

# Reinhold Environmental Ltd.

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2007 NOx Round Table & Expo  
Presentation

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*February 5-6, 2007 in Cincinnati, OH*

## *“Influence of SO<sub>3</sub> Removal on SCR Design and Operation”*

NO<sub>x</sub> Workshop  
Reinhold Environmental  
February 5-6, 2007

Rob Moser  
Jim Wilhelm  
Codan Development LLC

Jim Jarvis  
URS Corporation



# Presentation Outline

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- Impacts Created by Presence of  $\text{SO}_3$
- Benefits of Effective  $\text{SO}_3$  Removal
- Requirements for  $\text{SO}_3$  Technology
- SBS Injection™ Technology Description
- SBS Injection™ Development Programs

# Impacts Created by Presence of SO<sub>3</sub>

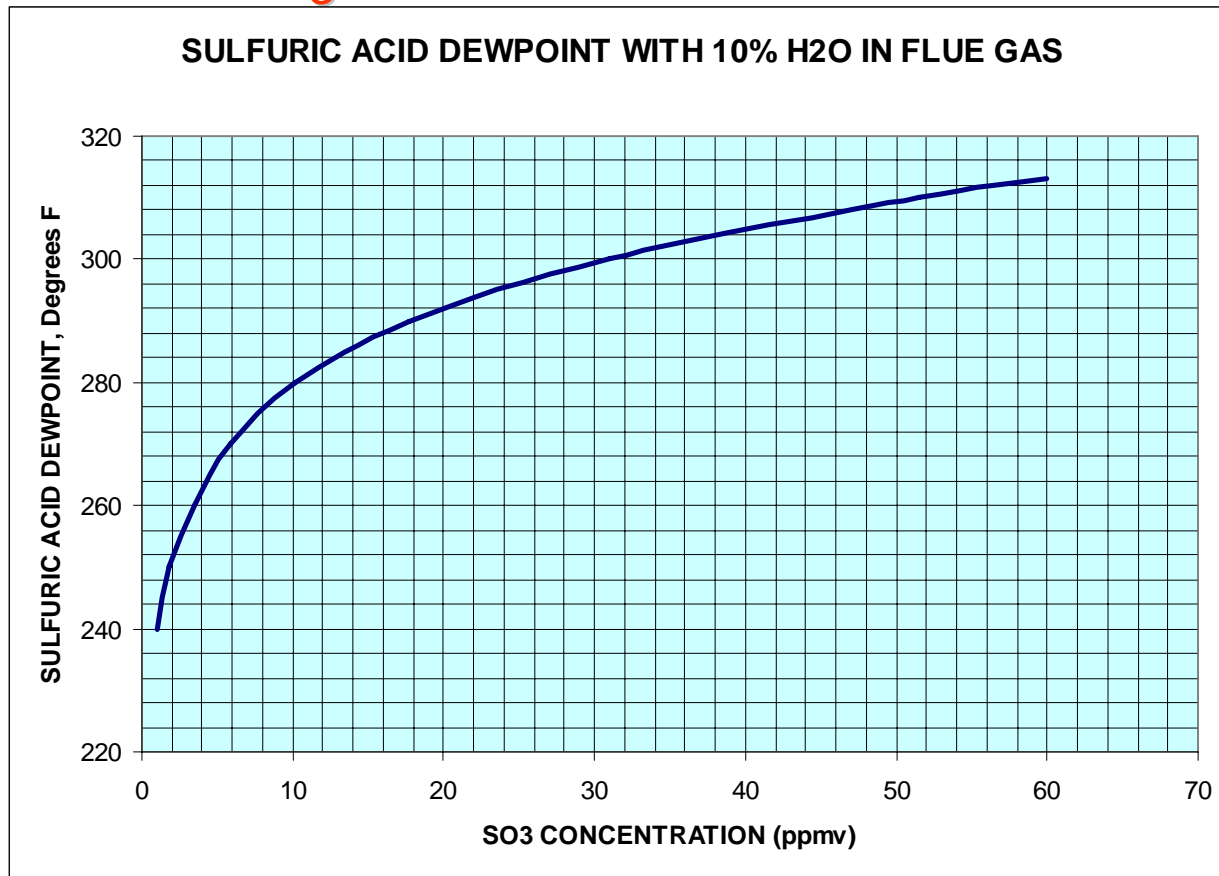
- Plume Opacity Issue: Coloration of Plume from Optical Light Scattering Effects of Sulfuric Acid Aerosol Particles



Sulfuric acid opacity can be visible when SO<sub>3</sub> > 5ppm

# Impacts Created by Presence of SO<sub>3</sub>

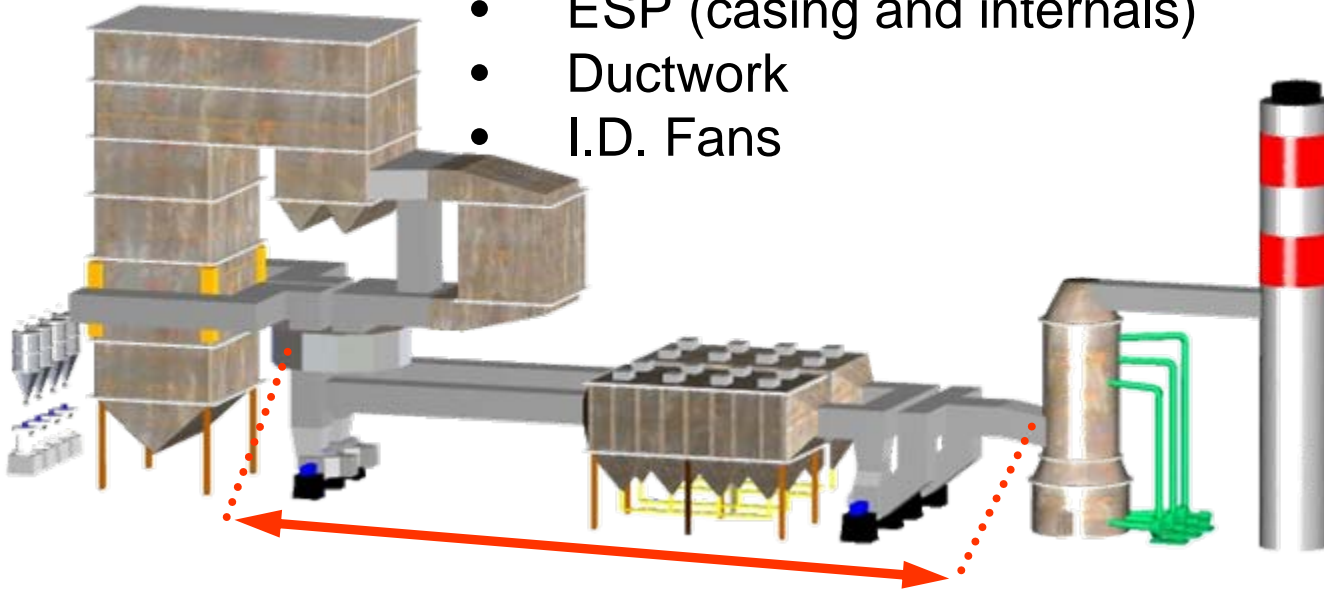
- Plume Effects: Opacity/Buoyancy
- Effect of SO<sub>3</sub> on Sulfuric Acid Dew Point



# Increased SO<sub>3</sub>: Acid Dew Point Implications

## Increased Acid Corrosion of all Back-End Components

- Air heater cold-end baskets
- ESP (casing and internals)
- Ductwork
- I.D. Fans



Increasing Air Heater Outlet Temperature Negatively Affects Heat Rate  
(Change of 35°F = 1% change in heat rate)

# Impacts Created by Presence of SO<sub>3</sub>

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- Plume Effects
  - ✓ Opacity & buoyancy issues
- Acid Dew Point Effects
  - ✓ Corrosion and/or heat rate issues
- Reaction with Ammonia to Form ABS
  - ✓ Occurs as gas cools in air heater (350-420°F)
    - × Plugging/pressure drop issues
    - × NH<sub>3</sub> slip criterion effects SCR design and operations
    - × NH<sub>3</sub> slip issue limits application of SNCR
  - ✓ Occurs within catalyst pores (530-630°F)

# Impacts Created by Presence of SO<sub>3</sub>

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- Plume Effects: Opacity & Buoyancy Issues
- Acid Dew Point Effects: Corrosion/Heat Rate Issues
- Reaction with NH<sub>3</sub>: Air Heater Fouling/SCR Implications
- Reaction with NH<sub>3</sub>: SCR Catalyst Plugging/Deactivation
- **Impacts on Use of Fabric Filters**
  - ✓ Corrosion issue (casing/components/bags)
  - ✓ Effects on filter cake properties
    - × Cake heavier, more sticky, more difficult to remove
    - × Implications on cleaning frequency/pressure drop

# Impacts Created by Presence of SO<sub>3</sub>

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- Plume Effects: Opacity & Buoyancy Issues
- Acid Dew Point Effects: Corrosion/Heat Rate Issues
- Reaction with NH<sub>3</sub>: Air Heater Fouling/SCR Implications
- Reaction with NH<sub>3</sub>: SCR Catalyst Plugging/Deactivation
- Negative Impacts on Use of Fabric Filters
- Adsorption on Fly Ash or Activated Carbon
  - ✓ By occupying these active adsorption sites, reduces ability of flyash/activated carbon to remove mercury

# Impacts Created by Presence of $\text{SO}_3$

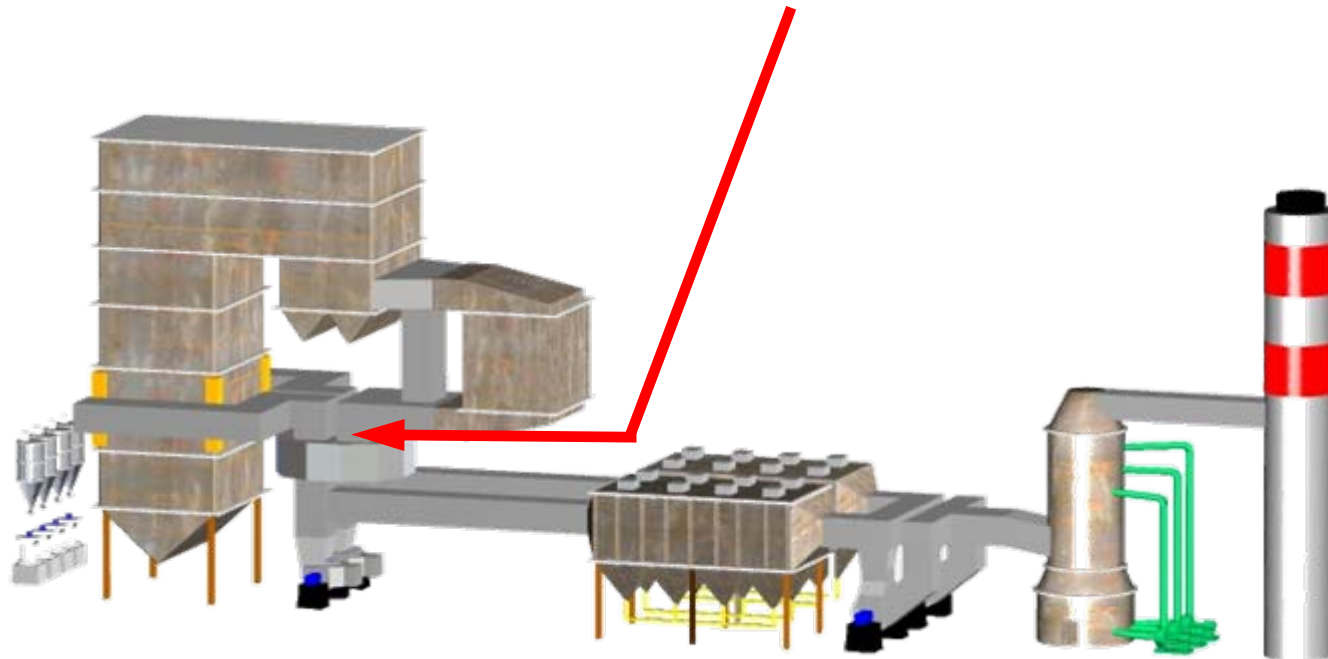
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- Plume Effects: Opacity & Buoyancy Issues
- Acid Dew Point Issues: Corrosion/Heat Rate
- Reaction with  $\text{NH}_3$ : Air Heater/SCR Implications
- Limits SCR Minimum Temp/Affects Catalyst Life
- Negative Impacts on Use of Fabric Filters
- Reduces Hg Retention by Fly Ash/Activated Carbon.
- Improves ESP Particulate Removal Efficiency

