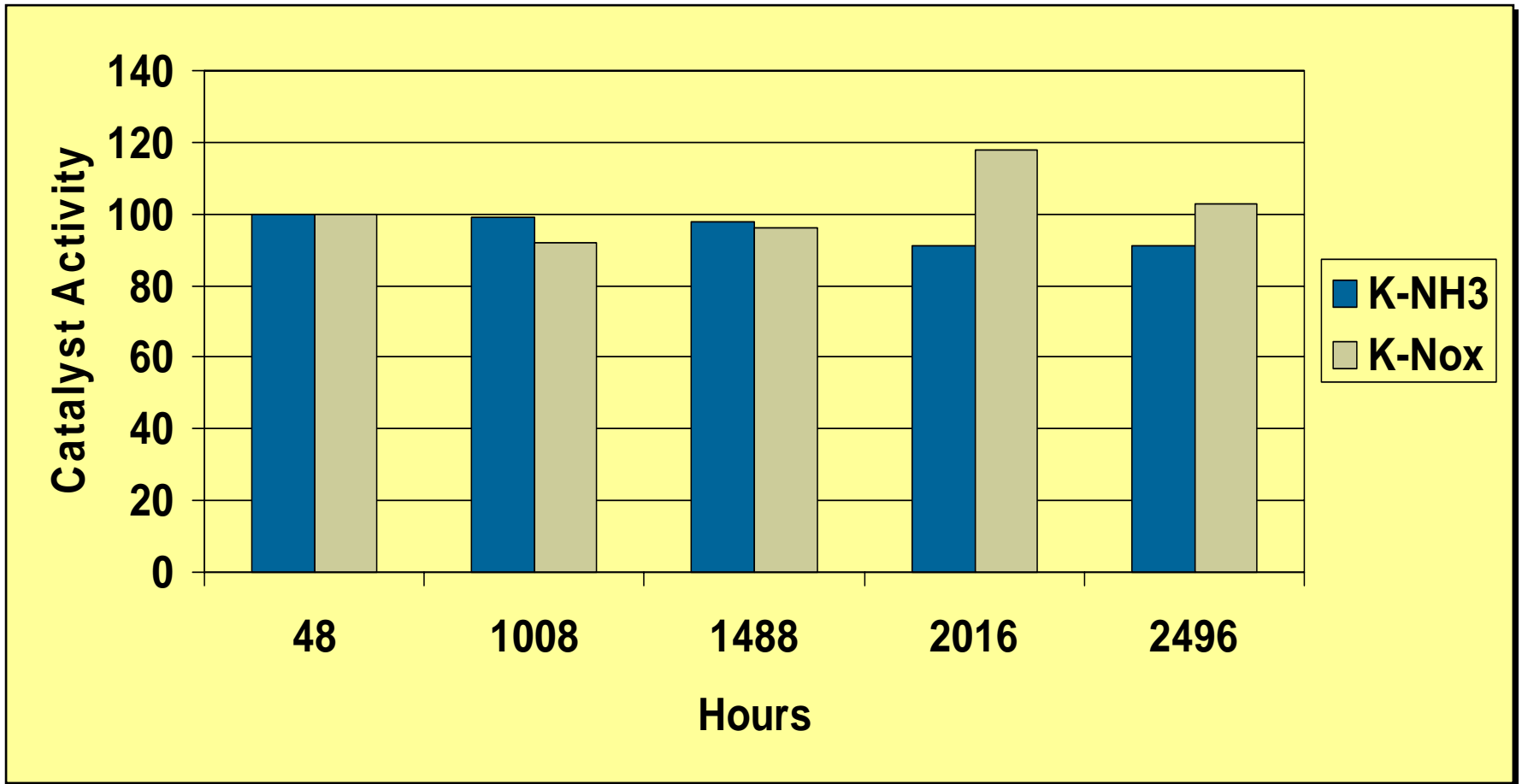
Fuel Impact - % Sulfur

- A diverse pore volume should allow operation below the ABS dew point

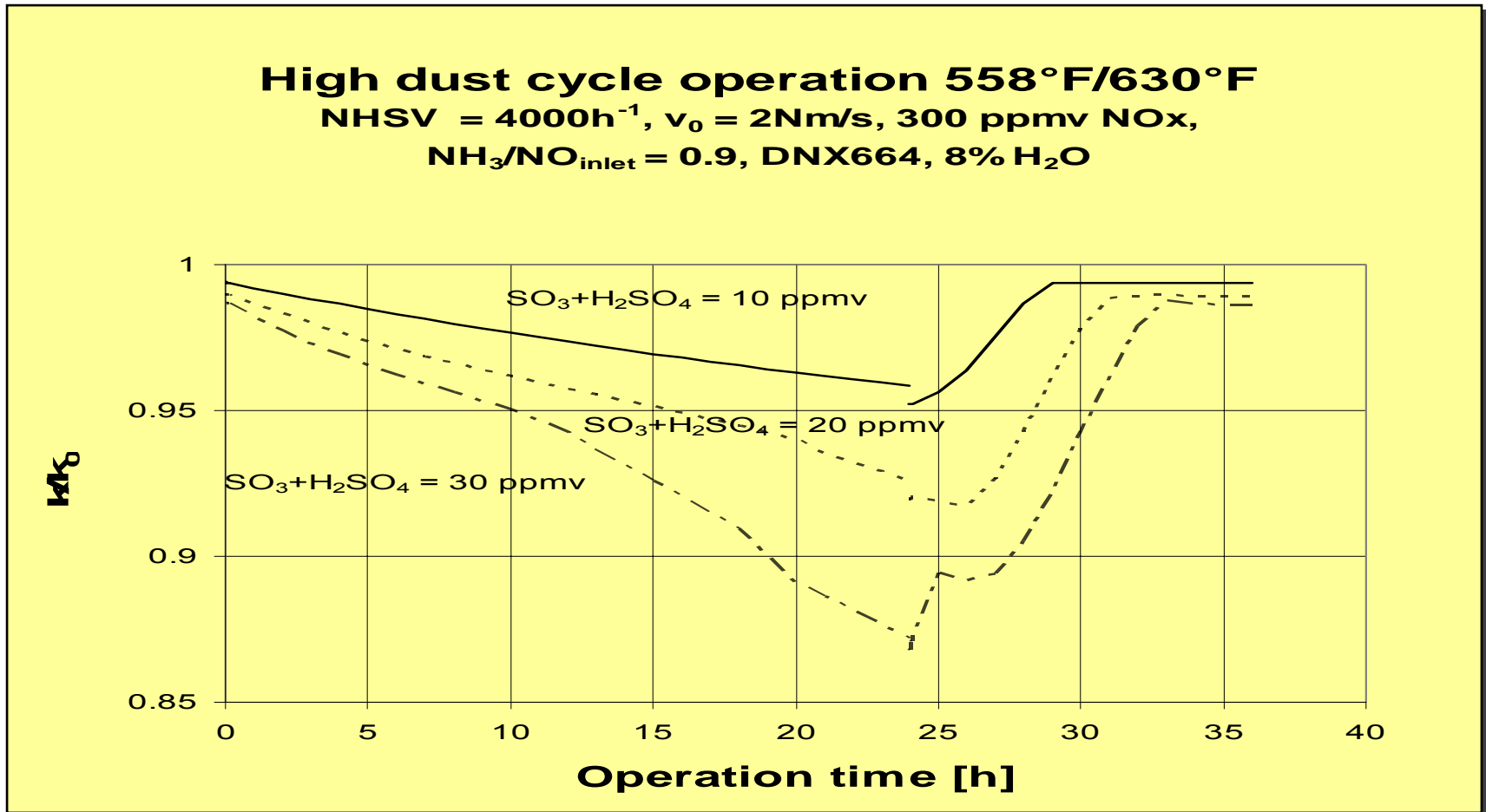


Fuel Impact - % Sulfur

- Cycle operation below and above the ABS dew point



Fuel Impact - % Sulfur



- Five (5) SCR nightly operate below ABS dew point, up to 300 continuous hours, within 10 degrees F of the ABS bulk dew point temperature.

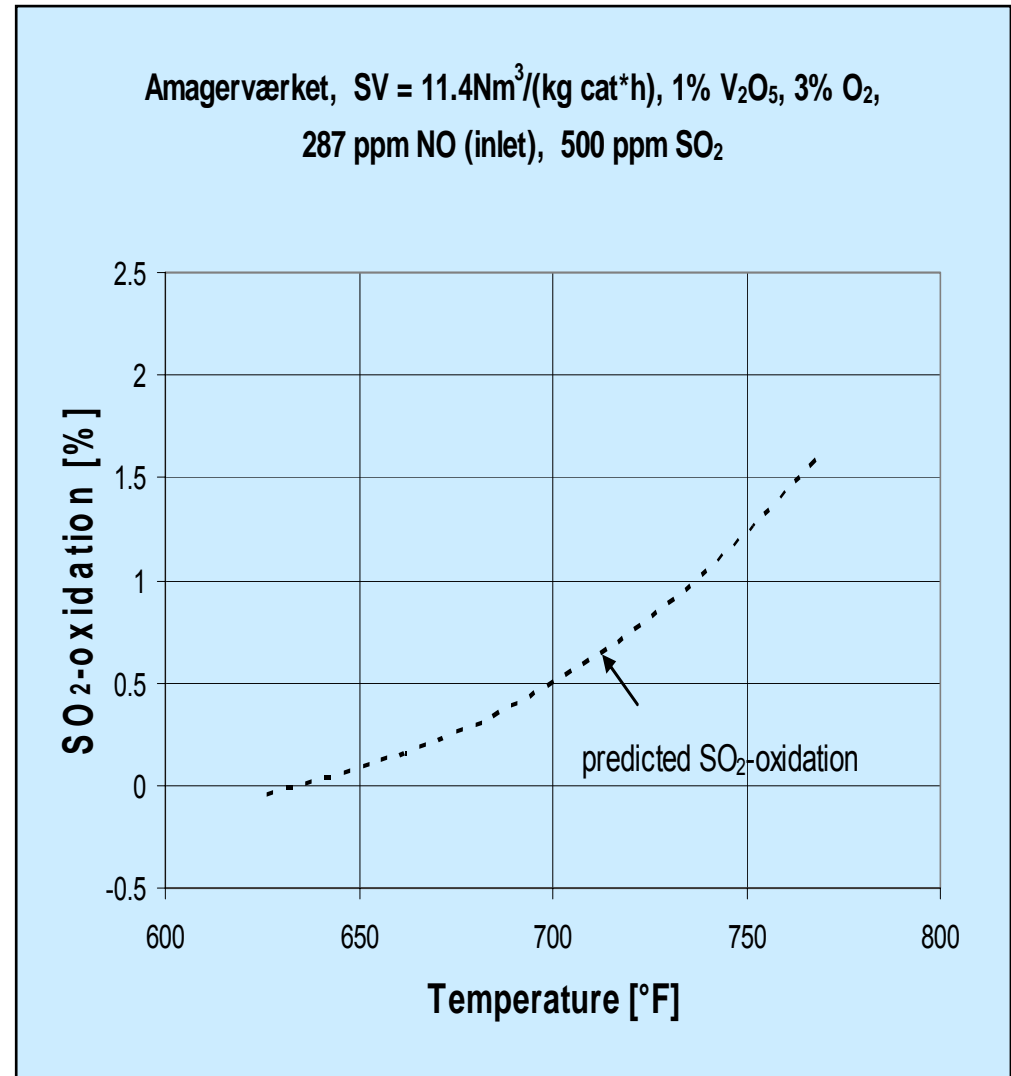
Fuel Impact - % Sulfur

■ Managing SO₂-oxidation

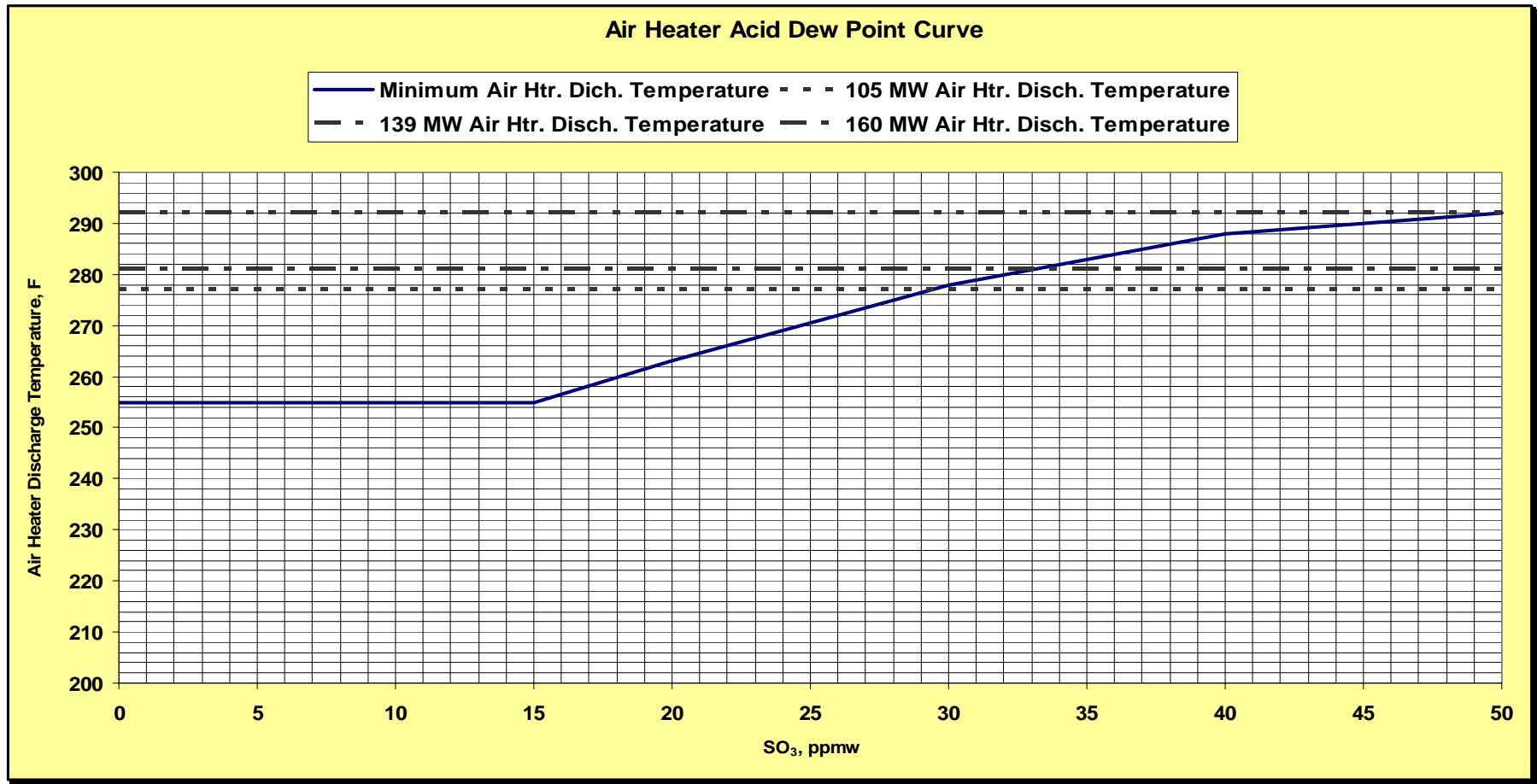
- Fuel
 - SO₂ concentration
 - Ash composition
- Boiler
 - Excess O₂
- Catalyst
 - Distribution of vanadium
 - Diverse pore structure

■ Current SO₂- oxidation experience

- High dust guarantees < 0.1%
- Operating experience < 0.1%



Fuel Impact - % Sulfur



- A 2.5 % sulfur coal, SO₂-oxidation of 0.5% increases SO₃ content > 10 ppm thereby increasing the acid dew point temperature.