# “Effective SO<sub>3</sub> Removal”

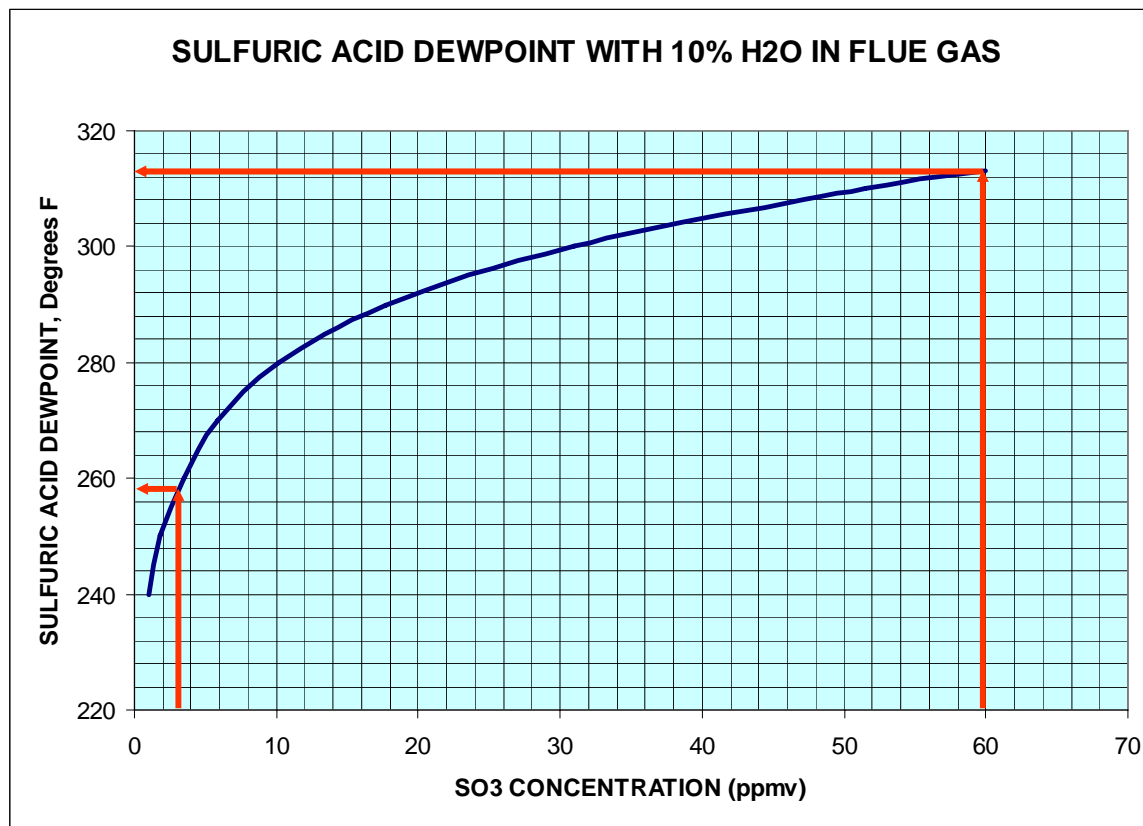
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Definition: Quick, Efficient Removal of SO<sub>3</sub>  
to Less than 3 ppm Entering Air Heater



# O&M Benefits of Effective SO<sub>3</sub> Removal

- Benefits from Reduced Acid Dew Point



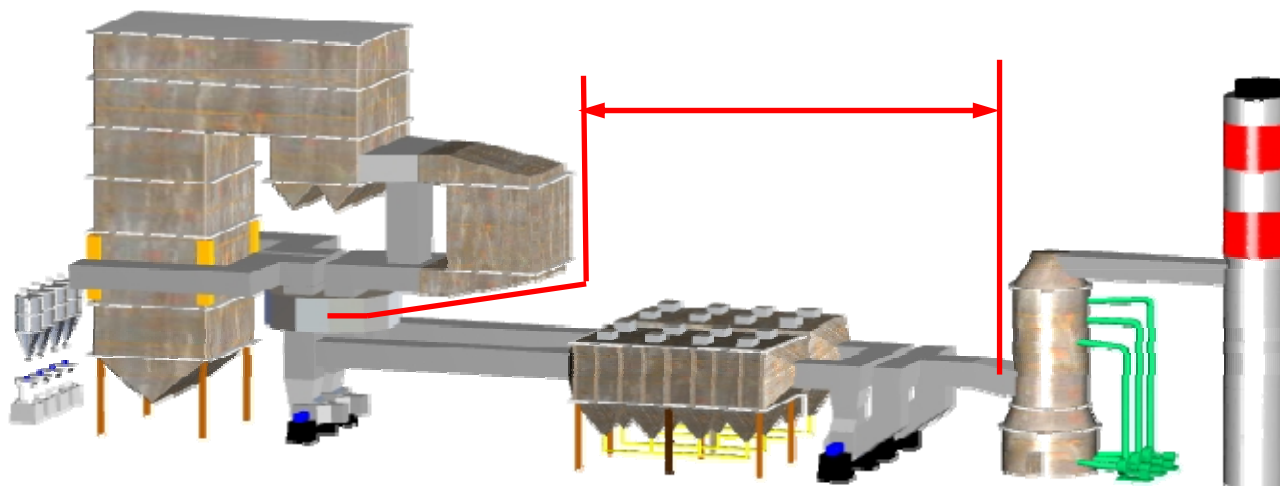
313°F @ 60 ppm

258°F @ 3 ppm

(at 10% H<sub>2</sub>O)

# O&M Benefits of Effective SO<sub>3</sub> Removal

- Benefits from Reduced Acid Dew Point
  - ✓ Reduced sulfuric acid corrosion for entire backend
    - ✗ Reduced dew point = lower potential for corrosion conditions
    - ✗ If conditions do occur, only 1/20 the amount of acid present
    - ✗ Could eliminate need for enamel coating of air heater cold end baskets



***Value Site-Specific: Could Exceed \$500k/yr***

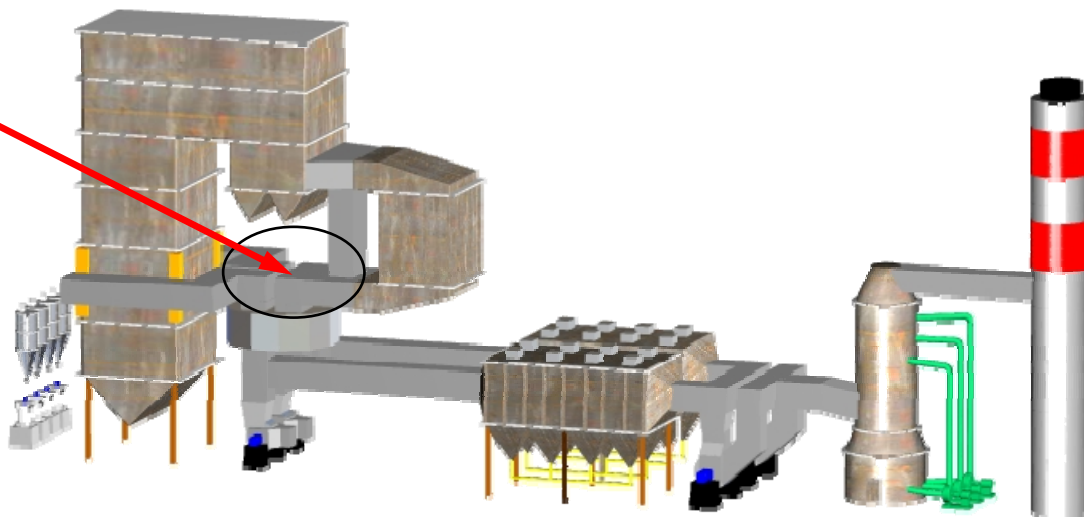
# O&M Benefits of Effective SO<sub>3</sub> Removal

## ● Benefits from Reduced Acid Dew Point

✓ Operate air heater at lower outlet gas temperature

Improve unit heat rate

- Basket modifications likely required
- Potential for \$800k/yr operating cost savings for 500 MW unit



# O&M Benefits of Effective SO<sub>3</sub> Removal

- Benefits from Reduced Acid Dew Point

- ✓ Operate air heater at lower outlet gas temperature

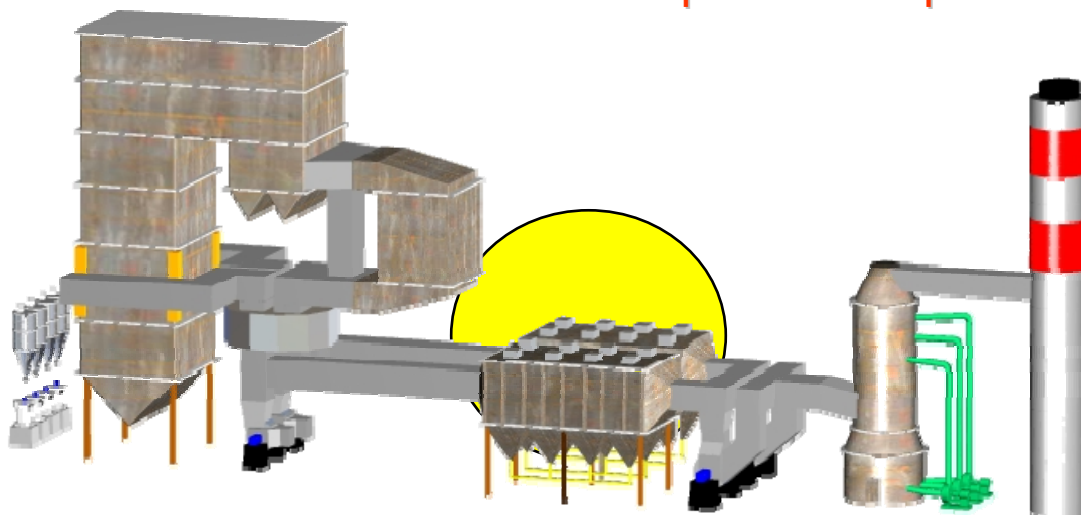
- ✗ Reduce gas volume (from 313F to 258F = ~7% decrease)

Reduced I.D. fan energy consumption

Increased ESP collection efficiency

- Increased specific collection area (SCA)

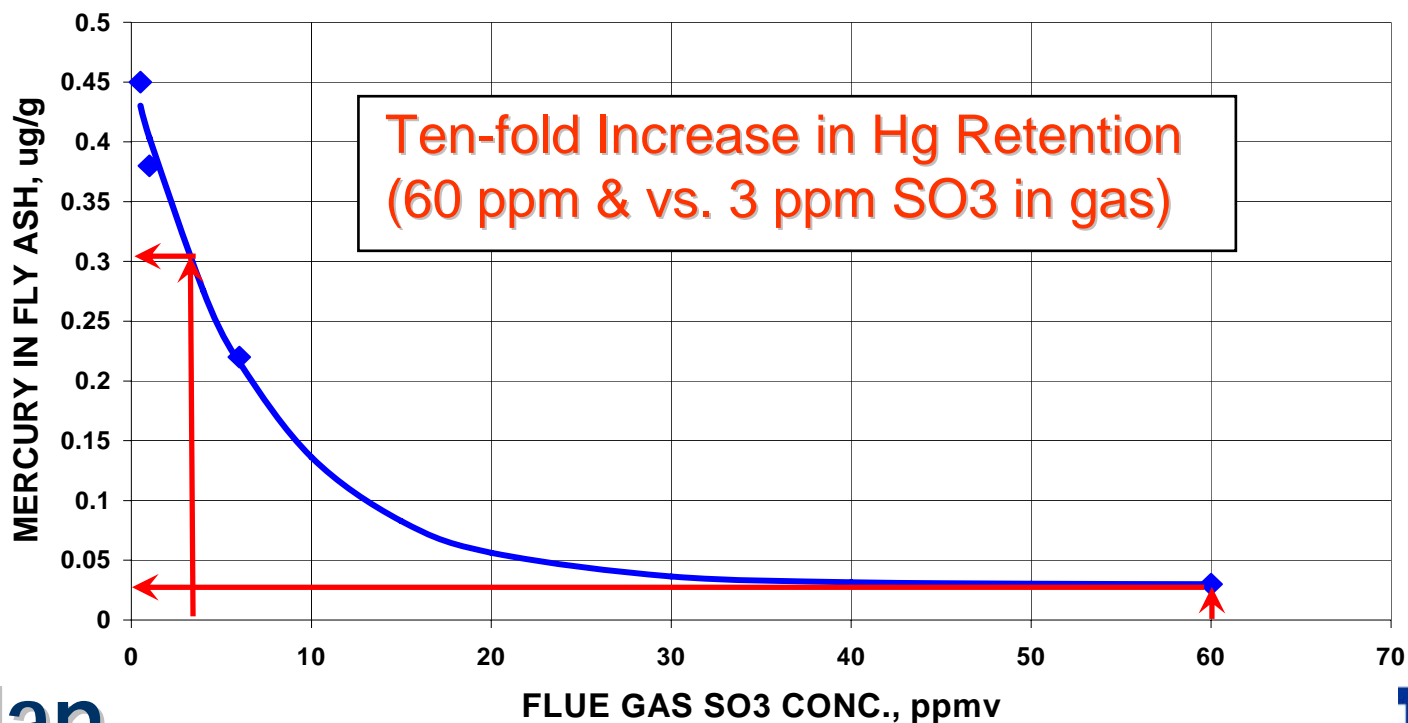
- Reduced temperature improves fly ash resistivity



# O&M Benefits of Effective SO<sub>3</sub> Removal

- With Respect to Mercury Control
  - ✓ Increase retention of mercury on native fly ash

## EFFECT OF SO<sub>3</sub> ON MERCURY COLLECTED WITH FLY ASH



# O&M Benefits of Effective SO<sub>3</sub> Removal

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## ● With Respect to Mercury Control

- ✓ Increase retention of Hg on native fly ash
- ✓ Increase efficiency & improve economics of activated carbon injection (ACI) technology
  - ✗ In absence of SO<sub>3</sub>, carbon will retain more mercury (removal efficiency increased, because carrying capacity increased)
  - ✗ Removal efficiency and carrying capacity both improved at lower temperatures – achievable with SO<sub>3</sub> removed.

# O&M Benefits of Effective SO<sub>3</sub> Removal

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## ● With Respect to Mercury Control

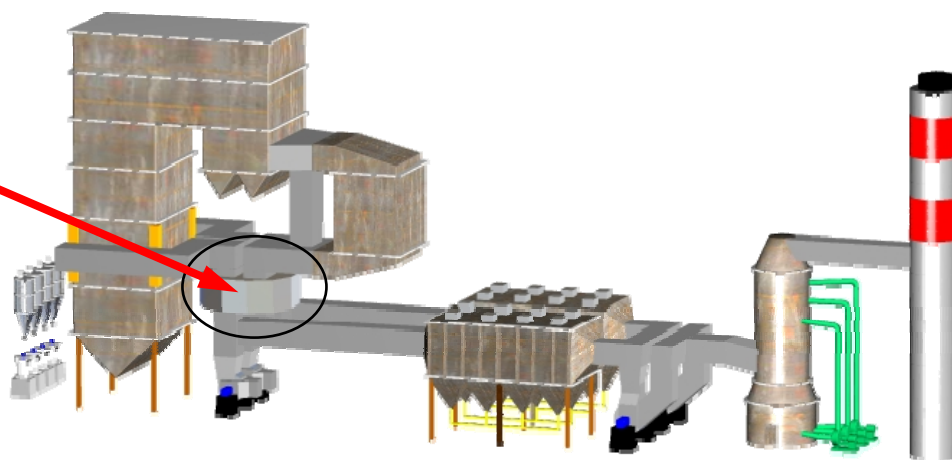
- ✓ Increase retention of mercury on native flyash
- ✓ Increase efficiency & improve economics of activated carbon injection (ACI) technology
  - ✗ In absence of SO<sub>3</sub>, carbon will retain more mercury (both removal efficiency and carrying capacity increased)
  - ✗ Improved carrying capacity of carbon plus lower temperature improve mercury removal efficiency.
  - ✗ **Could reduce amount of carbon required while increasing mercury removal**

# O&M Benefits of Effective SO<sub>3</sub> Removal

- Benefits of Limiting the Formation of Ammonium Bisulfate (ABS)

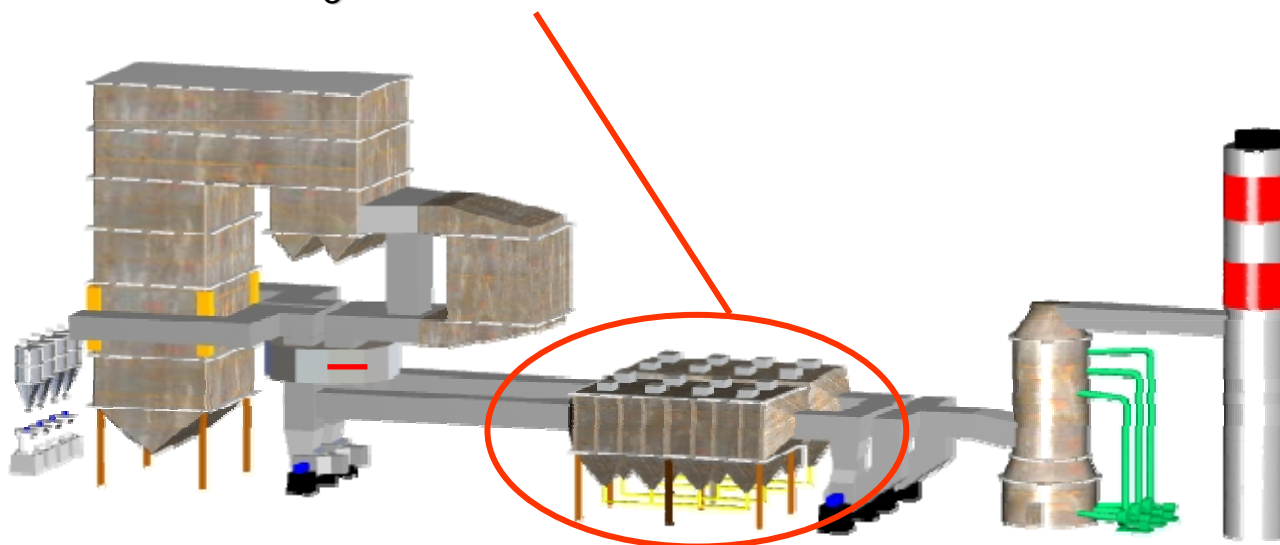
- ✓ Minimize air heater ABS fouling issues

- ✗ With low SO<sub>3</sub> concentrations, driving force for this reaction is small
- ✗ If NH<sub>3</sub> concentration is > 2 times SO<sub>3</sub> concentration, ammonium sulfate formation predominates (non-sticky powder)



# O&M Benefits of Effective SO<sub>3</sub> Removal

- ESP Particulate Removal Efficiency
  - ✓ Lower removal efficiency without condensed SO<sub>3</sub> on ash
  - ✓ Removal efficiency can be restored if sodium reagents are used for SO<sub>3</sub> removal



# Summary: SO<sub>3</sub> Impacts Eliminated

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- Plume Effects: Opacity & Buoyancy Issues
- Acid Dew Point Issues: Corrosion/Heat Rate
- Reaction with NH<sub>3</sub>: Air Heater/SCR Implications
- SCR Minimum Temp/Affects Catalyst Life
- Negative Impacts on Use of Fabric Filters
- Reduced Hg Retention by Fly Ash/Activated Carbon.
- ESP Particulate Removal Efficiency

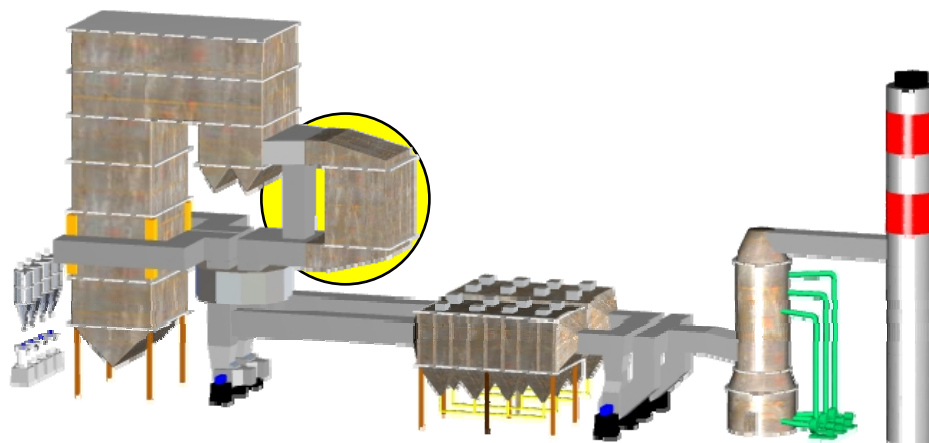
# SCR Improvement Opportunities

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- Reducing the  $\text{SO}_3$  Allows Ammonia Slip to be Raised without Air Heater Fouling
  - ✓ Limit ABS deposition in the air heater
  - ✓ Increased  $\text{NO}_x$  removal & credit generation
  - ✓ Optimized boiler operation (less staging needed)
  - ✓ Increased catalyst management options
  - ✓ Flexibility in  $\text{NH}_3$  distribution equipment design
- Improve SCR Catalyst Design with Less Concern About  $\text{SO}_2$  Oxidation
- Remove Combustion-Formed  $\text{SO}_3$  Upstream of SCR

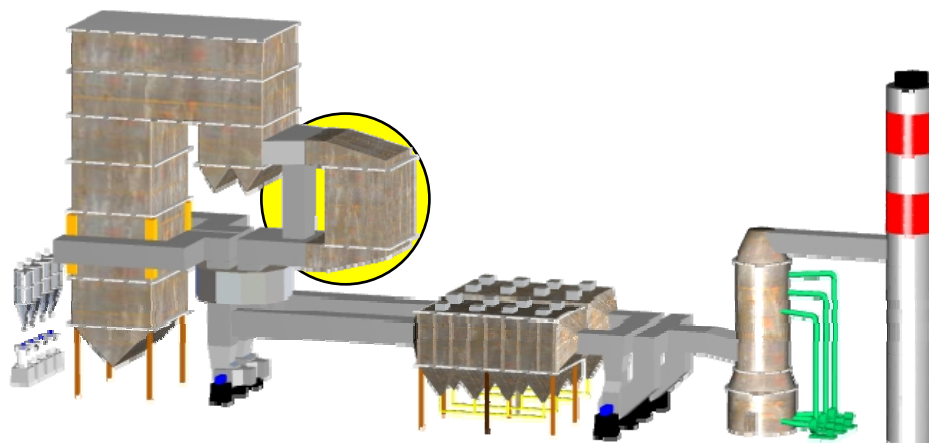
# O&M Benefits of Effective SO<sub>3</sub> Removal

- Benefits from Raising SCR NH<sub>3</sub> Slip
  - ✓ Limit ABS deposition in the air heater
    - × Maintaining 1 to 2 ppm maximum slip no longer critical
    - × Raising slip modestly (4 – 6 ppm) provides design & operating flexibility

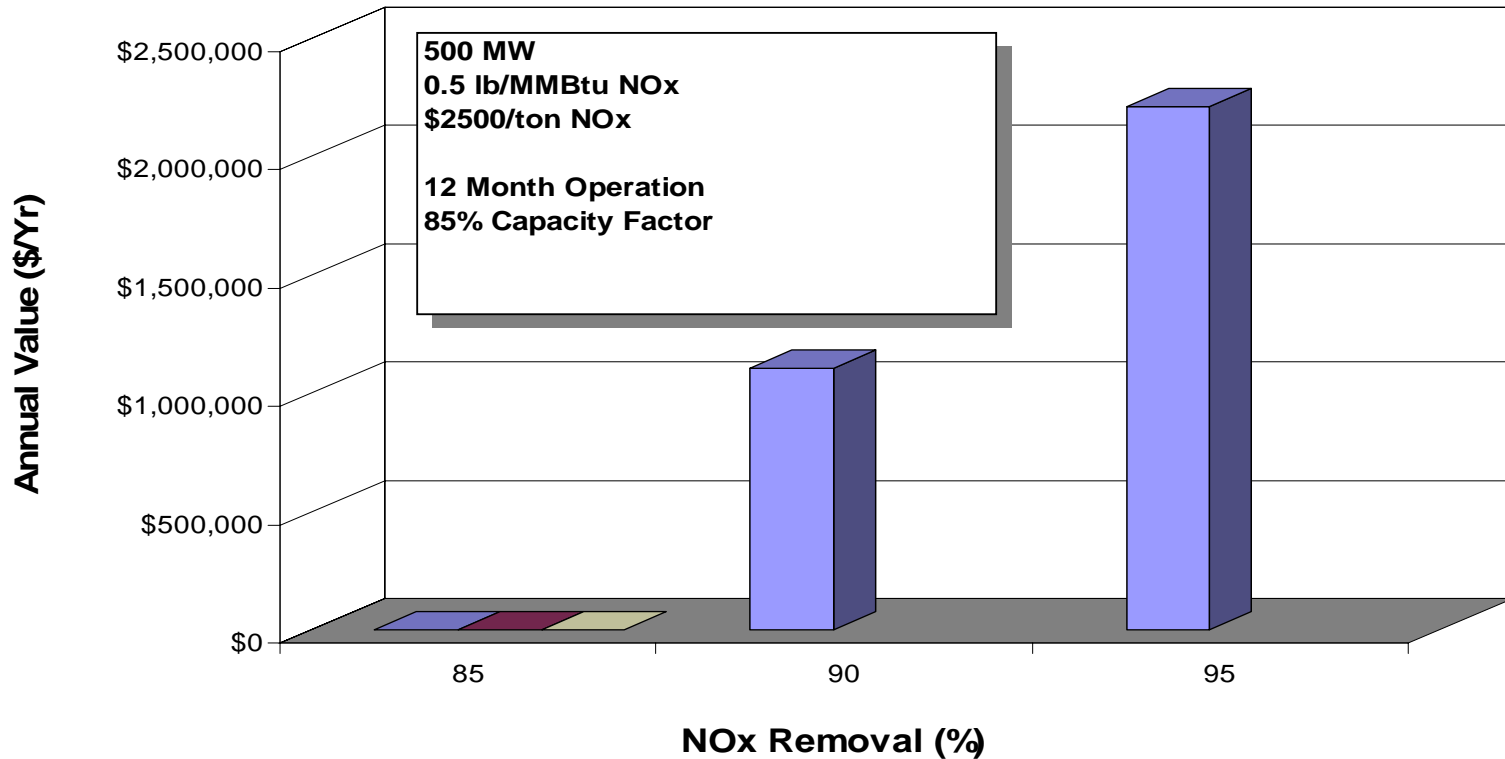


# O&M Benefits of Effective SO<sub>3</sub> Removal

- Benefits from Raising SCR NH<sub>3</sub> Slip
  - ✓ Limit ABS deposition in the air heater
  - ✓ Increase NO<sub>x</sub> removal efficiency & generate credits



# SCR Improvement Analysis – Removal



**Incremental NO<sub>x</sub> Removal Value =  
~ \$220k per 1% Improvement per Year**

*From 2005 EPRI SCR Workshop Proceedings*  
*“Availability of US SCR Fleet” by Clay Erickson*

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- “90% NO<sub>x</sub> Removal Elusive”
- “Low NO<sub>x</sub> Outlet Targets Offer Small Unforgiving Margins”
- Of the 130 units (70,170 MW) Analyzed:

NO <sub>x</sub> Rem'l %	# of Units	% of Units	# of MWs	% of MWs
>90%	18	14	9,165	13
87 -90%	34	26	19,757	28
80 – 86%	44	34	26,205	37
70 – 80%	34	26	15,043	22