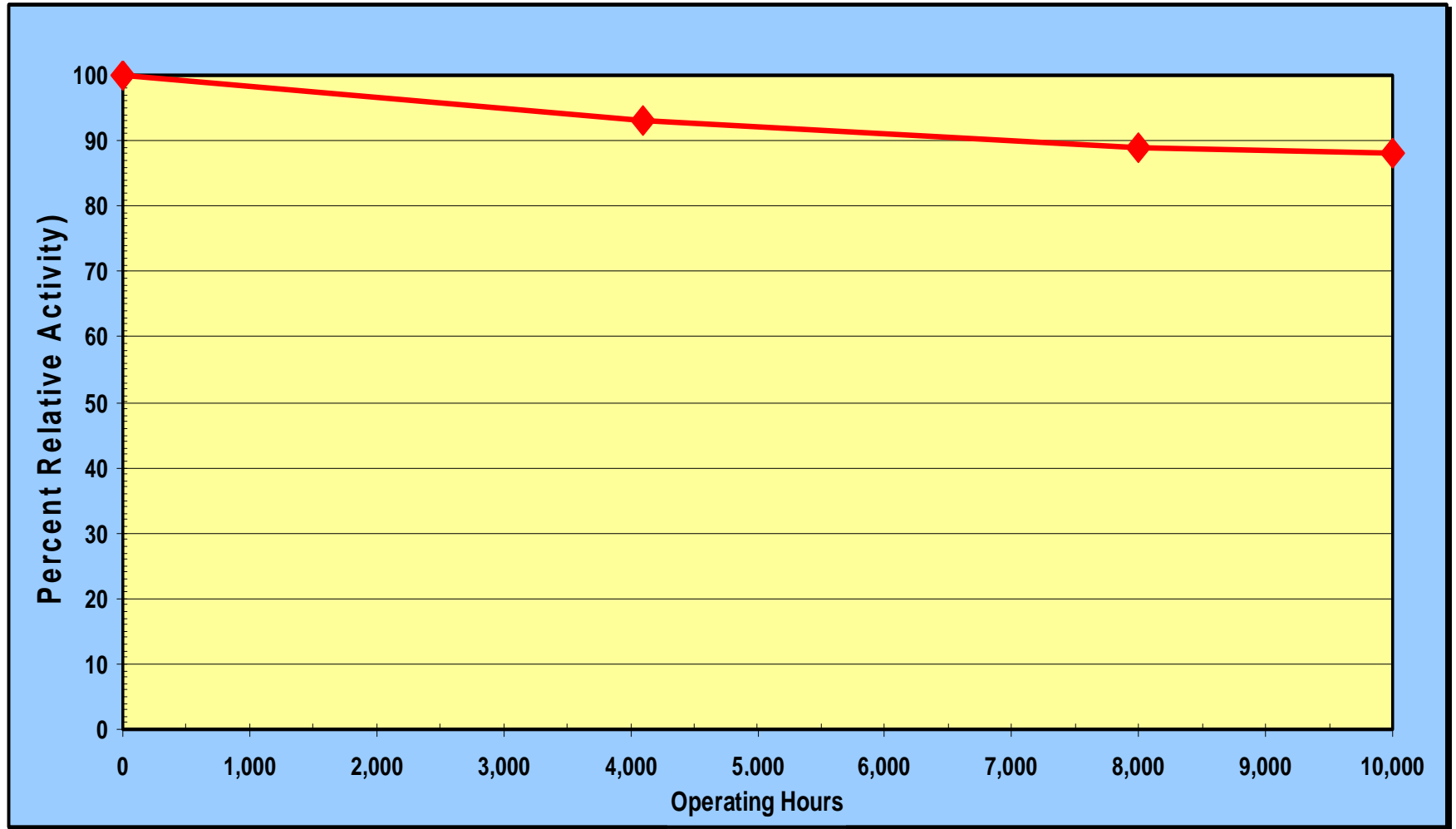
Catalyst Testing

- **Catalyst testing – Elements / Plates**
 - **When?**
 - Annual
 - End of ozone season
 - **Where should you sample?**
 - 1st layer
 - Look at flow model results
 - Area of highest dust loading
 - **What analysis should you request?**
 - Activity – STD / Unit Conditions
 - Chemical
 - Surface



Catalyst Testing

■ Activity Analysis Results



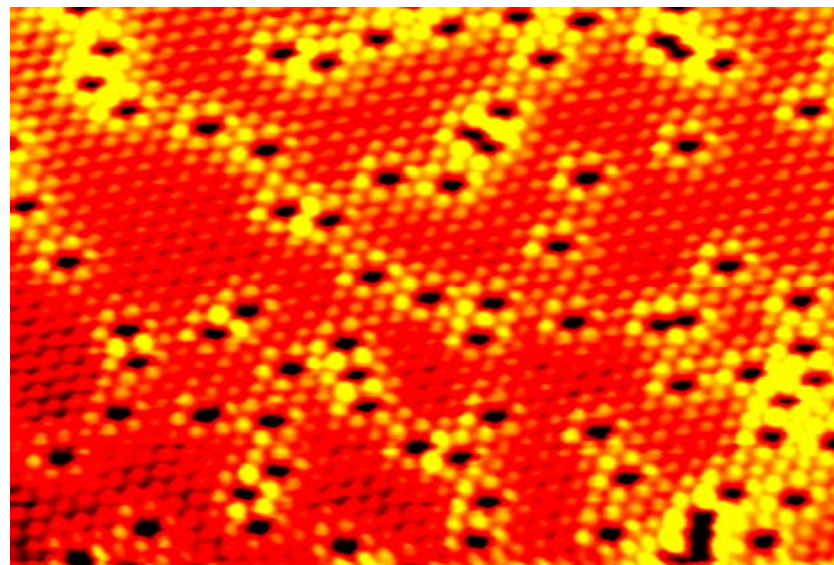
Catalyst Testing

- Chemical Analysis
 - Look at catalyst poisons

Sample	Reference	Reactor A		Reactor B	
		Original Layer 1	Duplicate Layer 1	Original Layer 1	Duplicate Layer 1
Al, %	1.12	1.52	1.43	1.76	1.67
As, ppm	< 10	6,070	6,315	4,805	4,170
Ca, %	2.56	2.53	2.43	2.69	2.45
Fe, %	0.04	0.68	0.58	1.16	0.93
K soluble, ppm	165	1,475	1,600	1,775	2,100
Mg, %	0.06	0.08	0.08	0.09	0.1
Na soluble, ppm	545	1,380	1,210	1,400	1,330
P, ppm	1,680	1,605	2,000	1,570	2,000
Si, %	5.24	5.78	5.69	6.32	6.05
S, %	0.03	0.55	0.57	0.58	0.59