# If NH<sub>3</sub> Slip could be increased (from 1-2 ppm to ~6 ppm) without Air Heater ABS Concerns

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	Estimated Additional NO <sub>x</sub> Removed
● 59% (~41,000 MW) could probably increase NO <sub>x</sub> removal by 10% or more	~72,000 tons
● An additional 28% (~20,000 MW) could probably increase NO <sub>x</sub> removal by 5% or more	~18,000 tons
● 13% (~9000 MW) could probably increase NO <sub>x</sub> removal by 2%	~3,000 tons

All units could achieve higher NO<sub>x</sub> removal with greater margins

Increases achieved with no additional SCR capital expenses and a only small increase in operating costs (more NH<sub>3</sub>)

# O&M Benefits of Effective SO<sub>3</sub> Removal

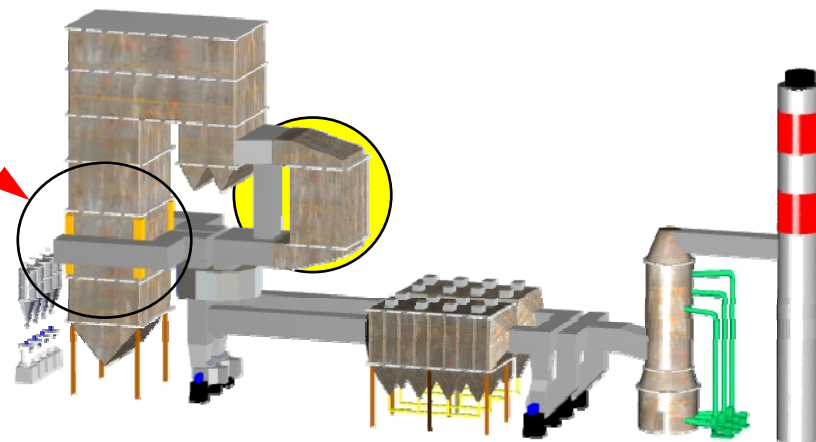
- Benefits from Raising SCR NH<sub>3</sub> slip

- ✓ Limit ABS deposition in the air heater
- ✓ Increase SCR NO<sub>x</sub> removal efficiency
- ✓ Optimize boiler operation

Reduce low-NO<sub>x</sub> combustion staging:  
(Allow SCR to remove more NO<sub>x</sub>)

Reduce fireside corrosion & tube leaks:  
(Forced outage = \$400k - \$600k per occurrence)

Improve heat rate by reduction of unburned carbon:  
(0.2% improvement > \$100/yr benefit for a 500 MW unit)



# O&M Benefits of Effective SO<sub>3</sub> Removal

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- Benefits from Raising SCR NH<sub>3</sub> slip
  - ✓ Limit ABS deposition in the air heater
  - ✓ Increase SCR NO<sub>x</sub> removal efficiency
  - ✓ Optimize boiler operation
  - ✓ Increased catalyst management options
    - × Increase NO<sub>x</sub> removal efficiency at given catalyst activity
    - × Fewer/less frequent catalyst replacements
    - × Extend useful life of catalyst until next planned outage

# O&M Benefits of Effective SO<sub>3</sub> Removal

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- Benefits from Raising SCR NH<sub>3</sub> slip
  - ✓ Limit ABS deposition in the air heater
  - ✓ Increase SCR NO<sub>x</sub> removal efficiency
  - ✓ Optimize boiler operation
  - ✓ Increased catalyst management options
  - ✓ Provide design flexibility with respect to NH<sub>3</sub>:NO<sub>x</sub> distribution
    - ✗ Higher NH<sub>3</sub>/NO<sub>x</sub> COV tolerable for given NO<sub>x</sub> efficiency
    - ✗ Allows more flexibility in NH<sub>3</sub> distribution and mixing equipment design

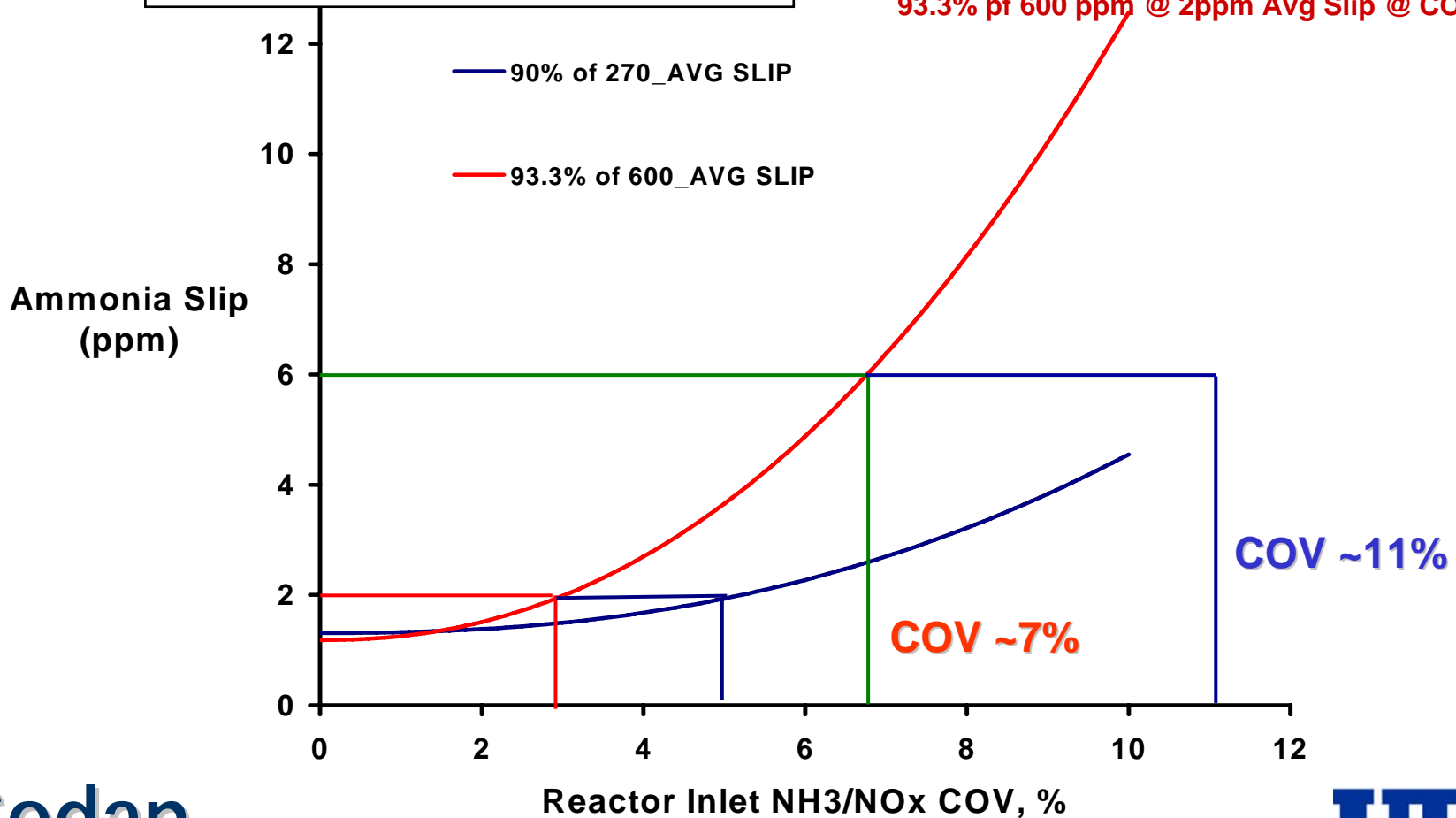
# “Uniformity Requirements and Challenges for Ultra-High SCR DeNOx System Design” by Kevin Rogers, B&W

## Average Ammonia Slip

End-of-Life Design:

90% of 270 ppm @ 2 ppm Avg Slip @ COV = 5%

93.3% pf 600 ppm @ 2ppm Avg Slip @ COV = 3%



# Optimize SCR Catalyst Formulation

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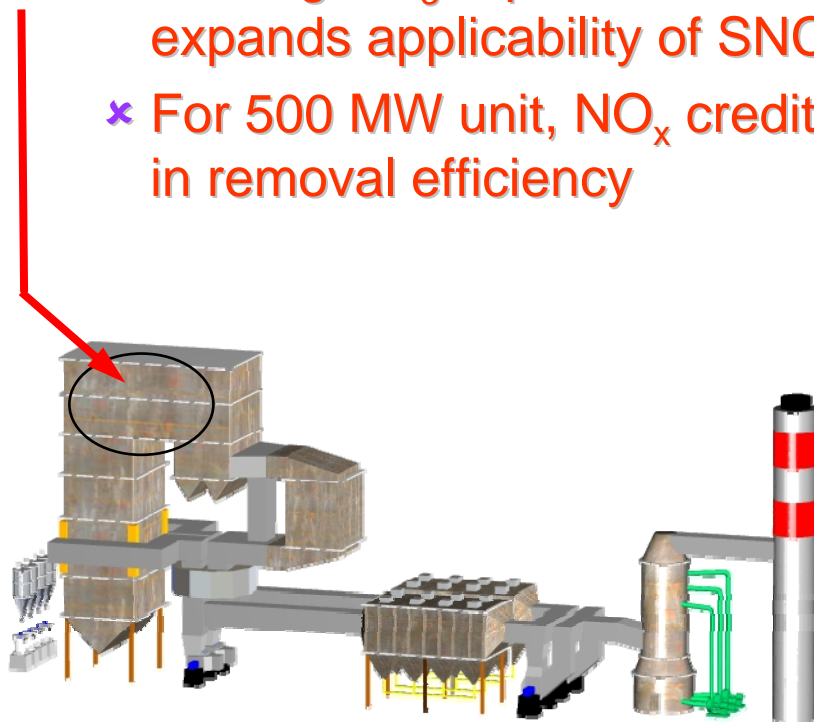
- Treat SO<sub>2</sub> to SO<sub>3</sub> Oxidation as an Economic Variable
  - ✓ Consider using Higher Activity Catalysts (Higher V & Increased TiO<sub>2</sub> Support)
- SCR Design and O&M Options
  - ✓ For new SCR's: Lower catalyst volume or higher NO<sub>x</sub> removal
  - ✓ For existing SCR's: Replacement with higher activity catalyst to improve NO<sub>x</sub> removal efficiency or avoid additional layer
  - ✓ Decouple SO<sub>2</sub> oxidation and mercury oxidation
    - ✗ Facilitate development of high NO<sub>x</sub> removal, high mercury oxidation catalysts
  - ✓ Synergy of high activity catalyst with increased NH<sub>3</sub> slip

# O&M Benefits of Effective SO<sub>3</sub> Removal

- Benefits from Raising NH<sub>3</sub> Slip

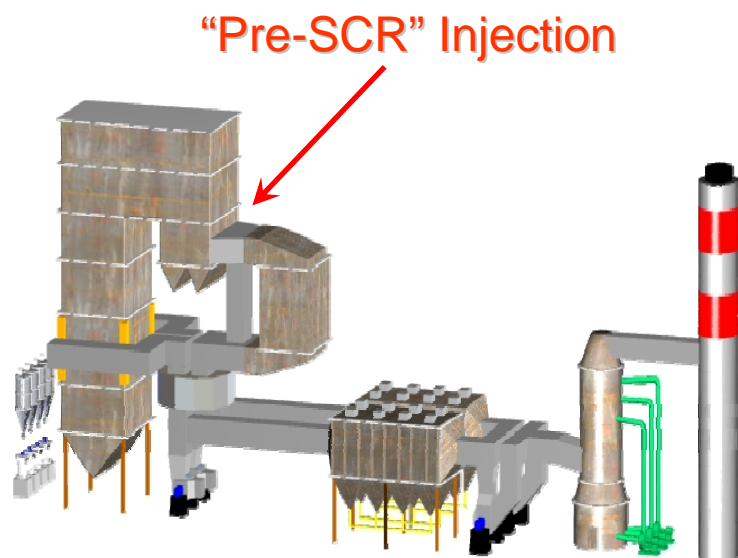
- ✓ With respect to SNCR technology

- ✗ Raising NH<sub>3</sub> slip increases NO<sub>x</sub> removal efficiency and expands applicability of SNCR to higher sulfur coals
- ✗ For 500 MW unit, NO<sub>x</sub> credit value is ~\$220k per % increase in removal efficiency



# O&M Benefits of Effective SO<sub>3</sub> Removal

## Remove Combustion-Formed SO<sub>3</sub> Upstream of the SCR



- ✓ Can allow lower Minimum Operating Temperature (MOT)
  - ✗ Reduce/eliminate formation of ABS within catalysts pores
  - ✗ Allow operation at lower loads without catalyst fouling
  - ✗ Reduced reliance on economizer bypass and associated maintenance & heat rate impacts
  - ✗ Reduced O&M costs

# Requirements of SO<sub>3</sub> Control Technology

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- Must Consistently Achieve High SO<sub>3</sub> Removal Efficiencies – Certainly >90%; Likely >95%
- Must Operate with Very High Reliability
- Must Not Negatively Impact Balance-of-Plant Operations
  - ✓ SCR (if technology used upstream of SCR)
  - ✓ Ductwork
  - ✓ Air heaters
  - ✓ ESP (must compensate for loss of “SO<sub>3</sub> conditioning”)

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- Fate of Excess NH<sub>3</sub> Needs Resolution

# Fate of Excess Ammonia

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- Will the Ammonia Form an Ammonium Sulfate Particle and be Collected in ESP?
  - ✓ If unreacted  $\text{SO}_3$  and  $\text{NH}_3$  are present, they will react to form particles or condense on ash.
  - ✓ Without  $\text{SO}_3$  present,  $\text{NH}_3$  will most likely remain in gas phase with little being collected in the ESP

# Fate of Excess Ammonia

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- Will the Ammonia Form an Ammonium Sulfate Particle and be Collected in ESP?
- Will the Ammonia that Passes through the ESP be Absorbed in the FGD System?
  - ✓ Yes
  - ✓  $\text{NH}_3$  would provide cations that increase sulfite concentration for scrubbers with inhibited oxidation, which will enhance  $\text{SO}_2$  removal efficiency.
  - ✓  $\text{NH}_3$  would cycle up and be washed out of the gypsum in LSFO scrubbers like chlorides.
  - ✓ FGD liquid blowdown or waste sludge treatment must account for  $\text{NH}_3$ .

# Requirements of SO<sub>3</sub> Control Technology

---

- Must Consistently Achieve High SO<sub>3</sub> Removal Efficiencies – Certainly >90%; Likely >95%
- Must Operate with Very High Reliability
- No Negative Balance-of-Plant Impacts
  - ✓ SCR (if technology used upstream of the SCR)
  - ✓ Ductwork
  - ✓ Air heaters
  - ✓ ESP (must compensate for loss of “SO<sub>3</sub> conditioning”)
  - ✓ Fate of Excess NH<sub>3</sub>
- SO<sub>3</sub> Removal Efficiency and Point of Removal are Vital Parameters to Achieve Benefits

# Potential Value of SO<sub>3</sub> Removal

- Benefits could be in the \$5-10 M/yr Range for a 500 MW Unit
- Costs Involved for SO<sub>3</sub> Control – Both Capital and Operating Costs
- Benefits Involve Additional Costs: Air Heater Modifications & NH<sub>3</sub> Costs
- Potential Benefits Many Times the Costs





# Presentation Outline

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- Impacts Created by Presence of  $\text{SO}_3$
- Benefits of Effective  $\text{SO}_3$  Removal
- Requirements for  $\text{SO}_3$  Technology
- SBS Injection™ Technology Description
  - ✓ Commercial installations
  - ✓ Opacity reduction
- SBS Injection™ Development Programs
  - ✓ Reagent testing
  - ✓ Mercury reduction
  - ✓ “Pre-SCR” reagent injection

# Technology Description

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- Injection of a Sodium Alkali Solution
- Injection Typically between SCR and ESP
  - ✓ Upstream or downstream of the air heater
- Solution is Flash Dried in Flue Gas
- Reaction between  $\text{SO}_3$  and Dried Particles
- Dry Reaction Products Collected in ESP/FF
- High  $\text{SO}_3$  Removal at Low Injection Rates

# Commercial Installations

Utility	Plant	Location	MW	Design SO <sub>3</sub>	Injection Location	Reagent	Operating Season	Startup Date
FirstEnergy	Mansfield 1-3	Shippingport, PA	3x860	80	Air Heater Inlet	Sodium Sulfite Solids	Year-round	March 2003
TVA	Widows Creek 7	Stevenson, AL	550	54	Air Heater Inlet	Sodium Sulfite Solution	Ozone	April 2003
NIPSCO	Bailly 8	Porter, IN	365	59	Air Heater Outlet	Sodium Carbonate Solution	Ozone	April 2004
Vectren	Culley 3	Newburgh, IN	287	48	Air Heater Inlet	Byproduct SBS Solution	Year-round	July 2004
PPL	Montour 1-2	Washingtonville, PA	2x765	42	Air Heater Outlet	Sodium Carbonate Solids	Ozone	Aug 2004
Cinergy	Gibson 1-5	E. Mt. Carmel, IN	5x650	110	Air Heater Outlet	Sodium Carbonate Solids	Ozone	May 2005

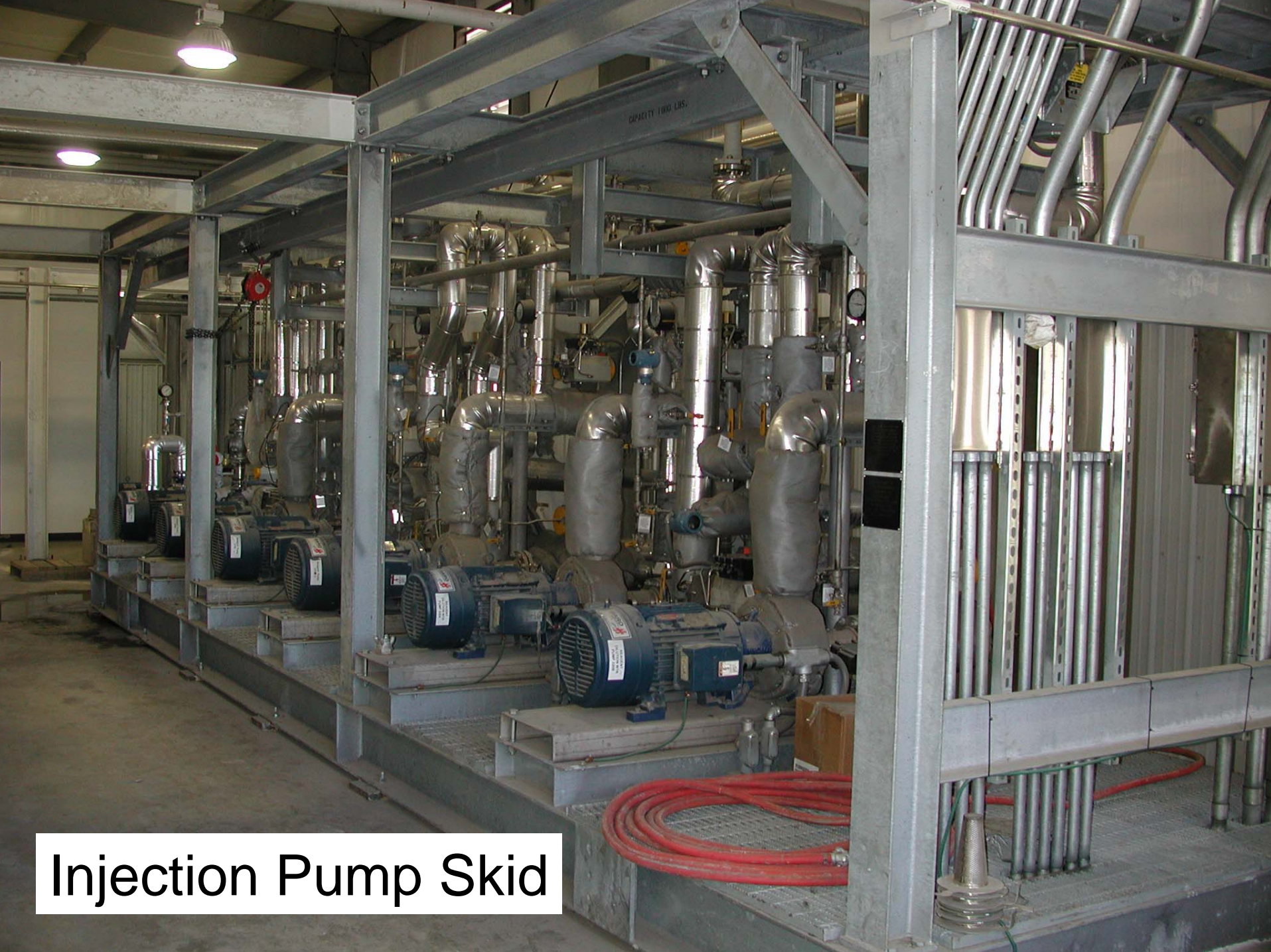
**Over 8500 MW Installed and Operating**

# Duke Energy Gibson Generating Station



# Reagent Storage Tanks and Pump Bldg





**Injection Pump Skid**



**Injection Metering Skids**



**Injection Lance System**

# Benefits of “Upstream” Injection

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- Reduced AH Fouling
  - ✓ Reduced sulfuric acid & ABS deposition
- Improved Heat Rate
  - ✓ Lower air heater outlet temperature & fan power
- Enhanced SCR Performance
  - ✓ NO<sub>x</sub> removal
  - ✓ Design & operating flexibility
  - ✓ Catalyst life & management strategy
- Enhanced Hg Removal
  - ✓ Lower flue gas temperature
  - ✓ Elimination of SO<sub>3</sub>

# Technology Development Activities

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- Pilot Reagent Evaluation Test Program
- Pilot “Enhanced” Mercury Removal Test
- Pilot “Pre-SCR” SBS Injection Evaluation
- Pilot Baghouse Demonstration
- Single-Fluid Atomization Demonstration

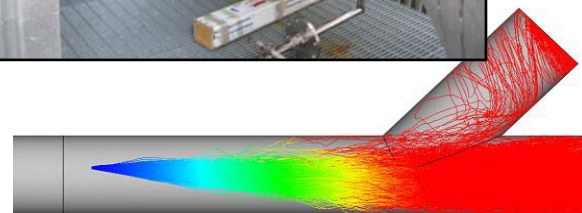
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- Pilot Reagent Evaluation Test Program
- Pilot “Enhanced” Mercury Removal Test
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- Pilot Baghouse Demonstration
- Single-Fluid Atomization Demonstration

# Pilot Reagent Test Program Overview

- Evaluate Alternative Reagents at “Upstream” Injection Location
  - ✓ Sodium sulfite
  - ✓ Sodium carbonate
  - ✓ Sodium hydroxide
  - ✓ Sodium thiosulfate
  - ✓ Dissolved iron
- Southern Company Plant Crist Mercury Research Center (~5 MW)
- SO<sub>3</sub> Removal versus Residence Time & Molar Ratio
- Mercury Removal versus SO<sub>3</sub>
- September/October 2006



# Project Participants

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- Indianapolis Power & Light
- FirstEnergy
- Hoosier Energy
- Southern Company/Gulf Power
- EPRI
- Southern Research Institute
- Southern Ionics
- Solvay

# Commercial Installations

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# Relative Reagent Costs

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<b>Reagent</b>	<b>Cost per Dry Ton, Delivered</b>	<b>Sodium Cost (Relative to Sodium Sulfite)</b>
Sodium Sulfite	\$350	1.00
Soda Ash	\$225	0.54
Trona (T-200)	\$150	0.52
Trona (T-50)	\$125	0.45
Caustic	\$350	0.63
Sodium Thiosulfate	\$350	1.25

- Most of the Tested Reagents would Reduce Operating Costs Relative to Sodium Sulfite by ~1/2
- Costs would be Lower for Sodium Thiosulfate Provided from Waste or Byproduct Streams

# Pilot Reagent Test Program Objectives

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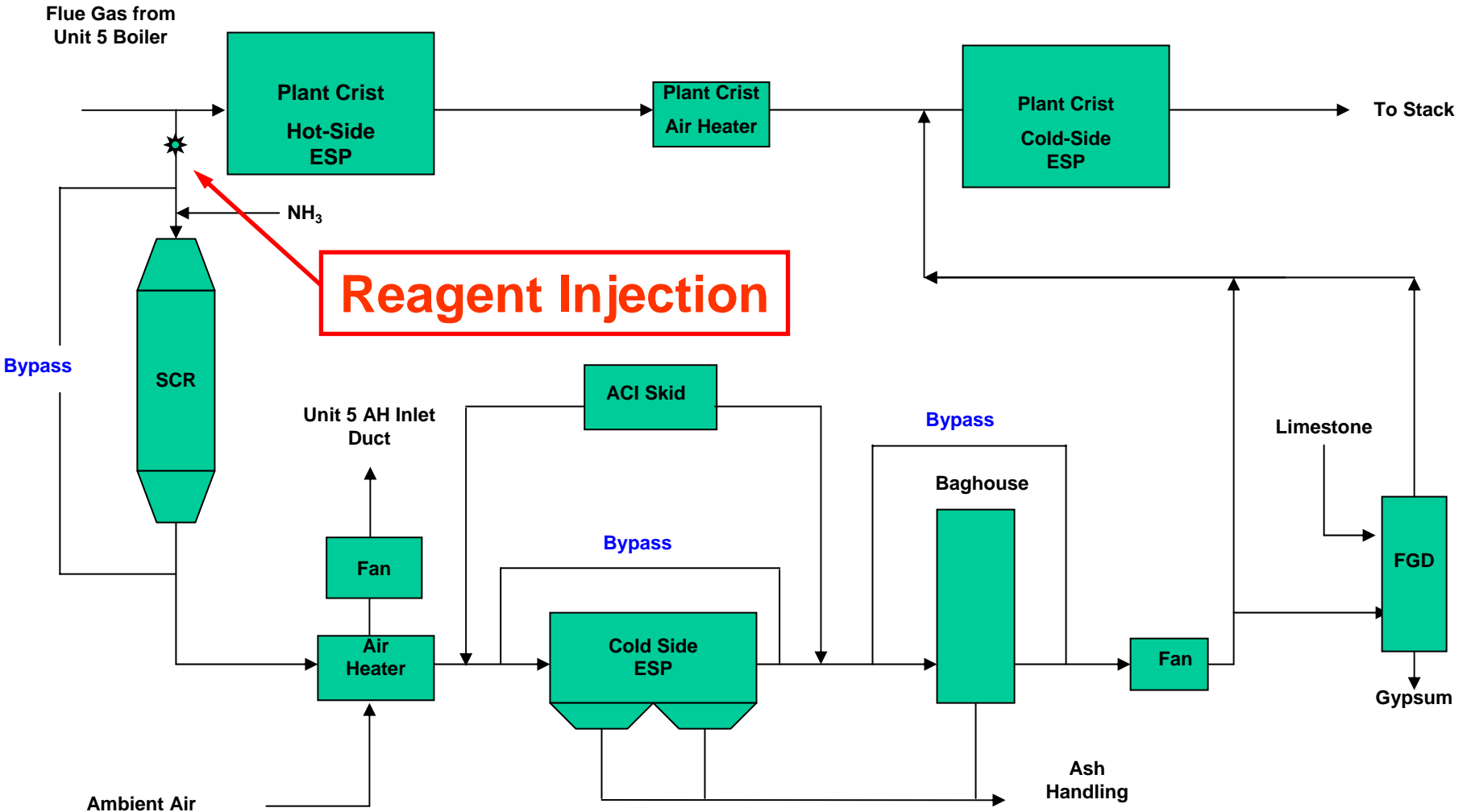
- Quantify Performance of Sodium Carbonate (Soda Ash) and Other Potential Reagents for “Upstream” Injection
  - ✓ Increase the number of reagents that can provide “upstream” injection benefits
  - ✓ Develop lower-cost reagent alternatives to reduce operating costs
- Collect Performance Data to Improve Process Performance Estimates
- Evaluate Effect of Varying  $\text{SO}_3$  Concentration on Mercury Removal with the Native Fly Ash