Catalyst Testing

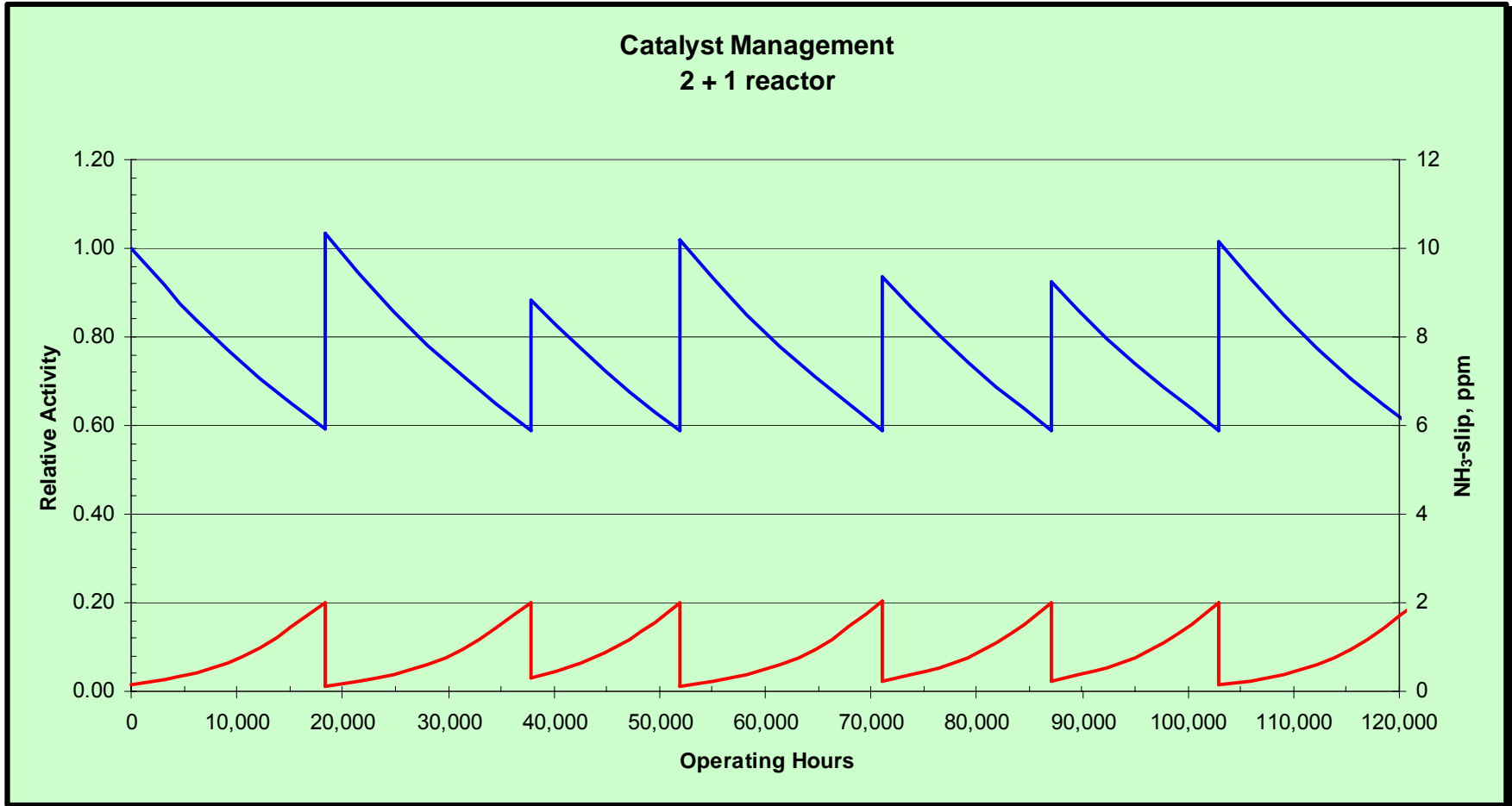
- Surface Analysis
 - Internal surface area (BET) pore volume



	Reference sample		Reactor A		Reactor B	
	A	B	Layer 1	Layer 3	Layer 1	Layer 3
Surface area BET, m ² /g	70	68	68	65	65	64

Catalyst Testing

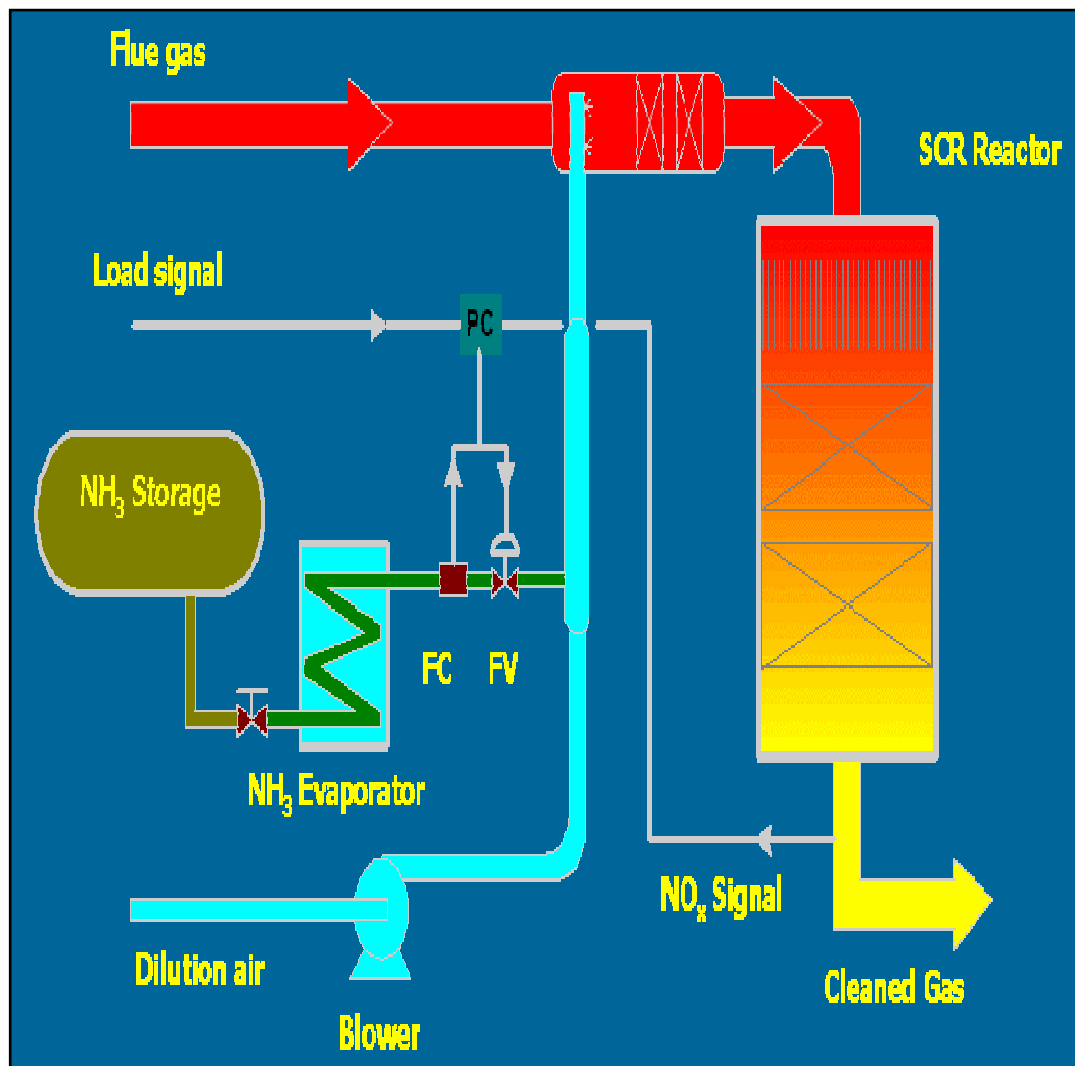
- Reactor specific catalyst management curve
 - To predict future catalyst additions