# Project Approach

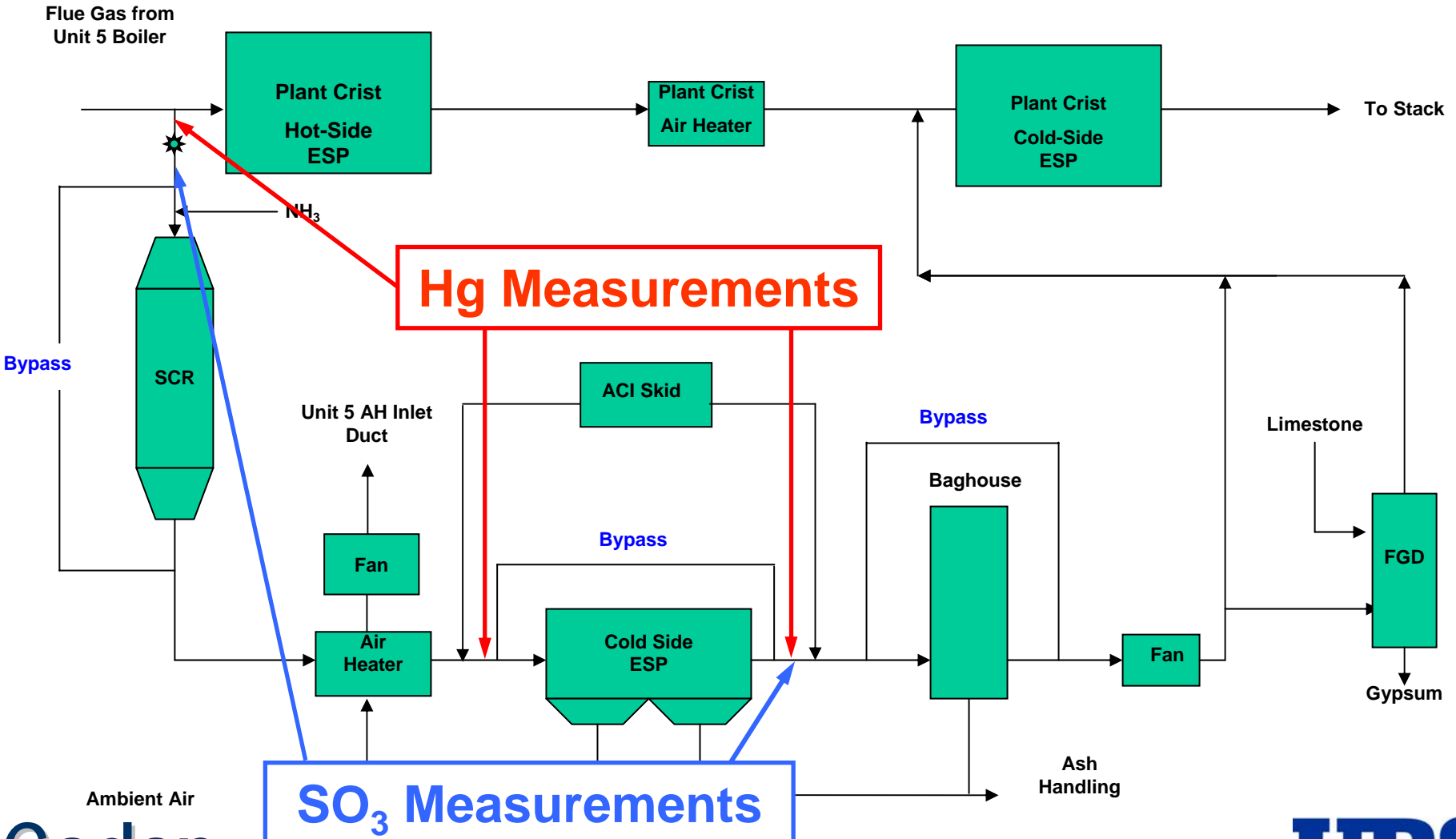
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- Testing Performed at the Mercury Research Center (MRC)
  - ✓ Unit 5 at Southern Company's Plant Crist
  - ✓ ~ 5 MW equivalent size
  - ✓ 0.5 to 1.0% sulfur coal
- Pilot Unit Equipped with SO<sub>2</sub>/SO<sub>3</sub> Injection
  - ✓ Significant control over these species
- Single, Commercial-Size Injection Nozzle
  - ✓ Test ports for SO<sub>3</sub> with varying residence times downstream of the injection location (1 – 4 seconds)
  - ✓ Mercury measurements at MRC facility inlet and at EPS Inlet/Outlet

# MRC Pilot Equipment Configuration

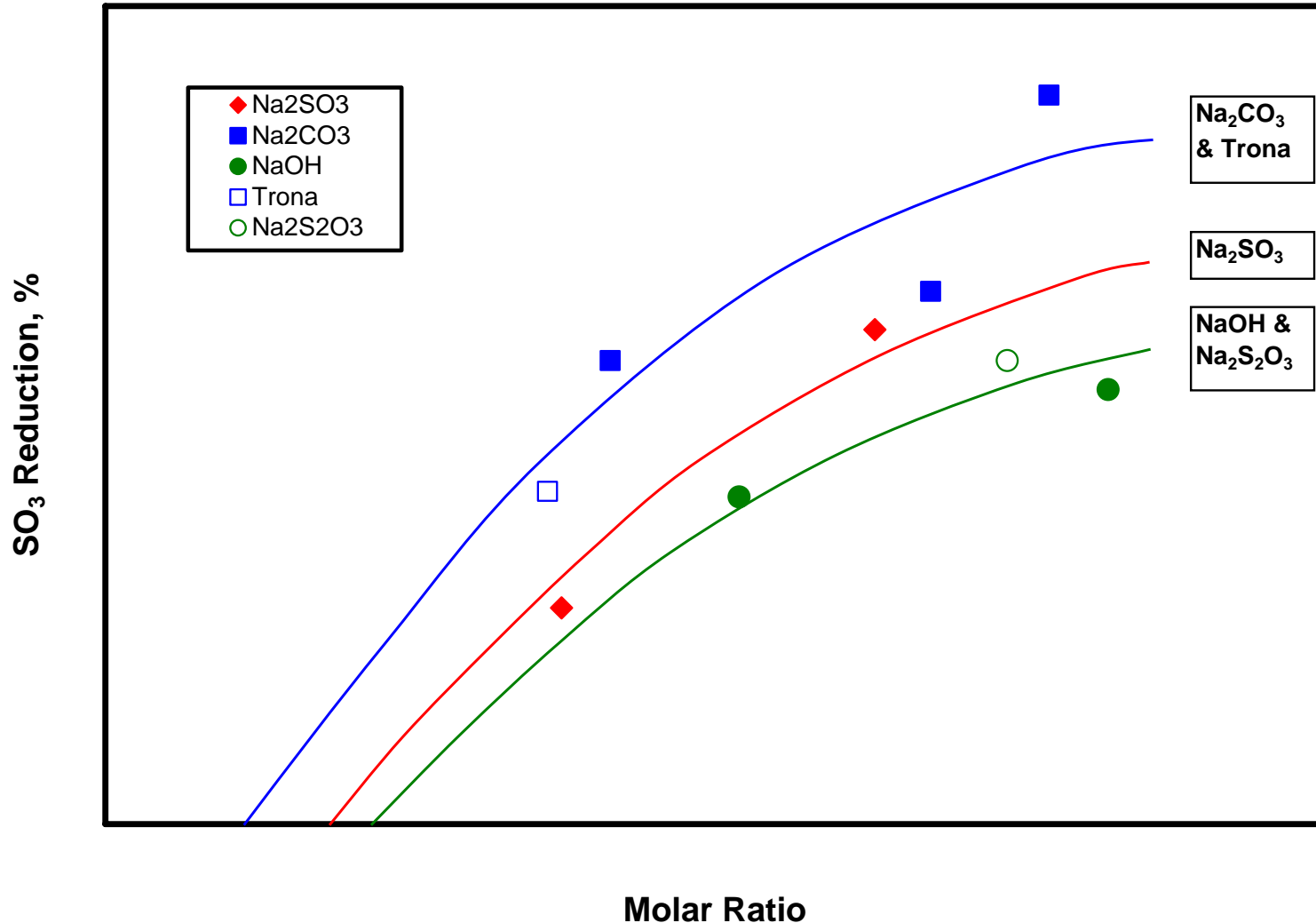


# MRC Pilot Equipment Configuration



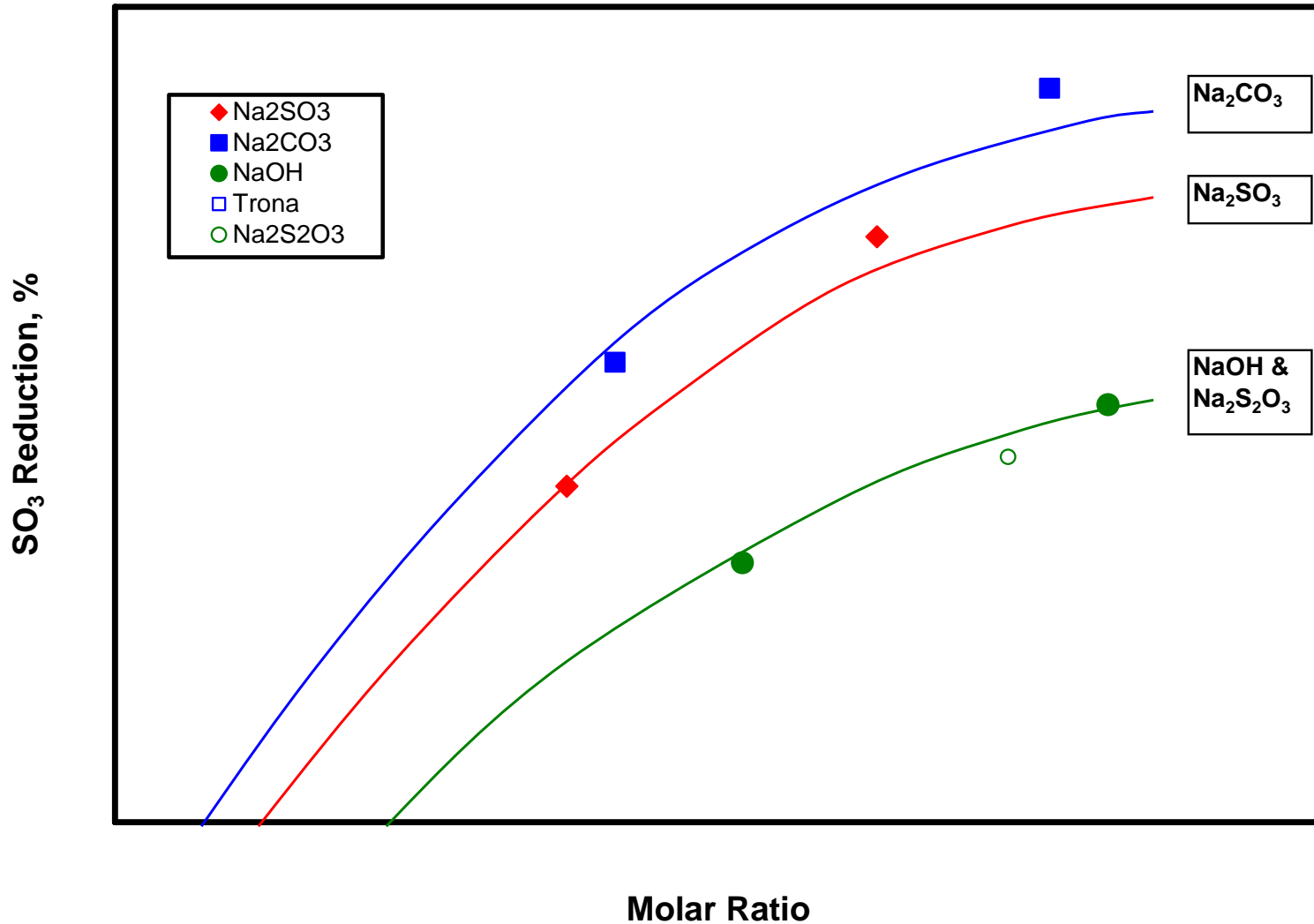
# SO<sub>3</sub> Reduction vs Molar Ratio

(Residence Time = 1.0 Seconds)



# SO<sub>3</sub> Reduction vs Molar Ratio

(Residence Time = 1.4 Seconds)



# Reagent Testing Results Summary

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- All of the Tested Reagents were Effective in Removing  $\text{SO}_3$
- $\text{SO}_3$  Removal Efficiencies with Soda Ash were As Good or Better than Sodium Sulfite
- “Upstream” Benefits can be Obtained with Lower Reagent Costs

# Mercury Removal Testing Approach

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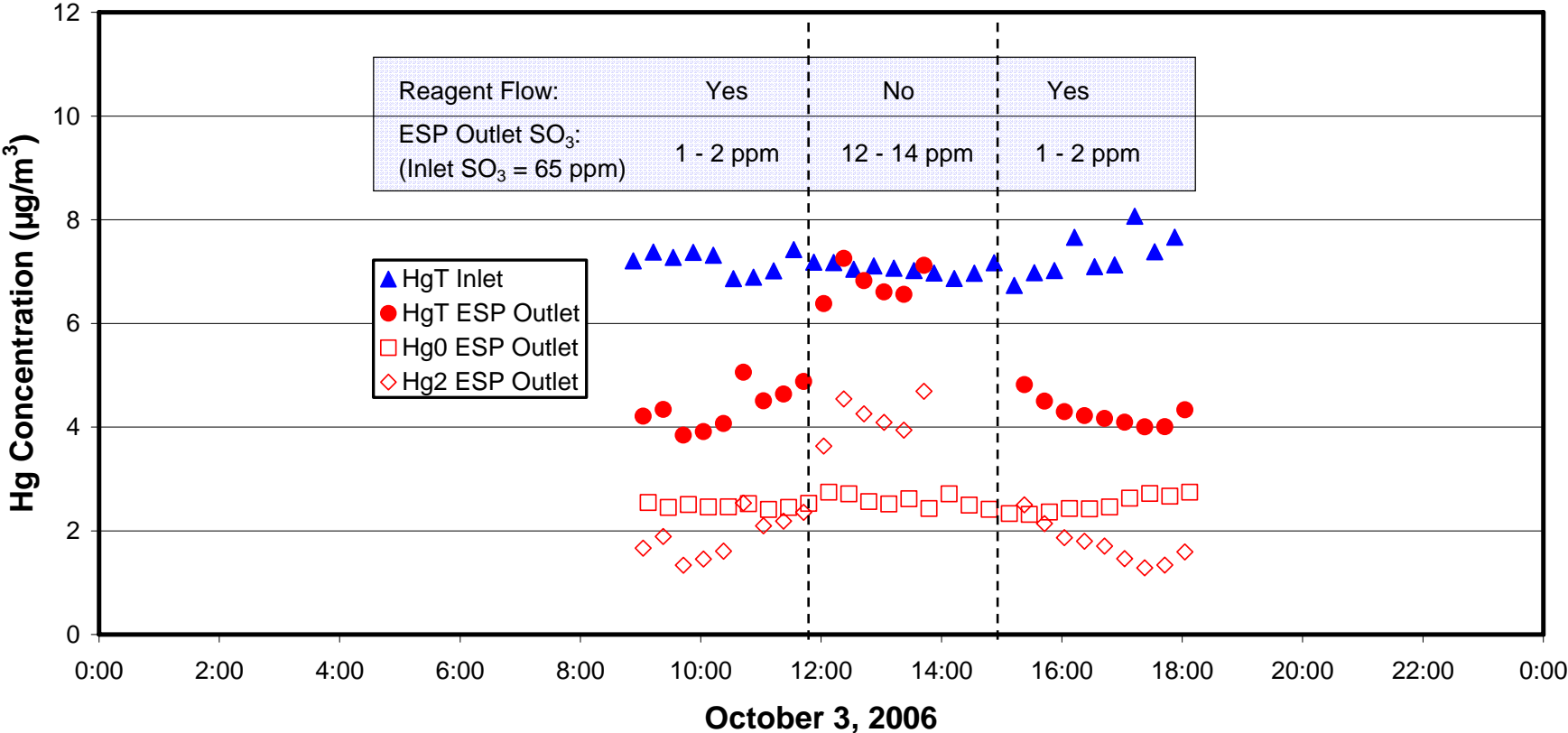
- Evaluate Impact of  $\text{SO}_3$  on Hg Capture by Carbon in Ash
- Data Collected During & Between Reagent Tests
  - ✓ Tests with/without  $\text{SO}_3$  injection
  - ✓ Tests with/without reagent injection
  - ✓ Relatively short test periods (screening tests)
  - ✓ Data collected for all five reagents

# Mercury Removal Testing Data Collection

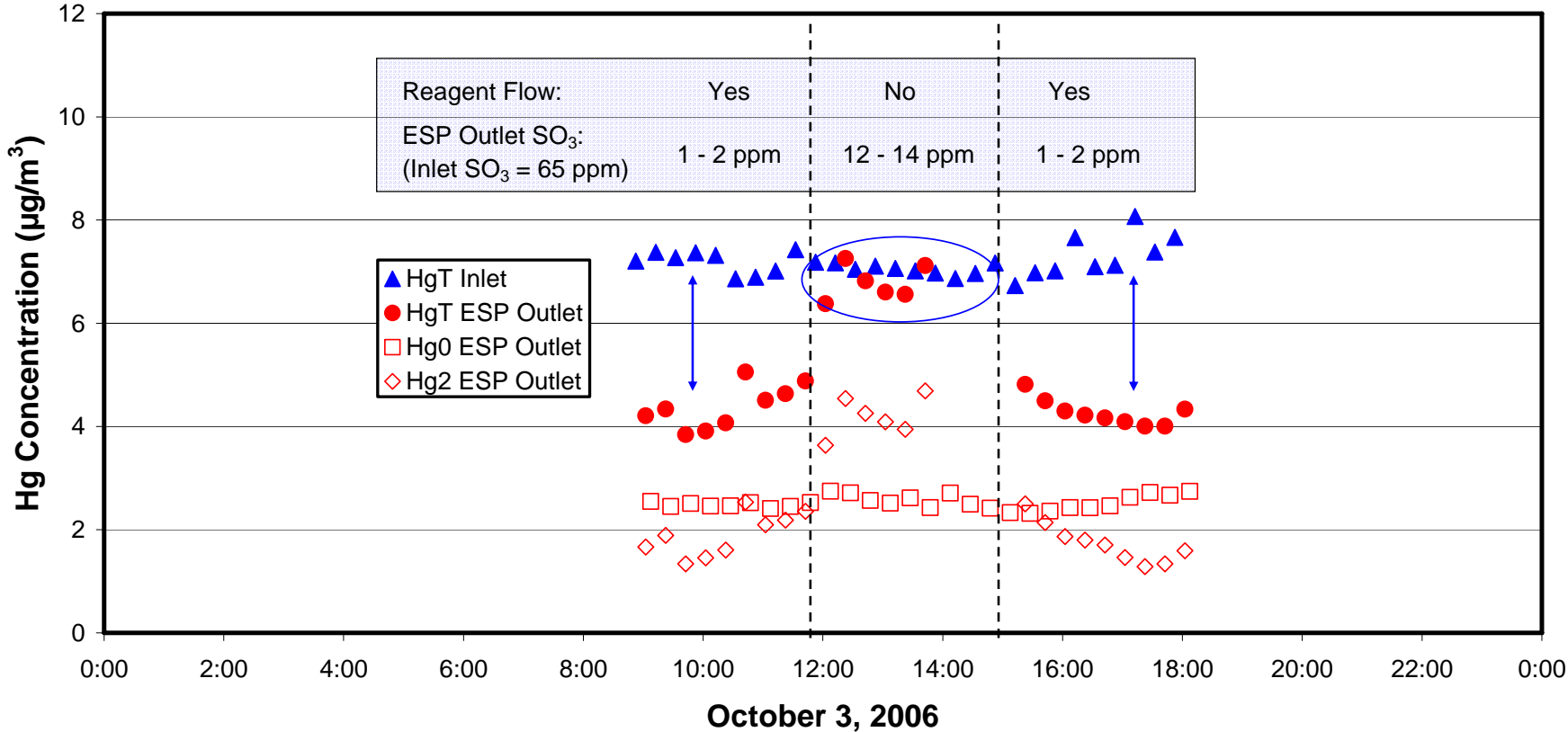
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- Hg Concentration Data Collected by Southern Research for EPRI
  - ✓ Tekran Model 2375A Mercury Vapor Analyzer with a semi-continuous monitoring system
  - ✓ Measurements of total & elemental mercury
  - ✓ Measurements at MRC inlet & ESP Outlet
- Operating Conditions
  - ✓ LOI concentrations of 3 – 5% (SRI)
  - ✓ AH & ESP outlet temperatures of 300 & 285°F for all tests

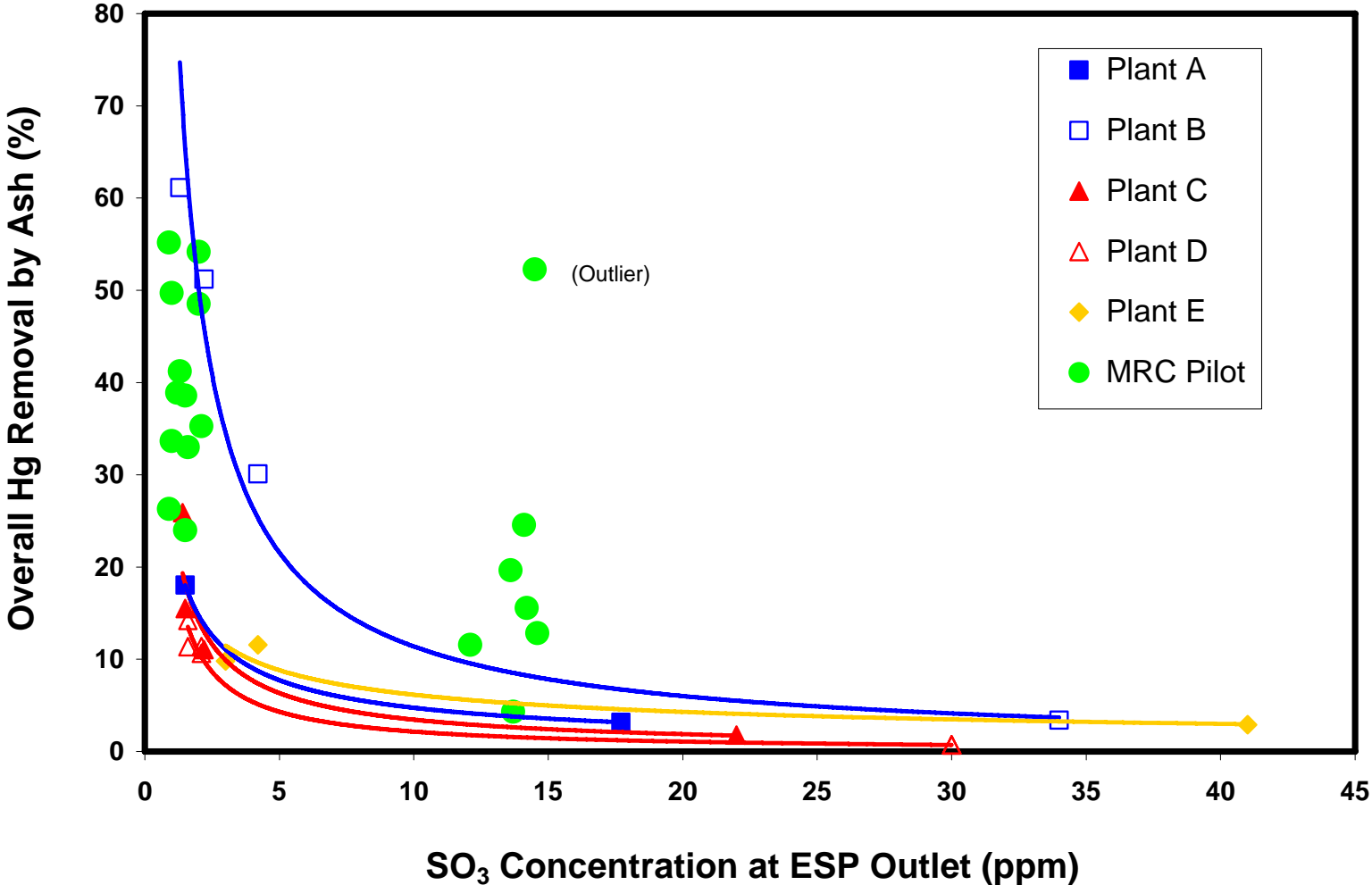
# Sample Mercury Removal Data (NaOH)