Inspection / Housekeeping

■ When / what should you inspect?

- At least annually
 - Inlet duct
 - AIG / Mixer
 - Turning vanes
 - Rectifier
 - SCR catalyst
 - Outlet duct
- Take advantage of short outages
- Take a quick look before start up in the spring
- Remove / prevent ash pile build up on catalyst



Catalyst Replacement – Case 1 (New)

- 600 MWe
- 85% operation, 7500 hours
- SCR layout, 2+1
- DeNOx efficiency, 90%
- Ammonia-slip, 2ppmvdc
- SO₂-oxidation, 0.5%

- New catalyst SOR = 1.0
- New catalyst deactivation = 13% per 10,000 hours
- New catalyst cost based on actual 2005 cost

Catalyst Replacement – Case 1 (New)

Catalyst Layers	New	Hours	New Catalyst Cost	Sealing Cost	Freight Cost	Removal Cost	Installation Cost
Spare	X	32,000	\$1,407,375	incl.	\$60,000	\$0	\$250,000
First	X	73,500	\$1,407,375	incl.	\$60,000	\$250,000	\$250,000
Second	X	102,500	\$1,407,375	incl.	\$60,000	\$250,000	\$250,000
Spare	X	136,000	\$1,407,375	incl.	\$60,000	\$250,000	\$250,000
First	X	169,500	\$1,407,375	incl.	\$60,000	\$250,000	\$250,000
Second	X	202,000	\$1,407,375	incl.	\$60,000	\$250,000	\$250,000

Sub - Total \$8,444,250 \$0 \$360,000 \$1,250,000 \$1,500,000

Total Cost over 27 years \$11,554,250

Catalyst Replacement – Case 2 (New / Clean)

- 600 MWe
- 85% operation, 7500 hours
- SCR layout, 2+1
- DeNOx efficiency, 90%
- Ammonia-slip, 2ppmvdc
- SO₂-oxidation, 0.5%

- New catalyst SOR = 1.0
- New catalyst deactivation = 13% per 10,000 hours
- New catalyst cost based on actual 2005 cost

- Clean catalyst SOR = 0.95
- Clean catalyst deactivation = 15 per 10,000 hours
- Damaged catalyst replacement = 10%
- Clean catalyst cost based on actual 2005 cost

Catalyst Replacement – Case 2 (New / Clean)

Catalyst Layers		Hours	Cleaning Catalyst Cost	New Catalyst Cost	Sealing Cost	Freight Cost	Removal Cost	Installation Cost
Spare	N	32,000	\$0	\$1,407,375	incl.	incl.	\$0	\$250,000
First	C	73,500	\$1,367,375	\$140,738	\$50,000	\$100,000	\$250,000	\$250,000
Second	C	125,000	\$1,367,375	\$140,738	\$50,000	\$100,000	\$250,000	\$250,000
Spare	C	151,000	\$1,367,375	\$140,738	\$50,000	\$100,000	\$250,000	\$250,000
First	C	174,500	\$1,367,375	\$140,738	\$50,000	\$100,000	\$250,000	\$250,000
Second	C	196,000	\$1,367,375	\$140,738	\$50,000	\$100,000	\$250,000	\$250,000

Sub - Total \$6,836,875 \$2,111,065 \$250,000 \$500,000 \$1,250,000 \$1,500,000

Total Cost over 27 years

\$12,447,940

Savings over New Catalyst, year =

(\$34,373)

w/o damaged catalyst

\$11,744,250

Savings over New Catalyst, year =

(\$7,308)

Catalyst Replacement – Case 3 (Clean)

- 600 MWe
- 85% operation, 7500 hours
- SCR layout, 2+1
- DeNOx efficiency, 90%
- Ammonia-slip, 2ppmvdc
- SO₂-oxidation, 0.5%

- Clean catalyst SOR = 0.95
- Clean catalyst deactivation = 15% per 10,000 hours
- Damaged catalyst replacement = 10%
- Clean catalyst cost based on actual 2005 cost quotation

Catalyst Replacement – Case 3 (Clean)

Catalyst Layers	Cleaned	Hours	Cleaning Catalyst Cost	New Catalyst Cost	Sealing Cost	Freight Cost	Removal Cost	Installation Cost
Spare	X	32,000	\$1,367,375	\$140,738	\$50,000	\$100,000	\$250,000	\$250,000
First	X	70,000	\$1,367,375	\$140,738	\$50,000	\$100,000	\$250,000	\$250,000
Second	X	95,000	\$1,367,375	\$140,738	\$50,000	\$100,000	\$250,000	\$250,000
Spare	X	123,000	\$1,367,375	\$140,738	\$50,000	\$100,000	\$250,000	\$250,000
First	X	149,000	\$1,367,375	\$140,738	\$50,000	\$100,000	\$250,000	\$250,000
Second	X	176,000	\$1,367,375	\$140,738	\$50,000	\$100,000	\$250,000	\$250,000
Spare	X	194,000	\$1,367,375	\$140,738	\$50,000	\$100,000	\$250,000	\$250,000

Sub - Total \$8,204,250 \$844,428 \$300,000 \$600,000 \$1,500,000 \$1,500,000

Total Cost over 27 years

\$12,948,678

Savings over New Catalyst, year =

(\$53,632)

w/o damaged catalyst

\$12,104,250

Savings over New Catalyst, year =

(\$21,154)

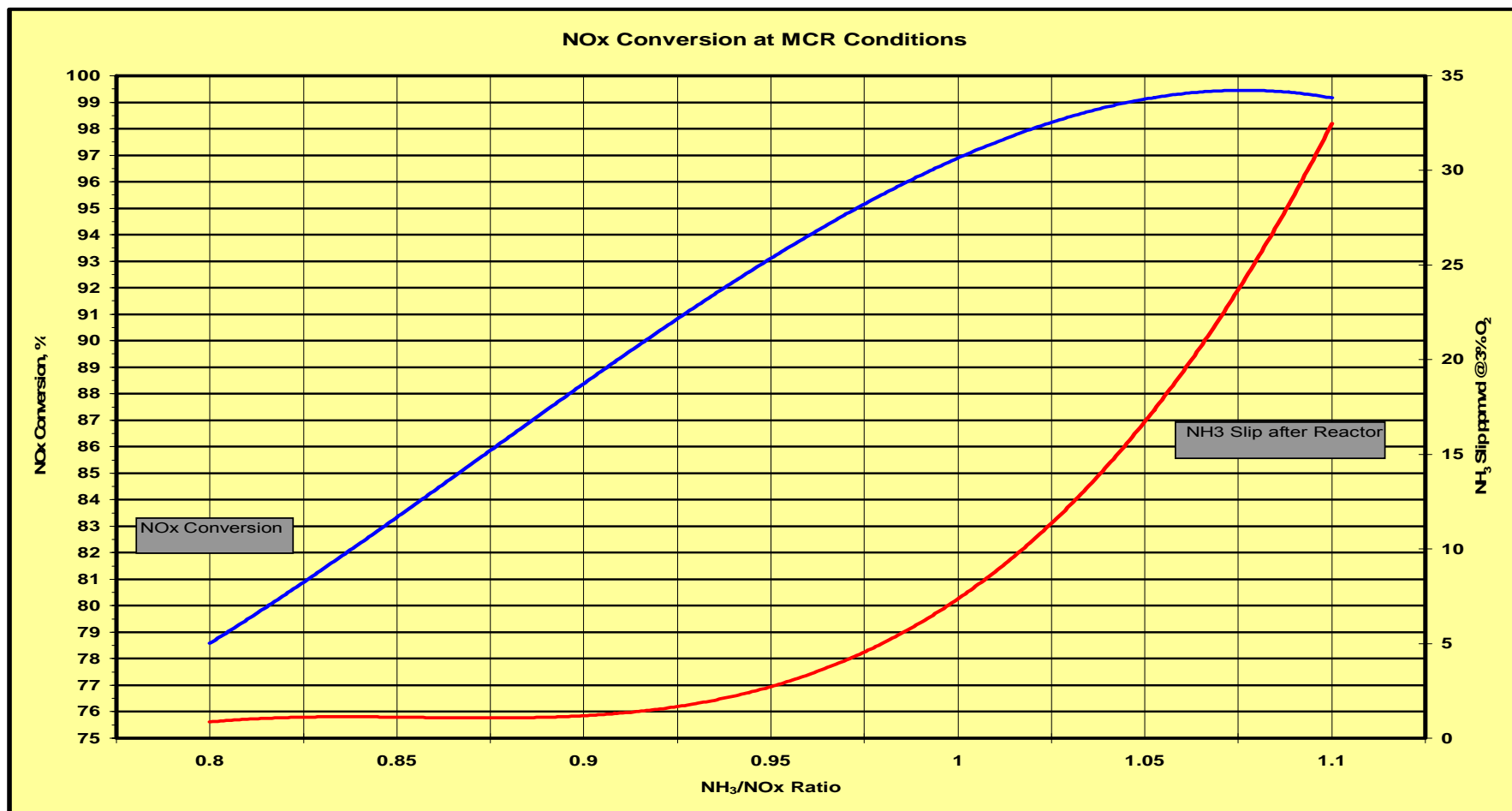
On Line Catalyst Monitoring

Measurements	Units	Daily	Daily	Daily	Daily	Daily
Unit load	MWe					
Flue gas flow	SCFM					
Oxygen in flue gas	% wet					
Flue gas temperature	°F					
NOx inlet	ppm wet					
NOx outlet	ppm wet					
Ammonia consumption	lb/hr					
Ammonia valve position	%					
Dilution air flow	SCFM					
Ammonia slip	ppm wet					
Ammonia in fly ash	ppm wet					
Pressure drop reactor	Inch WG					
Pressure drop 1 st layer	Inch WG					
Pressure drop 2 nd layer	Inch WG					
Pressure drop 3 rd layer	Inch WG					
Pressure drop APH	Inch WG					
Catalyst life	Hours					