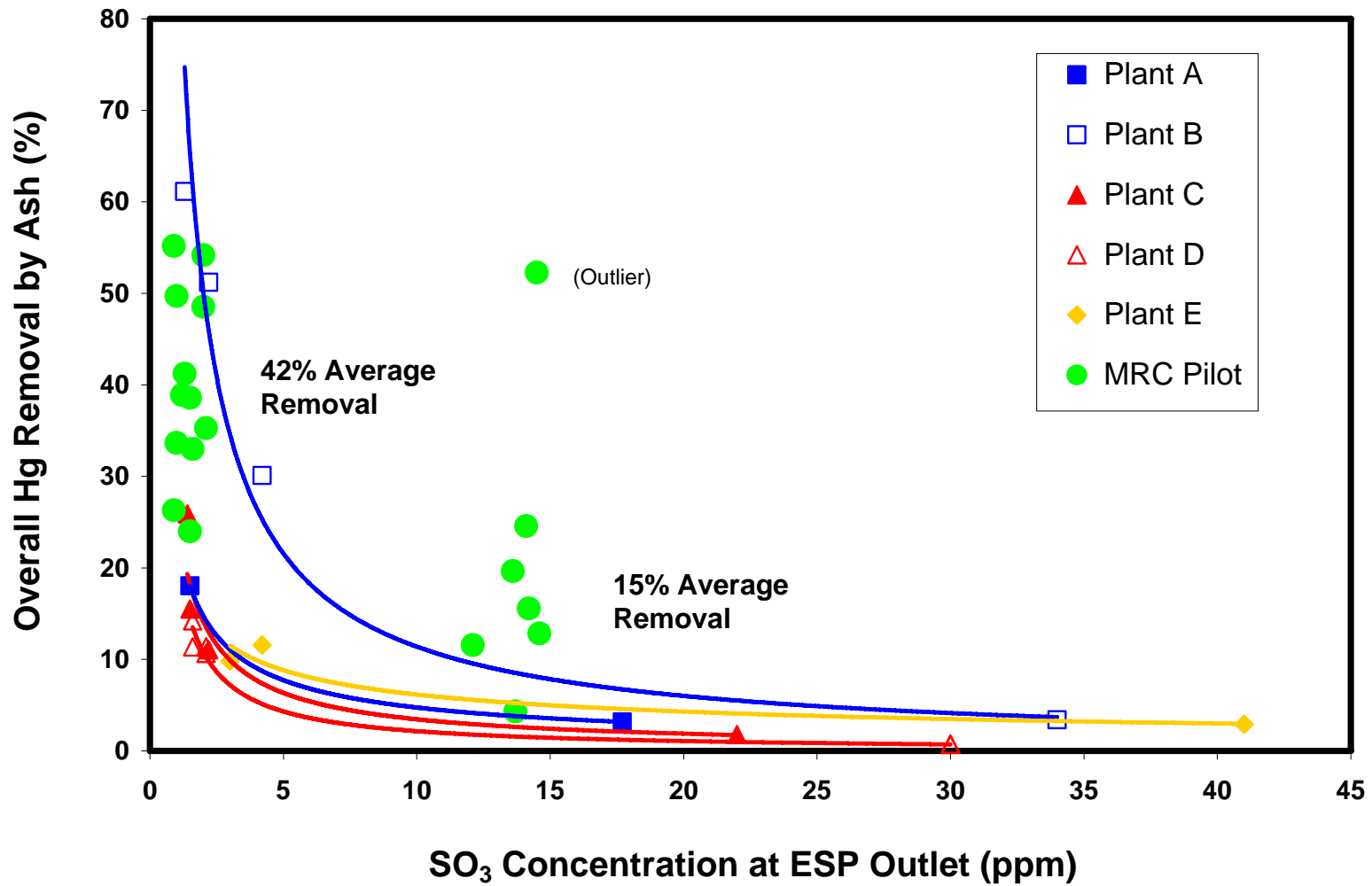


# Sample Mercury Removal Data (NaOH)



# Comparison of Pilot and Commercial-Scale Results





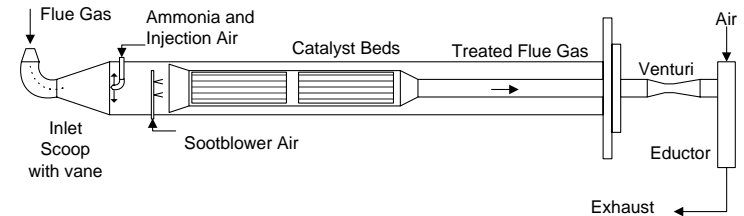
# Conclusions

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- Mercury Removal Efficiency was Affected by the  $\text{SO}_3$  Concentration at the ESP Outlet
  - ✓ Mercury removal increased at lower  $\text{SO}_3$  concentrations
- The Results Suggest that Oxidized Mercury was the Species that was Removed by the Fly Ash

# “Pre-SCR” Injection Testing Overview

- Evaluate Impact of Catalyst Exposure to Sodium Reagents
- Host Site - FirstEnergy Mansfield
  - ✓ SBS system in operation
- Catalyst “Slip Stream” Mini-Reactor (FERCO)
- Short-Term Testing
  - ✓ Exposure to SBS (675°F)
  - ✓ Simulate lay-up conditions
- Catalyst Analysis
  - ✓ Bulk and surface sodium
  - ✓ Activity (relative)
- November 2006



# Project Participants

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- FirstEnergy
- Ameren
- Dayton Power & Light
- Duke Energy
- FERCo
- CERAM Environmental
- Cormetech

# “Pre-SCR” Injection Testing Drivers

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- Allow “Upstream” SBS Injection where Residence Time is Limited
  - ✓ Use residence time within the SCR itself
- Allow a Reduction in the Minimum Operating Temperature (MOT) of the SCR
  - ✓ MOT limited by ABS formation in the catalyst
  - ✓ Many utilities installing scrubbers and then switching to higher sulfur coals, which increases the MOT
  - ✓ Increasing the flue gas temperature can be expensive or impractical, even with economizer bypass
  - ✓ “Pre-SCR” injection could remove combustion-generated  $\text{SO}_3$ , allowing reduction in MOT

# “Pre-SCR” Injection Testing Objectives

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- Determine Feasibility & Impacts of Sodium Reagent Upstream of the SCR
  - ✓ Will normal, high-temperature operation of the SCR be affected by the sodium particulate?
  - ✓ Will SO<sub>3</sub> removal upstream of the SCR reduce the MOT?
  - ✓ What conditions or operating practices will be required during outages or lay-ups to protect the activity of the catalyst?

# “Pre-SCR” Injection Testing Approach

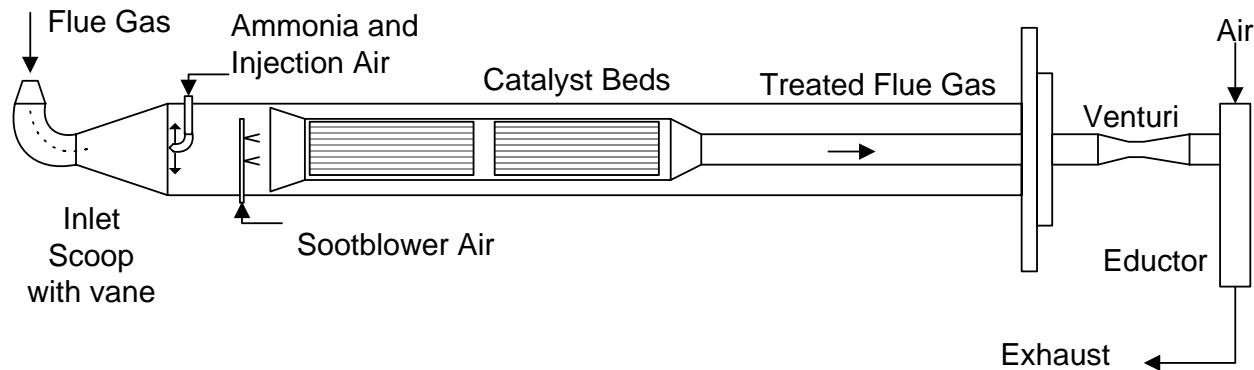
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- Phase I - Screening Tests
  - ✓ Short-term testing to determine feasibility
- Phase II - Longer-Term Testing
  - ✓ Quantify impacts of longer-term exposure on catalyst activity
- Phase III - Pilot and/or Full-Scale Testing
  - ✓ Demonstrate the Pre-SCR concept under actual operating conditions

# “Pre-SCR” Injection Testing Approach

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- Use FERCo Mini-Reactors to Expose Catalyst Samples to SBS-Containing Flue Gas



- Install the Equipment at FirstEnergy’s Mansfield Station, which has SBS Injection
  - ✓ A horizontal installation was required (FERCo modified the equipment to accomplish this)

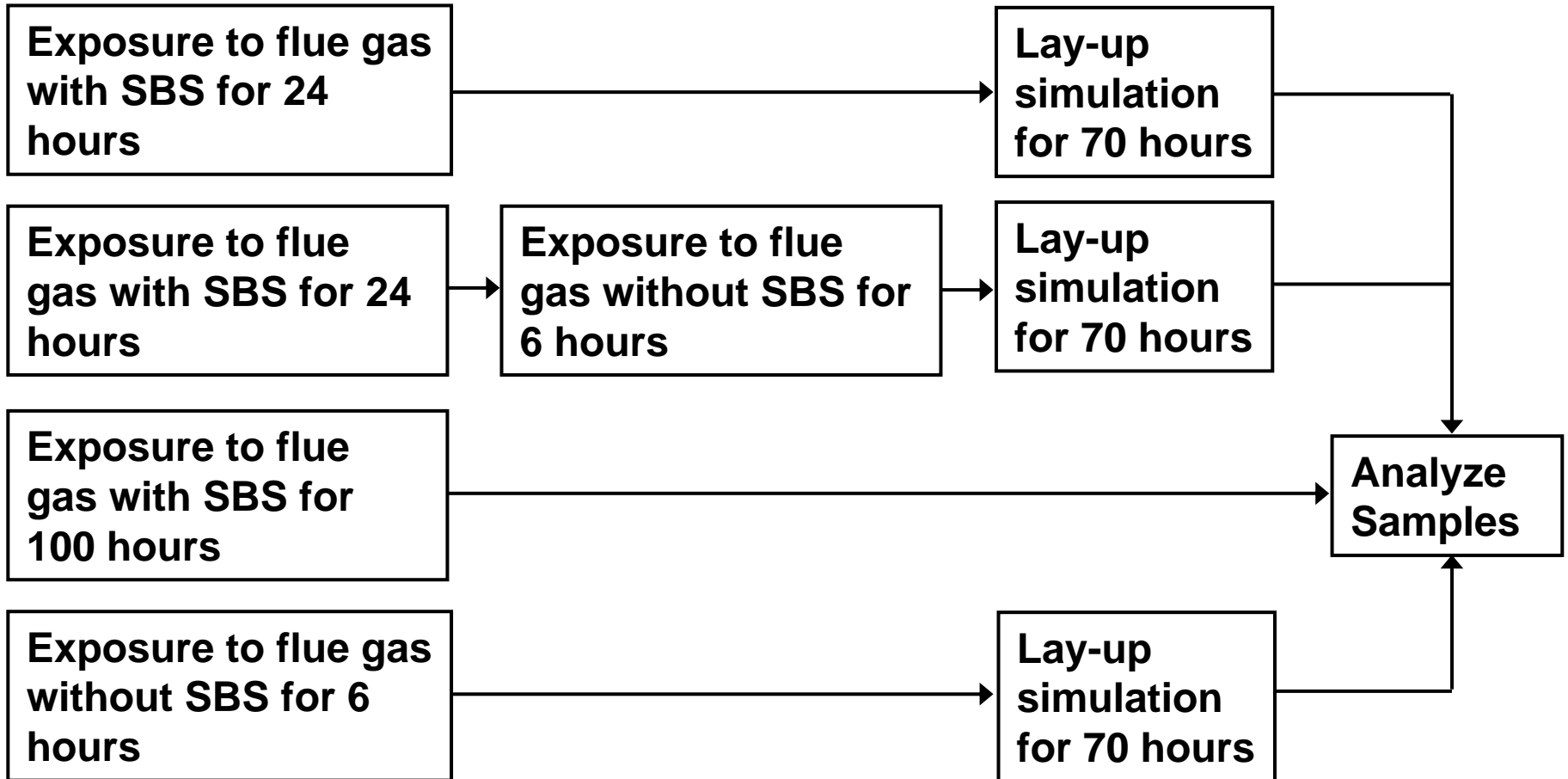
# “Pre-SCR” Test Plan

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- Exposure to Flue Gas
  - ✓ Duct Temperature of  $\sim 700^{\circ}\text{F}$
  - ✓ Gas Velocity Through Catalyst  $\sim 17$  aft/s
- Exposure to “Lay-Up” Conditions
  - ✓ Air at  $40^{\circ}\text{F}$  above ambient ( $<20\%$  RH)
  - ✓ Air at  $10^{\circ}\text{F}$  above ambient (ambient RH)
  - ✓ Humidified (saturated) ambient, then heated air to  $10^{\circ}\text{F}$  above ambient (about  $80\%$  RH)
- Analysis of Samples for Catalyst Activity & Bulk/Surface Sodium

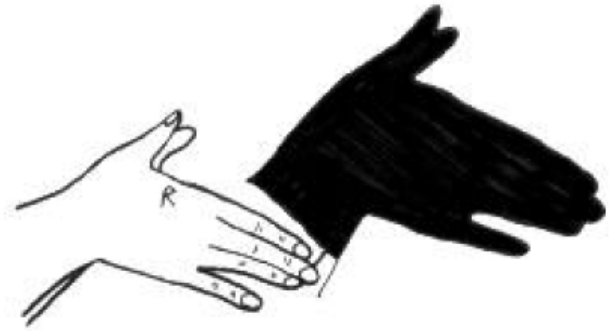
# “Pre-SCR” Test Plan

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# Just in Case There's No Data !

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# Wait! We Do Have Some Results!

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- One Catalyst Type Showed a Decline in Activity for Some Samples, the Other Type Did Not
- The Lay-Up Conditions used in the Testing Affected the Change in Activity
- It is Not Yet Clear if the Cause of the Activity Decline was the Sodium from the SBS Reagent or the Hostile Lay-Up Conditions
  - ✓ Samples not exposed to SBS also showed activity changes
- The “Scouring” Operating Procedure Did Not Seem to Provide Any Benefit
- Need to Examine Remaining Analytical Results

# Additional Testing

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- Conduct Phase II Longer-Term Testing
  - ✓ Age catalyst 3 – 6 months
  - ✓ Include  $\text{NH}_3$  injection & monitor activity “in-situ”
  - ✓ Evaluate ability to “control” SCR-generated  $\text{SO}_3$
  - ✓ Evaluate actual reduction in MOT (if possible)
  - ✓ Use vertical reactor configuration