On Line Catalyst Monitoring

■ NOx conversion graph

- Data can allow estimate of conversion and slip



On Line Catalyst Monitoring

■ Ammonia slip / Ammonia on fly ash

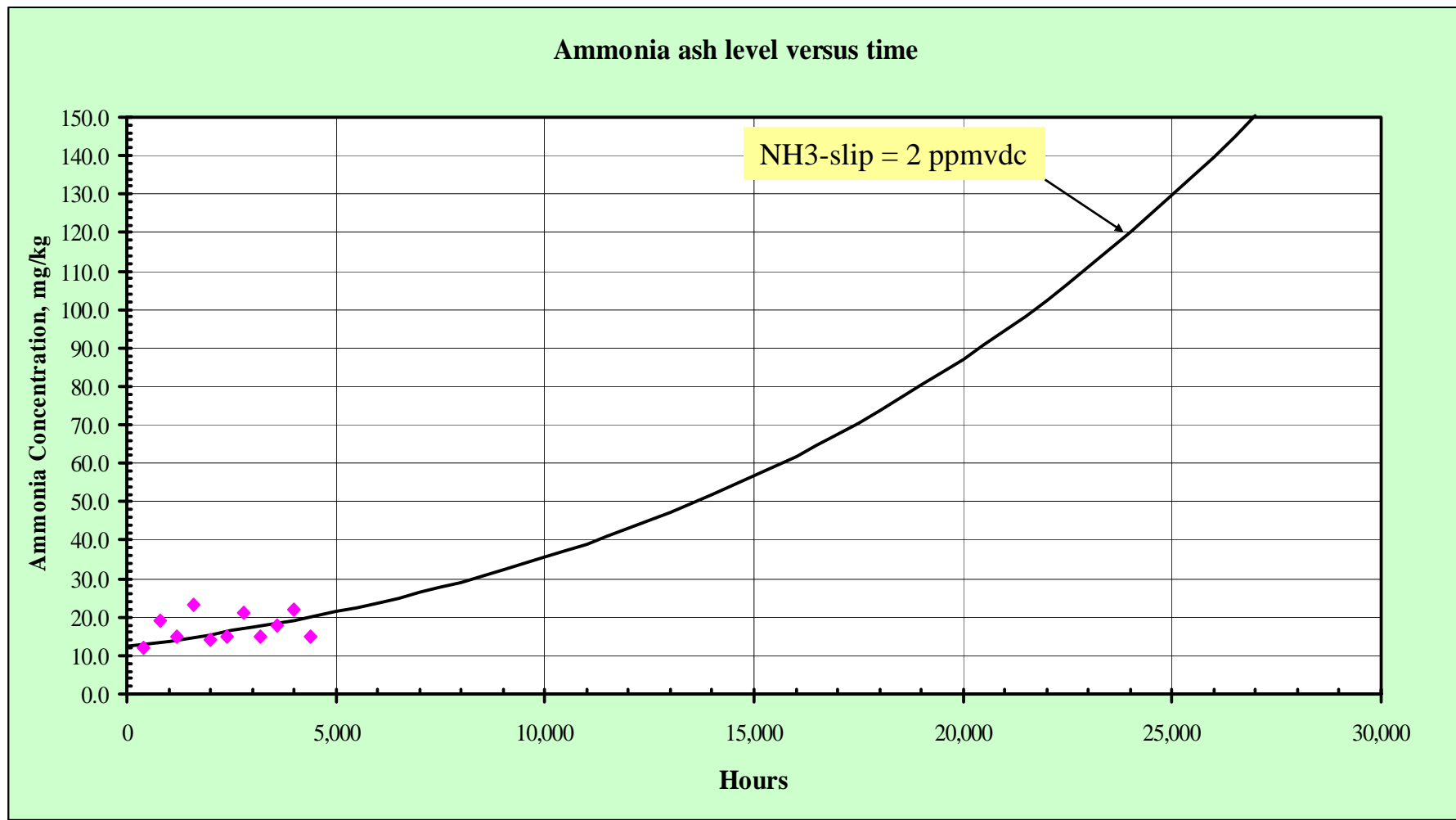
- How does ammonia get on the ash?
 - Acidic ashes attract ammonia
 - SO_3 attracts ammonia
- Without acid sites, ammonia will remain as gas
- Other factors that play a role
 - LOI (especially in PRB coal, very highly ammoniated ash)
 - Structure, mineralogy of ash
 - Local concentrations of NH_3
- Rule of Thumb: 2 ppm slip yields 100 ppm on ash
- ***Ammonia levels in ash are known to be quite variable***

On Line Catalyst Monitoring

- Ammonia slip / Ammonia fly ash sampling
 - Where should I measure?
 - Different ammonia levels in fly ash hoppers
 - Pick the hopper with the highest level
 - How often should I measure?
 - For best results sample hopper once a day
 - Combined samples for once a week ammonia analysis
 - Graph results

On Line Catalyst Monitoring

■ Performance test generated curve



Summary

■ Catalyst may be impacted with a fuel switch

- Ash loading
- ABS
- Catalyst poisons
- SO₂-oxidation

■ Catalyst sampling / analysis

- At least annually – activity, chemical analysis, and pore volume

■ Inspection

- At least once per year
- Quick look in the spring
- Take opportunities

■ Housekeeping

- Keep catalyst clean / free of pluggage

■ Catalyst replacement / cleaning

■ Catalyst On Line Monitoring

- Monitor catalyst / system performance
- Monitor NH_3 -slip
 - Analyzer
 - Fly ash sampling – daily samples combined for weekly analysis

Summary

- Catalyst may be impacted with a fuel switch
 - Ash loading
 - ABS
 - Catalyst poisons
 - SO₂-oxidation
- On line monitoring
 - Monitor catalyst / system performance
 - Ammonia ash sample - daily sample / combined samples for weekly analysis
- Off line monitoring
 - Catalyst sampling / analysis @ least annually – activity, chemical analysis, and pore volume
 - Inspection
 - Complete inspection annually
 - Quick look in the spring
 - Take opportunities
- Housekeeping
 - Keep catalyst clean / free of pluggage

Summary

- Keep care of your investment to prevent this

